

# Biofuel impact on Diesel engine after-treatment: deactivation mechanisms and soot reactivity

E. Iojoiu<sup>1</sup>, V. Lauga<sup>1</sup>, J. Abboud<sup>2</sup>, G. Legros<sup>2</sup>, A. Matynia<sup>2</sup>, J. Bonnety<sup>2</sup>, P. Da Costa<sup>2</sup>, J. Schobing<sup>3</sup>, A. Brillard<sup>3</sup>, G. Leyssens<sup>3</sup>, V. Tschamber<sup>3</sup>, P. Anguita<sup>4</sup>, J.M. Garcia-Vargas<sup>4</sup>, L. Retailleau<sup>4</sup>, S. Gil<sup>4</sup>, A. Giroir-Fendler<sup>4</sup>, M.-L. Tarot<sup>5</sup>, F. Can<sup>5</sup>, D. Duprez<sup>5</sup>, X. Courtois<sup>5</sup>

<sup>1</sup> Renault Trucks - Volvo Group Trucks Technology, Powertrain Engineering, 99 route de Lyon – 69806 Saint-Priest Cedex, France

<sup>2</sup> Sorbonne Universités, UPMC, Univ. Paris 6, CNRS, UMR 7190, Institut Jean Le Rond d'Alembert, 2 place de la gare de ceinture, 78210 Saint-Cyr-L'École, France,

<sup>3</sup> Laboratoire Gestion des Risques et Environnement, Université de Haute Alsace, 3b rue A. Werner 68093 Mulhouse cedex, France

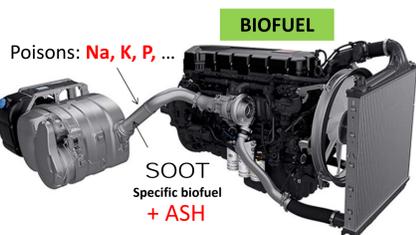
<sup>4</sup> Univ Lyon, Université Claude Bernard Lyon 1, CNRS, IRCELYON, 2 avenue Albert Einstein, Villeurbanne, F-69622, France

<sup>5</sup> Université de Poitiers-CNRS, Institut de Chimie des Milieux et des Matériaux de Poitiers (IC2MP), UMR 7285, 4 rue Michel Brunet, TSA 51106, 86073 Poitiers Cedex 9, France

**BACKGROUND:** The new emission standards for diesel engines empower the need of complex and high efficient after-treatment systems, the durability being a crucial aspect. When biofuel is used, the after-treatment catalytic system is exposed to large amounts of poisons, the particles composition being as well impacted. The comprehension of the involved deactivation mechanisms as well as soot reactivity is a complex and multidisciplinary challenge.

**METHODOLOGY:** One focus was the study of the deactivation of the DOC and SCR catalysts through poisoning. Limited information is available about the physics and chemistry of the particles formed when biodiesel is used. The second focus was therefore devoted to the impact on soot reactivity in mechanistic and kinetic terms using model and real soot.

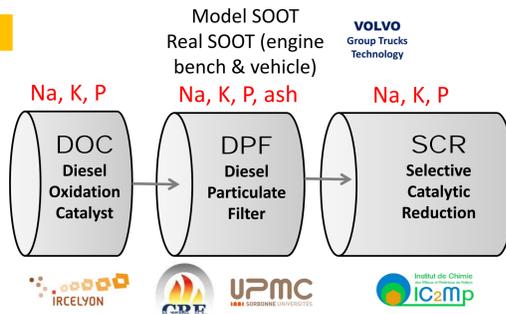
## HOW AFTER-TREATMENT SYSTEM IS AGED USING BIOFUEL ?



## APPBIO PROJECT

Model poisons:  
Na, K, P

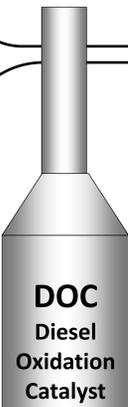
Exhaust specification  
Catalyst benchmark  
Technical requirements  
Soot production  
Catalyst poisoning



Global kinetic model to predict the after-treatment ageing

## HOW SOOT LOADING & REGENERATION ARE IMPACTED BY USING BIOFUEL ?

Reactivity – Durability/ Characterization – Mechanisms – Ageing – Regeneration



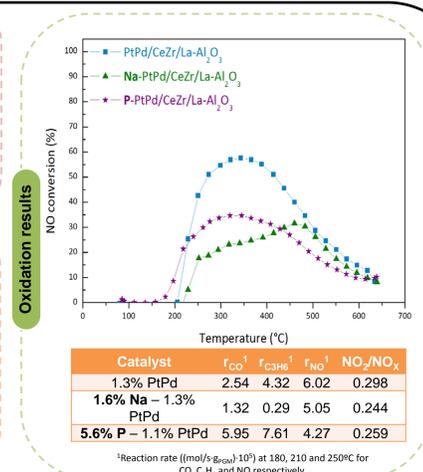
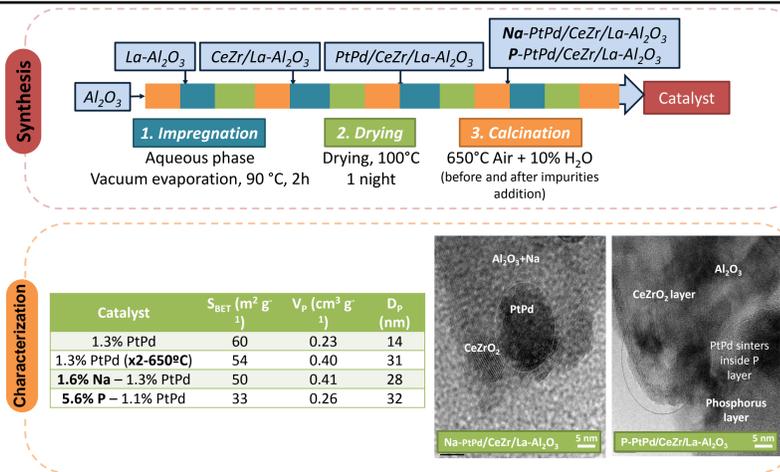
DOC  
Diesel  
Oxidation  
Catalyst



The DOC was exposed to Na and P impurities, which are specific of the use of biofuel, being studied their impact on the catalytic performance.

BET specific surface area decreased and the total pore volume increased after addition of Na, due to the second hydrothermal treatment. In presence of P, partial blocking of the smallest mesopores decreasing pore volume was detected. TEM images showed a homogenous distribution of Na on the alumina bulk, which could modify the Al<sub>2</sub>O<sub>3</sub> acid sites. On the other hand, bigger Pd-Pt particles were formed and a layer of phosphorus coated on the alumina surface was observed in the case of P-poisoned catalyst.

Catalytic results have shown that **Na have a negative impact on CO oxidation**, whereas **an improvement can be observed in presence of P**. In addition, **P poisoned catalyst enhance C<sub>3</sub>H<sub>6</sub> conversion**, while **Na impurities have the contrary effect**. All elements have shown a **negative effect on NO oxidation**.

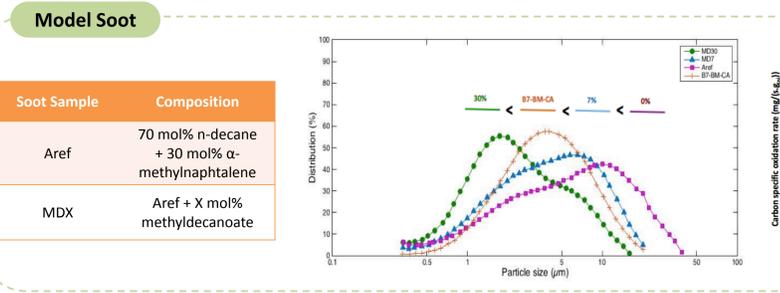


DPF  
Diesel  
Particulate  
Filter



The real soot samples were collected from filters operated on a medium-duty truck in real driving conditions or from engine bench using standard or 100% biofuel as well as doped biofuel. The use of biodiesel significantly reduces the soot formation. There is no significant impact of biodiesel on the soot specific surface, the higher value being obtained for an accelerated soot loading. Experiments performed at laboratory scale showed no relationship between volatile fraction and real soot reactivity under passive regeneration conditions. Adding **alkali metals** to the real Diesel soot **enhances soot oxidative reactivity** in the whole temperature range (200 – 600 °C), regardless the cycle of production applied. Those species act as catalyst for soot oxidation process. The **kinetic of soot oxidation**, in presence of water in the feed gas, at low temperature (≤ 400 °C), is **significantly increased in the presence of phosphorus**.

Model soot were collected on a glass microfiber filter in the post flame region of an atmospheric axisymmetric co-flow diffusion flame burner. Particle size distribution and oxidative reactivity of model soot from the burner are in the same range as real soot derived from Diesel engine.



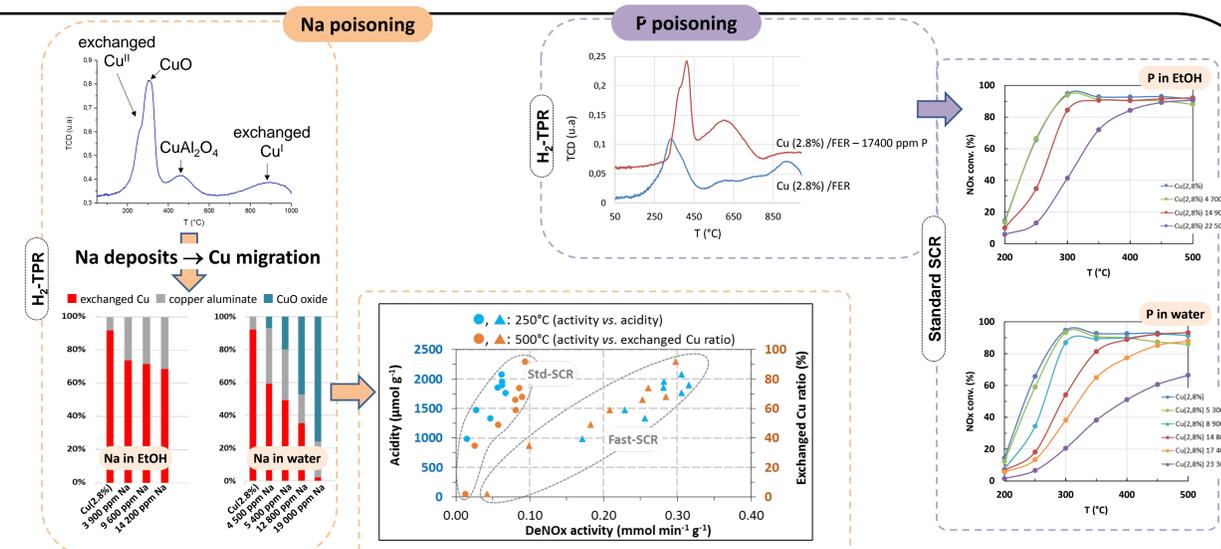
SCR  
Selective  
Catalytic  
Reduction



The impact of Na and P on the SCR activity of Cu/Fe catalyst was studied depending on the mineral loading (until ≈2<sub>wt</sub>%) and the impregnation solvent (H<sub>2</sub>O or ethanol).

Acidity (measured by NH<sub>3</sub> adsorption) was poisoned after Na addition which directly affected the NO<sub>x</sub> conversion at low temperature (250°C). **Na impregnation in water led to a stronger catalyst deactivation than in ethanol, because water favors the migration of the Cu exchanged species, leading to the formation of CuO extra framework species.** It appears that the deNO<sub>x</sub> efficiency at high temperature (500°C) is clearly related to the amount of active exchanged copper.

**Cu-P interactions were evidenced** after phosphorus addition, leading to a decrease in redox behaviors (NO oxidation and NH<sub>3</sub> oxidation) and consequently in the SCR activity (especially at low temperature and in Standard SCR condition). Again, lower deactivations were observed when the wet impregnations were performed in ethanol.



## CONCLUSIONS

Impact of biofuel poisoning elements to DOC & SCR performance has been identified, Na having the highest deactivation potential; SCR deactivation mechanism has been proposed.

Comparison between real and model soot as well as the impact of Na, K and P on soot oxidation have been successfully studied.

Biofuel use leads to more poisoning, especially due to Na, with a direct impact on DOC performance in terms of NO<sub>2</sub> formation which might indirectly affect soot oxidation rate and NO<sub>x</sub> reduction.

## PERSPECTIVES

Collected data by academic partners combined with results obtained on engine bench and vehicles will be used to build kinetic models that will be integrated to empower an ageing predictive model taking into account the use of biodiesel;

Adapted/innovative systems with improved poisoning resistance.

Global kinetic model - biofuel impact

Chemical composition of biofuel (Na, K, P, etc)

Biofuel consumption

AGEING PREDICTION

NO & hydrocarbon oxidation

NO<sub>x</sub>

Soot oxidation

Ash deposit