INVESTIGATION OF THE INFLUENCE OF AMMONIA AND HYDROGEN ADDITION ON SOOT FORMATION IN CO-FLOW LAMINAR DIFFUSION FLAMES

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MOTIVATION

- Ammonia and hydrogen are promising carbon free fuels that can be produced by using renewable energy sources in the future. Ammonia has higher volumetric energy density than liquid hydrogen. Ammonia is also easier to store and transport. Infrastructure is well established. Poor combustion characteristics limit ammonia use as a fuel.
- Co-firing ammonia with hydrocarbon fuels could enable wider use in combustion applications and help lower CO₂ emissions. When co-firing, soot production must also be considered as soot is a harmful pollutant. Experiments can help validate mechanisms and soot formation models.
- In this study, we will provide a comparison between ammonia and hydrogen addition to ethylene-air diffusion flame to investigate the effects on soot formation.

 $Q_{fuel\,mix} =$

 $Q_{NH3 or H2}$

 $Q_{fuelmix}$





Laser measurement setup for co-flow burner inside the pressure vessel **Temperature:** Type S thermocouple (PtRh-10%/Pt) is inserted into the flame at different locations. Temperature is averaged over 20 s at each location. Temperature profiles help reveal soot deposition zones [1]. **Qualitative PAH:** Planar LIF technique captures signal from excited PAHs in the flame using a lower energy 266nm laser ~3.5 mJ. Fluorescence signals from 3-4 ring PAHs and larger PAHs were captured using 400,450 and 500 nm bandpass filters respectively [2,3].

Soot volume fraction (SVF): Planar LII is used to heat soot particles with high energy laser pulses ~25 mJ. ICCD camera with 435nm BP filter captures the incandescence signal from heated soot particles. LII is calibrated to previous 2D LOSA measurements to calculate SVF [4]. **Soot Particle Size Distribution:** Soot particles are sampled from the flame with a quartz probe at different locations along the centerline. After a two-stage dilution, particles are sent to the SMPS to measure size and number concentration. (work in progress)

HAB (mm) **3-4 ring PAHs from LIF:** LIF signal reduces to minimum levels by 25% NH_3 addition.





Soot Volume Fraction from LII HAB [mm] HAB [mm] **Ammonia addition:** chemical effects Hydrogen addition: thermal effects were more dominant were more dominant NH₃ addition to diluted ethylene/air NDFs H₂ addition to diluted ethylene/air NDFs N2 diluted Ethylene 10 2350 2350 Adiabatic Flame

Tested flame conditions for the study. Conditions are shown as a function of the fraction of added NH_3 (top) or H_2 (bottom) in the fuel flow.

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CONCLUSION

- Soot volume fraction decays exponentially with ammonia vol%.
- Soot volume fraction increases linearly with hydrogen vol%.
- PAH-LIF signal reaches a minimum by 25% ammonia in fuel flow.
- PAH-LIF measurements in hydrogen addition cases were affected by quenching due to the increase in flame temperature.

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