# BLACK CARBON AND JEONGHOON LEE KOREA UNIV. of TECH. & EDU. ELEMENTAL CARBON CONCENTRATIONS OF Black carbon (BC) is formed by the incomplete combustion of hydro-SPARK-GENERATED CARBON PARTICLES

measurement

BLACK CARBON CONCENTRATIONS ARE COMPARED WITH ELEMENTAL CARBON **CONCENTRATIONS FOR CARBON NANO-**PARTICLES GENERATED USING A SPARK-DISCHARGER. THE SHAPE OF CARBON IZING THE OPTICAL PROPERTIES AS WELL TION AND THE MASS CONCENTRATION.



#### INTRODUCTION

incomplete combustion of hydrocarbon fuels and intensively absorbs light of all wavelengths of solar radiation. BC is emitted with & CPC. other particles and gases such as  $SO_2$ ,  $NO_x$  and organic carbon (OC). These other gases and particles than BC sometimes create uncertainty associated with measurement techniques which are affected by the presence of the 'others'. In the present study, to remove the effect of the other particles and gases than BC, we have initiated the investigation of the nearly 'pure' BC. The optical properties such as the absorphysical properties such as the characterize the pure BC. Then we compare the BC with the elemental carbon (EC) to quantify the relation between the BC and the EC.

#### **Particle Generation**

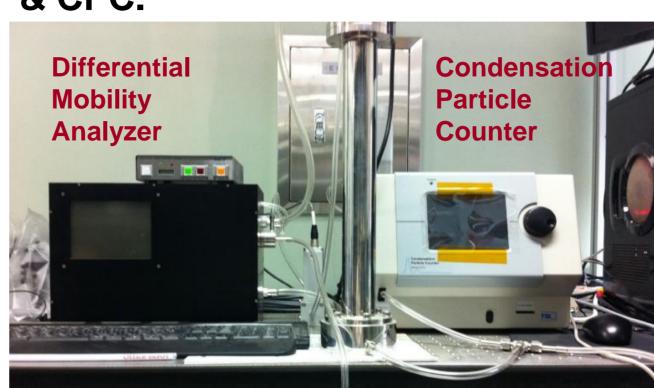
A spark discharger (PALAS GmbH, DNP 2000) generates black carbon (BC) by continuous sparks created by high voltage. The generated particle is called as 'PALAS carbon', which is known to similar to diesel



**Mass Concentration Measurement** Multi-angle Absorption Photometer (MAAP, Thermo Scientific 5012) was used to measure the BC mass concentration, which is converted from light absorption coefficient. When aerosol samples are attached on the fiber filter tape, light from a light source (670 nm) scatters back to hemisphere. The MAAP has three detectors to revise reflection (back hemisphere) and transmission (forward hemisphere) for correction of any backward scattering. Sampling time was 1 min. Aerosol was introduced into the MAAP through a sampling inlet.

**Size Distribution Measurement** 

The size distributions of the sparkgenerated soot were measured by the differential mobility analyzer (DMA, home-designed) equipped with a condensation particle counter (CPC, TSI 3775). The aerosol was dried using a diffusion drier before the aerosol is introduced into DMA

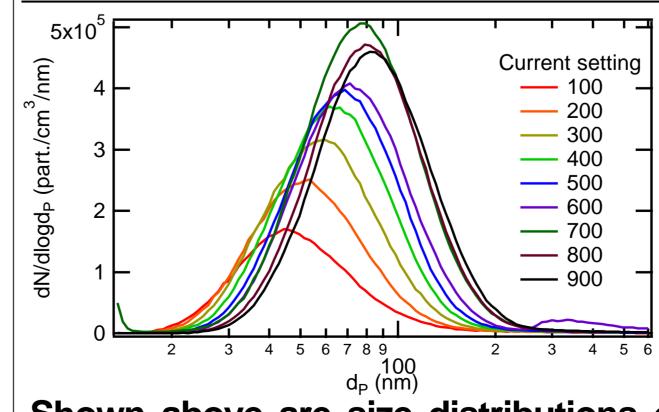


**Elemental Carbon Measurement** Elemental carbon concentra-tions are measured by an EC/OC Analyzer | Effective Mass of BC (Sunset, USA), which uses ption cross section as well as the thermal-optical method. In an oxygen-free helium atmosphere, PALAS effective density will be estimated to carbon collected onto quartz filter is heated to remove all OC from the sample. As the organ-ic compounds are vaporized, they are immediately oxidized to  $CO_2$ . After the PALAS  $\stackrel{\triangle}{\sim}$ carbon on the filter is cooled to a relevant temperature, a 2% O<sub>2</sub>/He mixture is injected into the sample oven. The sample oven temperature is then increased up to a given temperature, during which the EC is oxidized to CO<sub>2</sub> due to the presence of oxygen. This CO<sub>2</sub> is then converted to CH<sub>4</sub> and detected by the Flame densities for various PALAS carbon Ionization Detection (FID) for the quantification of the EC.



### **RESULTS & DISCUSSION**

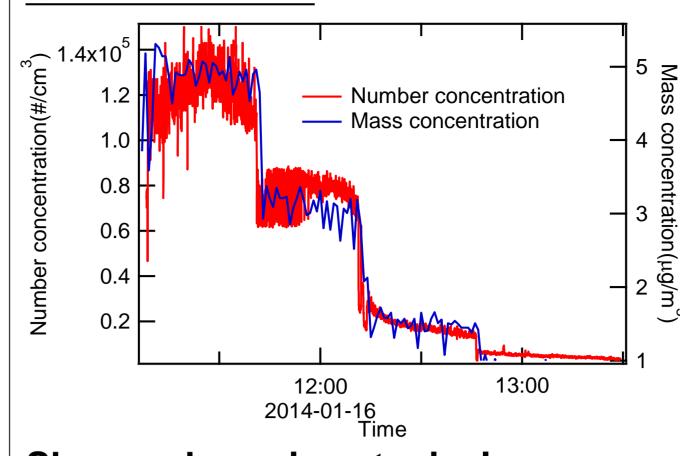
**Size Distribution of PALAS soot** 



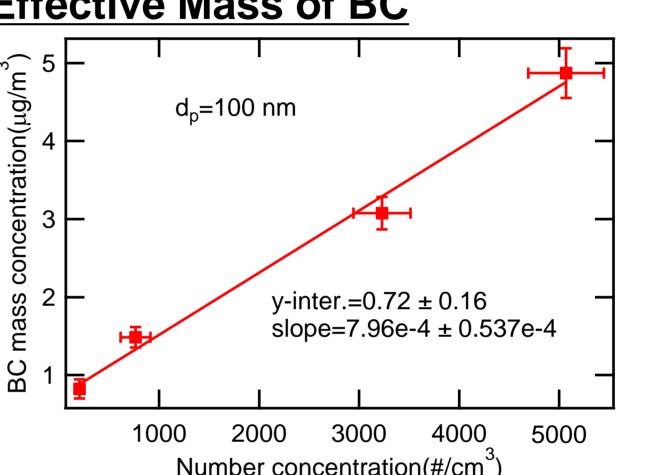
volumes were obtained.

### **Mass Concentration & Number**

Concentration



Shown above is a typical measurement data of BC mass concentration. Number concentration was also overlaid on the same graph. The BC mass concentration follows the number concentration within discernable detection resolution. This data will be used to obtain the effective mass of unit BC.

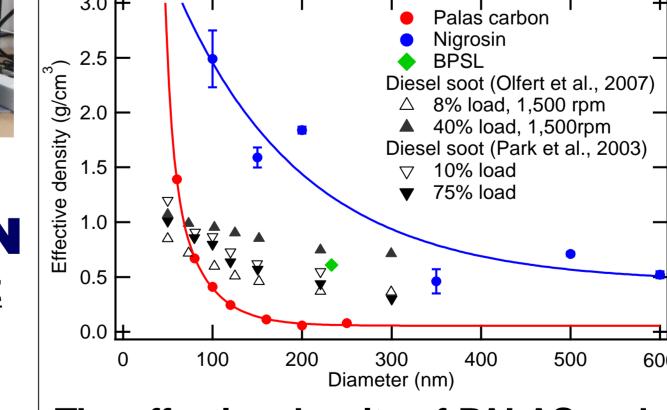


Number concentration(#/cm<sup>3</sup>)

The mass concentration for a typical size of spark-generated PALAS carbon was plotted as a function of the number concentration. The slope of the linear fit curve corresponds to the effective mass per unit particle. Effective particles were estimated as shown

## **Effective Density for Various**

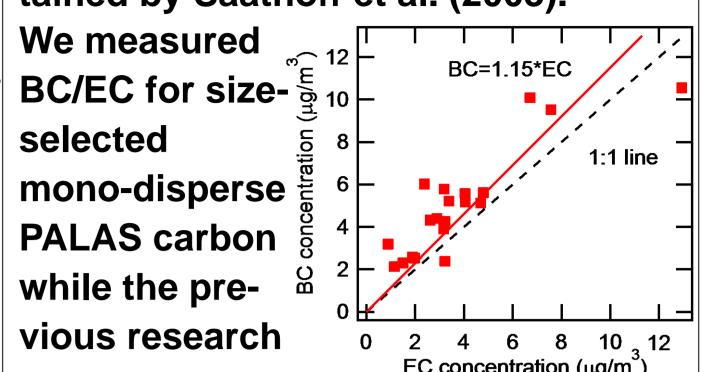
**Absorbing Particles** 



The effective density of PALAS carbon decreases as the particle size increases. This implies that the morphology of the PALAS carbon changes from a compact aggregate to a loose agglomerate. The effective density of nigro-Shown above are size distributions of sin and black dyed polystyrene latex PALAS carbon measured for various (BPSL) is higher than the PALAS carcurrent settings. As the current increal bon aggregate because nigrosin and ses the mode diameter increases. From BPSL are spherical, that is, of no void. the mobility diameter, the equivalent The result of diesel soot lies between those of PALAS carbon and nigrosin.

#### BC/EC Comparison (PALAS carbon)

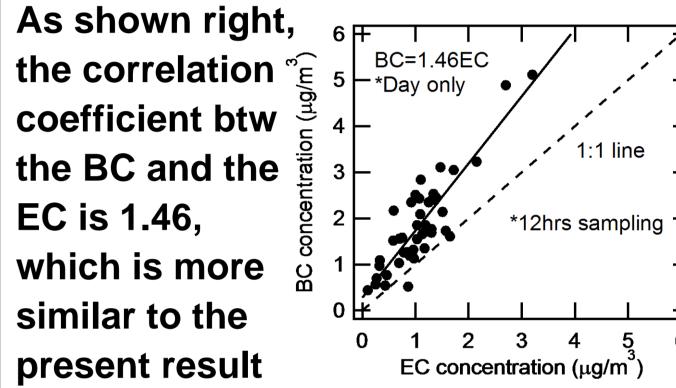
As shown below, the correlation coefficient between the BC and the EC for PALAS carbons ranged from 60 nm to 250 nm is 1.15, which is higher than the values of 0.15 and 0.26 obtained by Saathoff et al. (2003).



did for poly-disperse one, which might cause overlap onto filter in BC measurement, resulting in underestimation of BC concentration compared to EC.

#### **BC/EC Comparison (Ambient)**

As shown right, coefficient btw the BC and the EC is 1.46, which is more similar to the present result



than the previous study. This result implies that PALAS carbon can be a proxy for ambient absorbing aerosol. The caveat, however, is that this holds good for the size-selected one at low level of concentration.

#### CONCLUSIONS

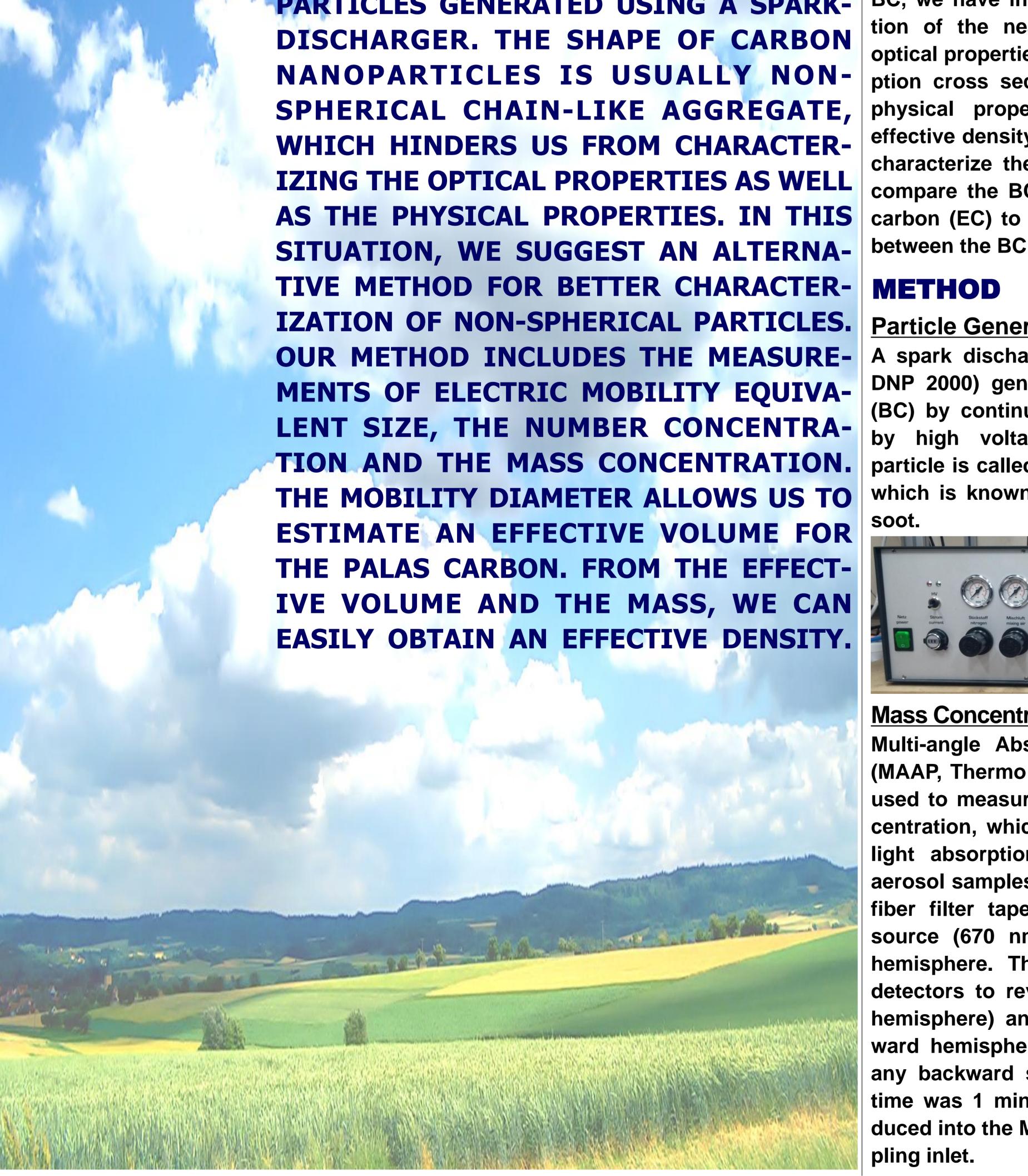
- BC and EC concentrations were compared for the PALAS carbon particles generated in a sparkdischarger.
- Effective density for PALAS carbon was determined using BC mass concentration, number concentration and mobility equivalent dia-
- BC/EC correlation coefficient for the size-selected PALAS carbon was 1.15, which showed similar trend to the value for ambient aerosol but higher than the values measured in a previous study.

#### **ACKNOWLEDGMENT**

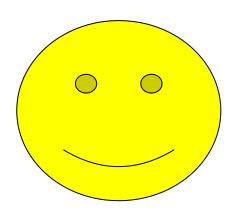
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#### REFERENCES

Olfert et al., 2007, J. Aerosol Sci.; Park et al., 2003, Env. Sci. Technol.; Saathoff et al., 2003, J. Aerosol Sci.



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