#### Volatile nano particle formation by diesel cars equipped with modern catalytic filter systems: New Insights from First Gaseous Sulphuric Acid Measurements

F.Arnold(1\*), T.Schuck(1), A.Weller(1), L.Pirjola(2), J. Keskinen(3), T. Rönkkö(3), T. Lähde(4), K. Hämeri(2,4), and D.Rothe(5)

(1) Max-Planck-Institute for Nuclear Physics, Heidelberg, Germany

- (2) University of Helsinki, Helsinki, Finland
- (3) Tampere University of Technology, Tampere, Finland
- (4) Finnish Institute of Occupational Health, Helsinki, Finland
- (5) MAN Nutzfahrzeuge, Nürnberg, Germany
- (\*) corresponding author F.A. (frank.arnold@mpi-hd.mpg.de)

#### Talk given at 10 ETH conference on "Combustion Generated Nano Particles", Zürich, 21-23 August 2006

Abstract. Gaseous sulfuric acid (GSA) is thought to represent an important if not the most important nucleating gas present in modern diesel automobile exhaust. It triggers the formation of new aerosol particles which grow by condensation and coagulation. Here we report on the first measurements of gaseous sulfuric acid in automobile exhaust. Two experiments have been made: Experiment 1 was a feasibility study involving a stationary passenger diesel car (operated in very low engine load conditions) equipped with a catalytic diesel particle filter system (CDF). Experiment 2 involved systematic measurements with a heavy duty diesel truck engine on a test bench at MAN Nutzfahrzeuge company in Nürnberg (Germany).

In experiment 1 the diesel fuel used had an ultra-low fuel sulfur content FSC of only 5 mg/kg. Here the exhaust leaving the tail pipe was passed through a tunnel where it experienced natural dilution with ambient air.. Measured GSA number concentrations reached up to  $1 \times 10^9$  cm<sup>-3</sup>. Freshly nucleated particles with diameters larger than 3 nm which were also measured. reached concentrations of up to  $1 \times 10^5$  cm<sup>-3</sup>. They were positively correlated with GSA for GSA exceeding a threshold value in the range of  $5 \times 10^7 - 2 \times 10^8$  cm<sup>-3</sup>. This suggests that GSA was involved in the formation of new volatile particles. However particle growth was strongly increased by condensation of condensable organic molecules (COM).

In experiment 2 two different FSC (7 and 36 mg/kg) were alternatively used. Engine load EL was systematically varied (25., 50, 75, and 100%). Three after treatment scenarios were investigated: CDF,

Also experiment 2 indicates that particle formation by nucleation is driven by GSA while particle growth is driven not only by GSA but also by COM.

# Volatile nano particle formation by modern diesel cars: New Insights from First Gaseous Sulphuric Acid Measurements

F.Arnold(1\*), T.Schuck(1), A.Weller(1), L.Pirjola(2), J. Keskinen(3), T. Rönkkö(3), T. Lähde(4), K. Hämeri(2,4), and D.Rothe(5)

(1) MPI, Heidelberg, Germany
(2) University of Helsinki, Helsinki, Finland
(3) Tampere University of Technology, Tampere, Finland
(4) Finnish Institute of Occupational Health, Helsinki, Finland
(5) MAN Nutzfahrzeuge, Nürnberg, Germany
(\*) corresponding author

Talk given at at 10 th ETH conference on "Combustion Generated Nano Particles", Zürich, 21-23 August 2006

#### **Background Information**

#### Sulphuric Acid Molecule H2SO4

- Represents powerful aerosol precursor
- Is also emitted by combustion of sulphurcontaining fossil fuels

## SULFURIC ACID MOLECULE H2SO4

- Most important property : large GA
  - → proton transfer to other molecule with large PA (Atmosphere : H2O)
  - → Gas-Phase Hydrates H2SO4(H2O)n
  - → Nucleation (H2SO4)a(H2O)w cluster

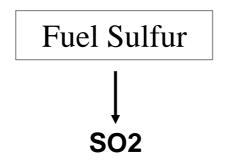
## SULFURIC ACID MOLECULE H2SO4

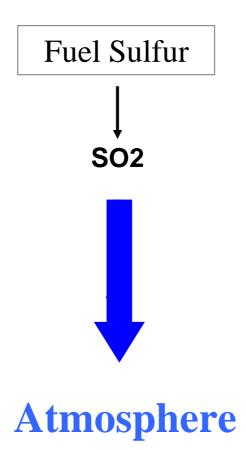
- Most important property : large GA
  - → proton transfer to other molecule with large PA (Atmosphere : H2O)
  - → Gas-Phase Hydrates H2SO4(H2O)n
  - → Nucleation (H2SO4)a(H2O)w cluster
- Atmosphere :

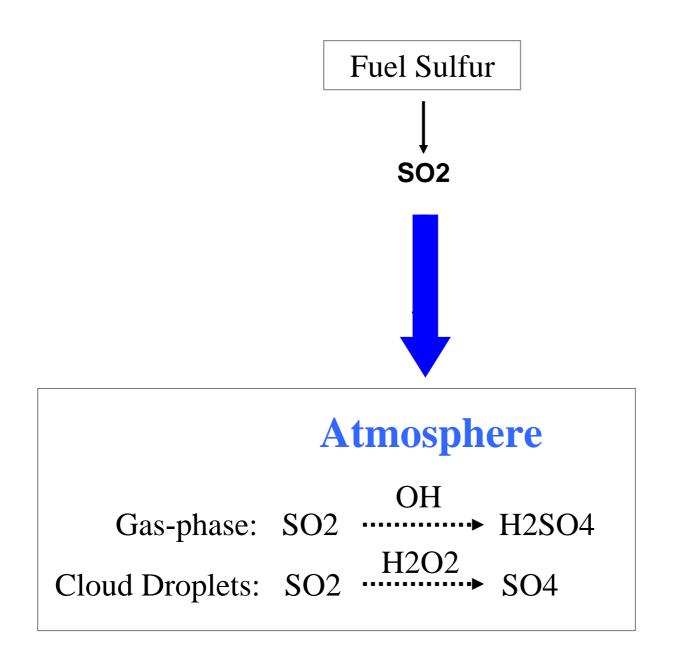
Secondary H2SO4: formed in Atmosphere from SO2 Primary H2SO4 : released from combustion

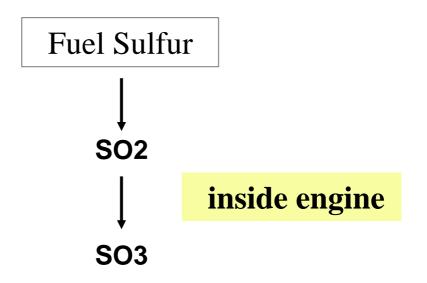
Example: mobile combustion sources (Aircraft, Ships, cars)

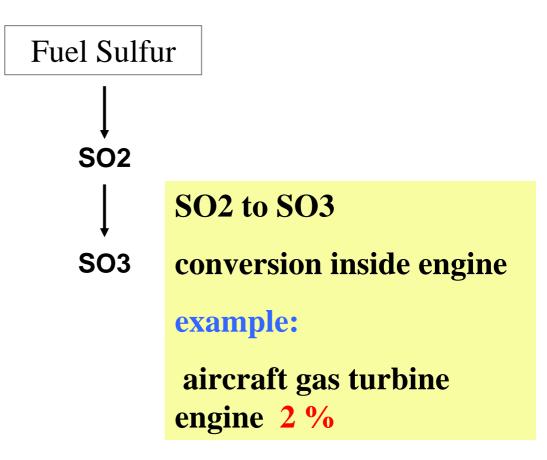
#### Simplified Scheme of Fuel Sulphur Conversion

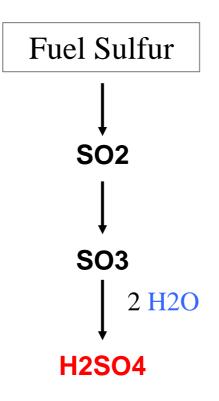


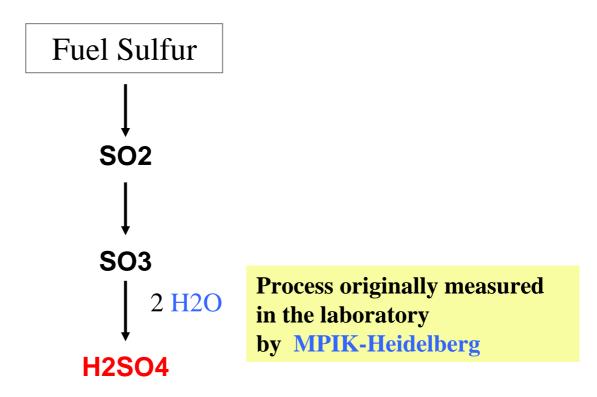


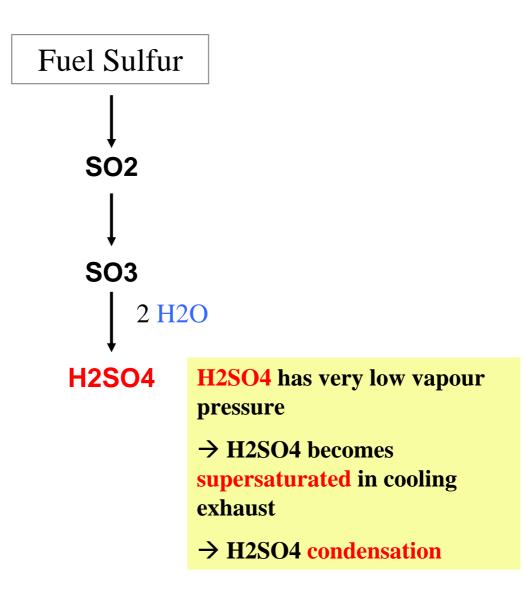


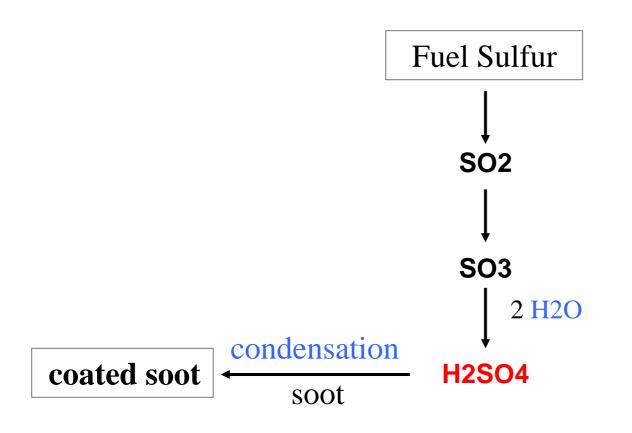


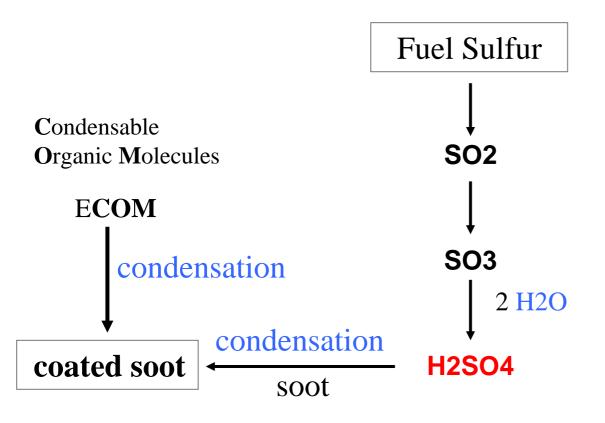


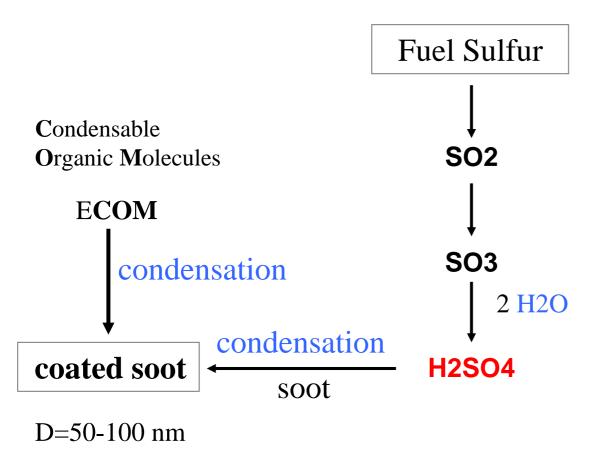


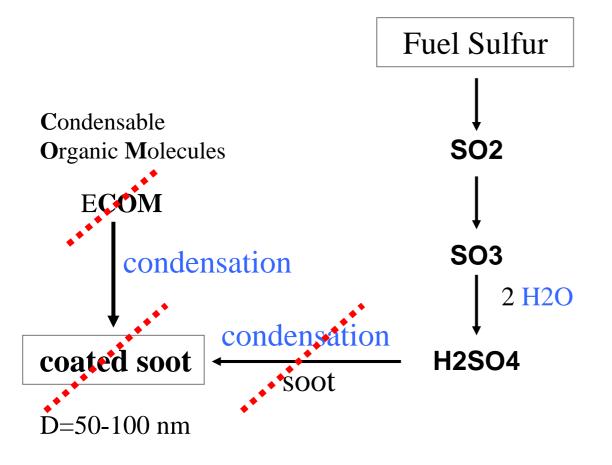




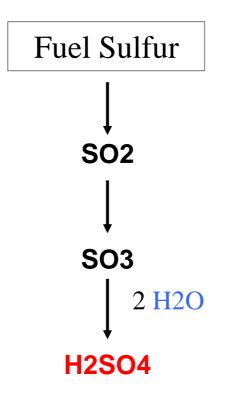




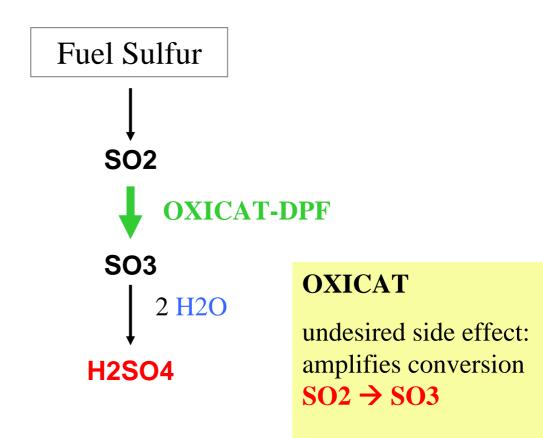


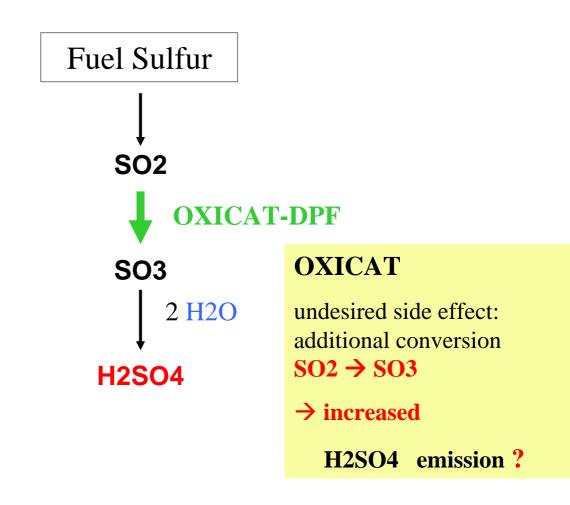


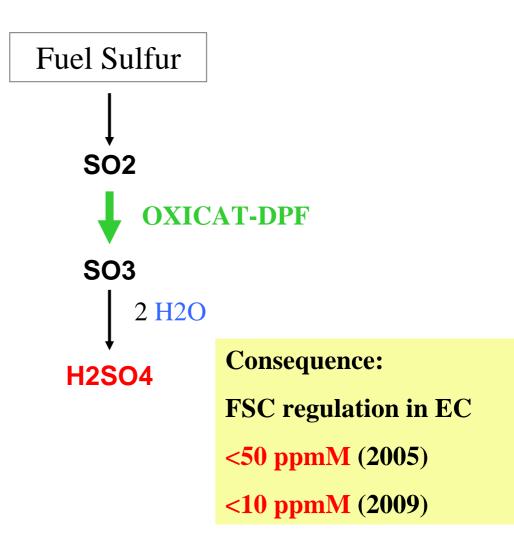
Modern diesel car is equipped with OXICAT-DPF

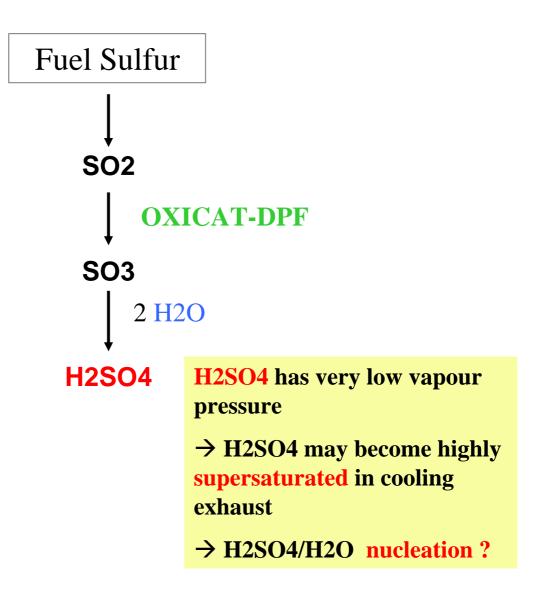


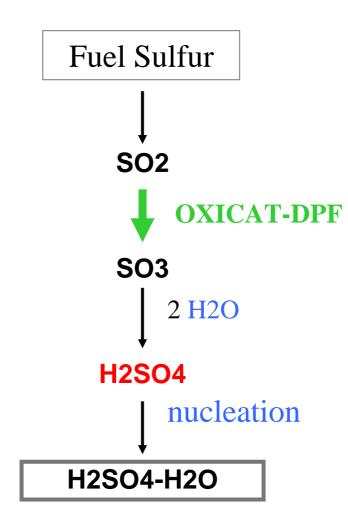
Modern diesel car is equipped with OXICAT-DPF

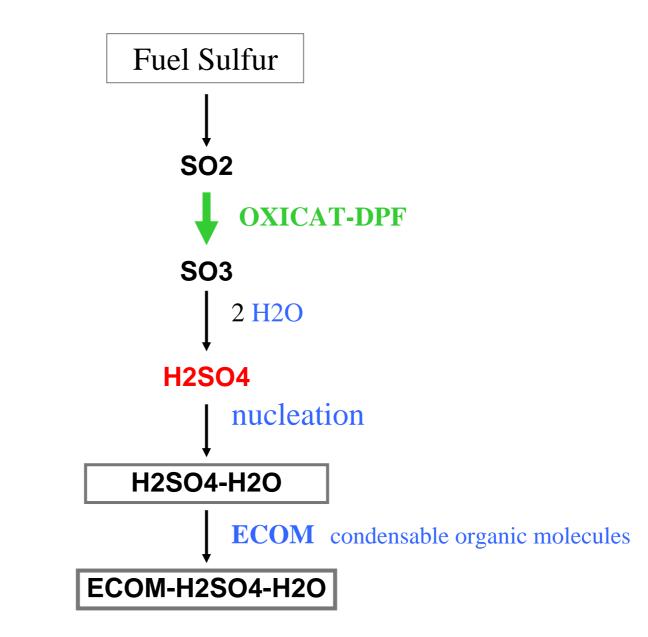


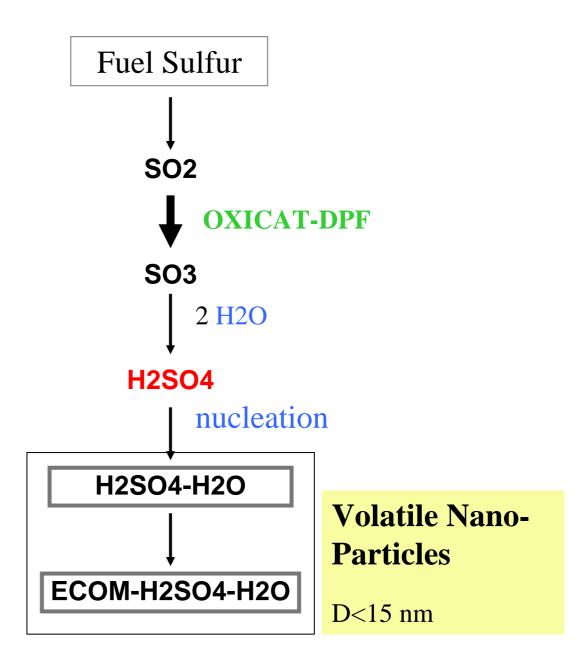






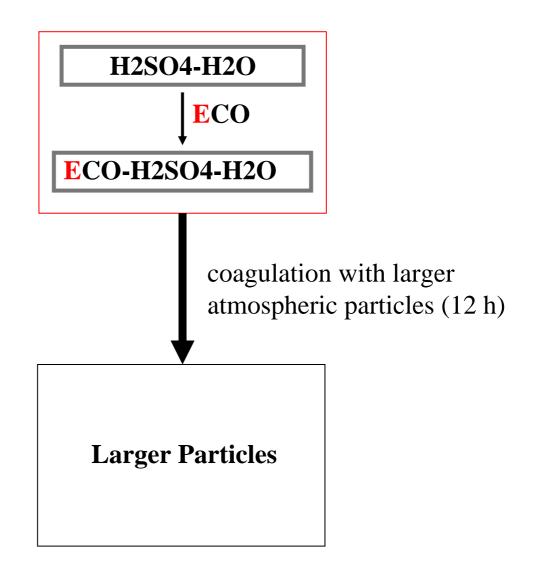


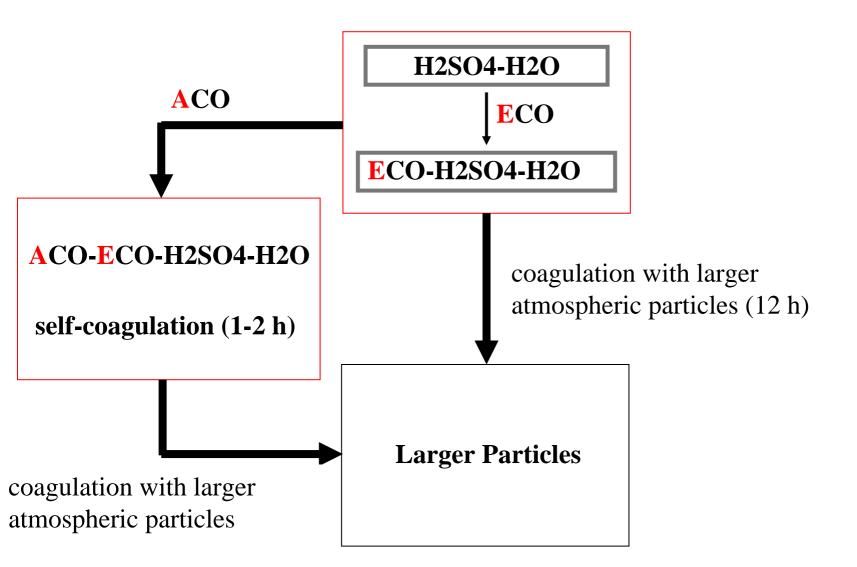


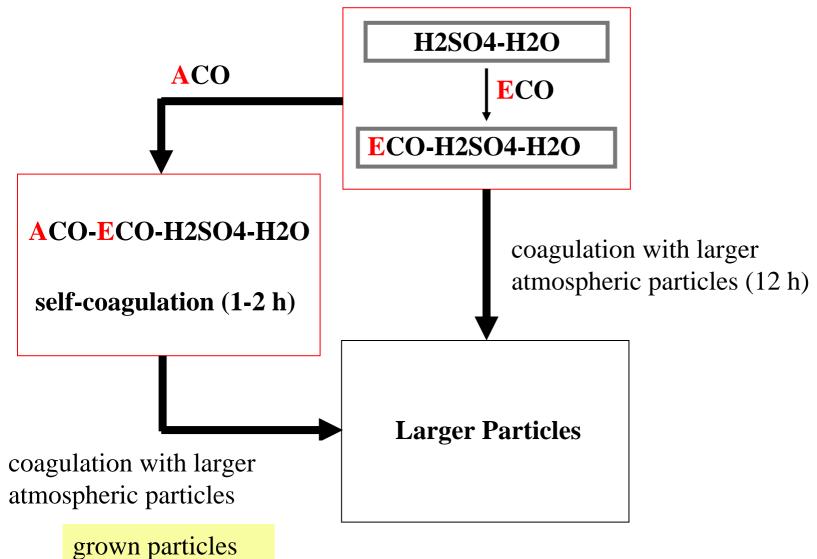


#### Fate of small particles in Atmosphere

Coagulation







live longer !

#### **Open Questions**

- Fraction F of fuel sulphur converted to gaseous H2SO4 emitted
- New particle formation by nucleation downstream of tail pipe
- New particle growth

#### Investigations of Nano-Particle Formation and Growth

- Stationary passenger car at Hyytiälä (Finland) (first test)
- Diesel engine test bed at MAN Nürnberg (Germany) (systematic measurements under well defined conditions)

#### Gaseous H2SO4 Measurements in Diesel Car Exhaust (First Test)

• Stationary experiment (low engine load EL)

#### Gaseous H2SO4 Measurements in Diesel Car Exhaust (First Test)

- Stationary experiment (low engine load EL)
- Engine frequency varied

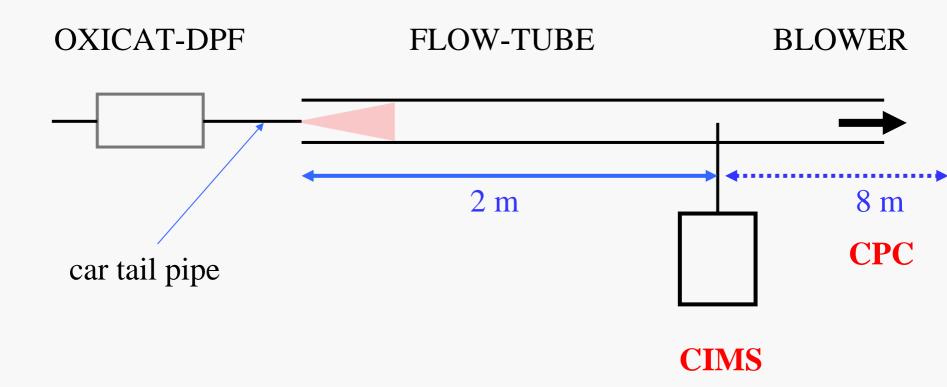
- Stationary experiment (low engine load EL)
- Engine frequency varied
- FSC=5 ppm

- Stationary experiment (low engine load EL)
- Engine frequency varied
- FSC=5 ppm
- Passenger diesel car: Peugeot 607 (year 2004) equipped with OXICAT-DPF

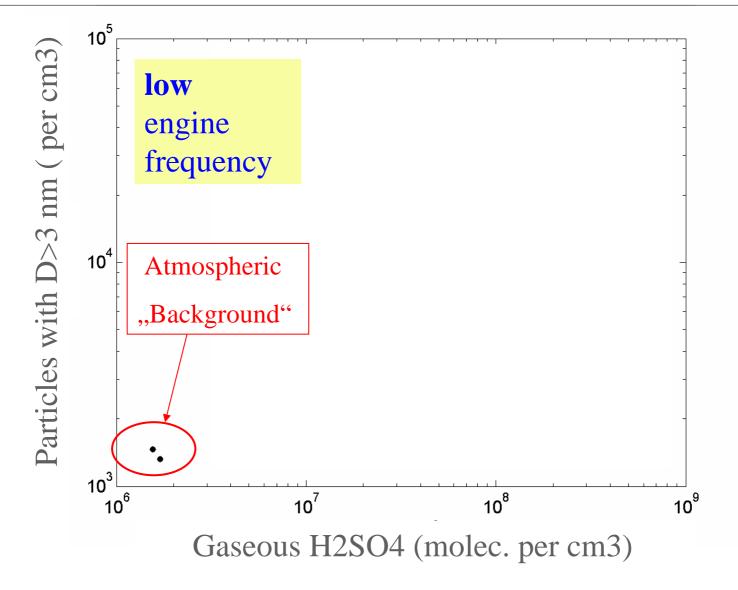
- Stationary experiment (low engine load EL)
- Engine frequency varied
- FSC=5 ppm
- Passenger diesel car: Peugeot 607 (year 2004) equipped with OXICAT-DPF
- H2SO4 measurement 2 m downstream of exhaust pipe

- Stationary experiment (low engine load EL)
- Engine frequency varied
- FSC=5 ppm
- Passenger diesel car: Peugeot 607 (year 2004) equipped with OXICAT-DPF
- H2SO4 measurement 2 m downstream of exhaust pipe
- H2SO4 measurement by ITCIMS-method originally developed by MPIK- Heidelberg

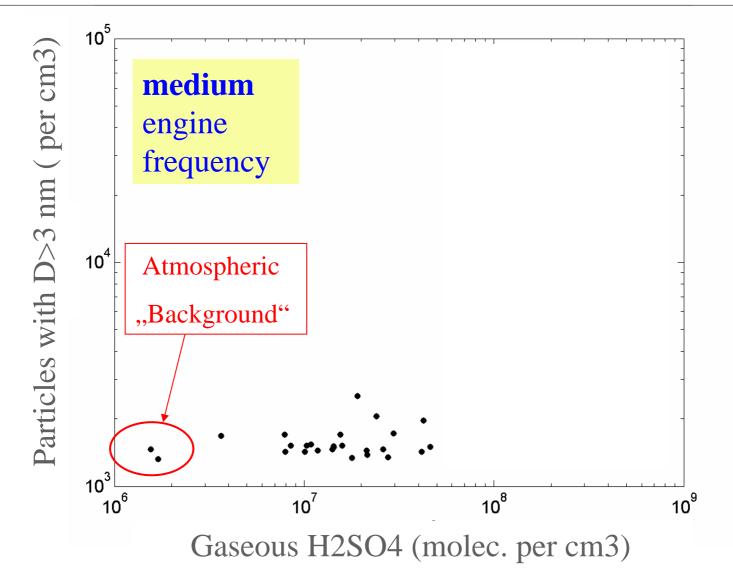
#### Stationary Diesel Car Experiment



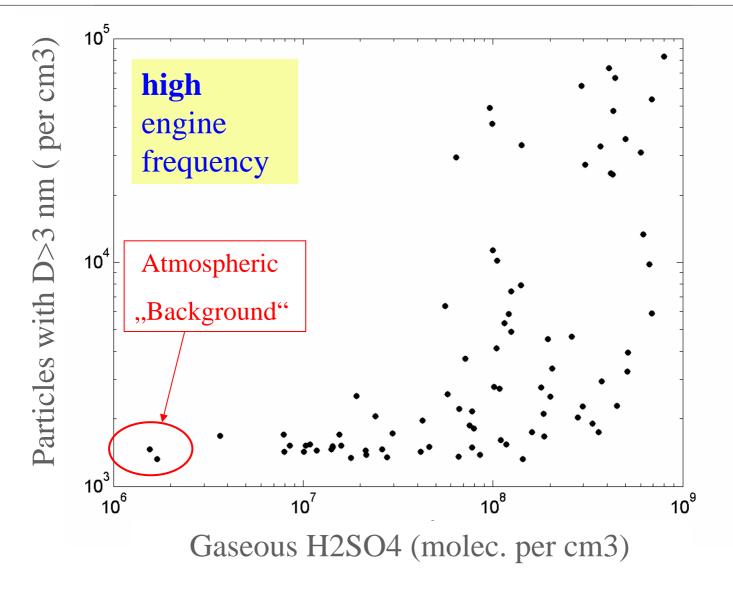
Gaseous H2SO4 measured in Stationary Experiment with modern diesel passenger car (FSC=5 ppmM); as engine frequency is increased H2SO4 increases



Gaseous H2SO4 measured in Stationary Experiment with modern diesel passenger car (FSC=5 ppmM) ; as engine frequency is increased H2SO4 increases



Gaseous H2SO4 measured in Stationary Experiment with modern diesel passenger car (FSC=5 ppmM); as engine frequency is increased H2SO4 increases



## Diesel Engine exhaust Measurements at MAN Nutzfahrzeuge (Nürnberg, Germany)

• Heavy duty diesel vehicle engine

### MAN heavy duty Diesel engine

- Type: MAN D 2066 LF 31
- Six cylinders (10.5 l) with common rail fuel injection
- Euro 4 norm
- Max power: 324 kW at 1900 rpm
- Max torque: 2100 Nm at 1000-1400 rpm

Diesel Engine exhaust Measurements at MAN Nutzfahrzeuge (Nürnberg, Germany)

- Heavy duty diesel vehicle engine
- Engine load: 25, 50, 75, 100 %
- FSC : 7 , 36 ppmM

## Diesel Engine exhaust Measurements at MAN Nutzfahrzeuge (Nürnberg, Germany)

- Heavy duty diesel vehicle engine
- Engine load: 25, 50, 75, 100 %
- FSC : 7 , 36 ppmM
- exhaust aftertreatment scenarios
  - no aftertreatment
  - OXICAT (volume=5 I, 40 g Pt/ft3)
  - CDPF (sinter material coated with Pt (20gPt/m2)

# Figures showing data of measurements at MAN

presently cannot be included since they are contained in a paper submitted to NATURE journal

### Summary

- Gaseous H2SO4 in exhaust very low when catalyst not used
- Gaseous H2SO4 in exhaust is much higher when catalyst is used
  - OXICAT :
  - CDPF :
- Volatile nano particle formation by H2SO4/H2O nucleation
- Volatile nano particle growth by condensation of H2SO4/H2O
- Additional growth by ECOM condensation on H2SO4/H2O particles

### Conclusions

- A modern diesel car equipped with a catalyst combusting low FSC (50 ppmM) fuel produces much more H2SO4 than a diesel car without catalyst combusting high FSC (300 ppmM) fuel !
- Consequently a modern diesel car with catalyst produces more volatile nano particles
- Volatile nano particles are also formed via secondary H2SO4 which stems from combustion not related to car traffic

### Acknowledgements

- Thanks to members of our MPIK Heidelberg research group
- Thanks to cooperation partners from Finland
- Thanks to MAN company
- Thanks to Dr. Jacob (MAN) for helping to plan the experiments at MAN

#### THANK YOU FOR YOUR INTEREST