



### Investigation of In-Cylinder Soot Formation and Oxidation during Transient Engine Operation

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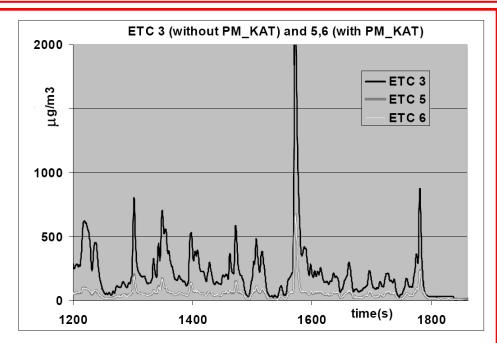
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# INTRODUCTION



Source: Schindler et al, 2004-01-0968

- Soot emissions measured using an AVL Micro Soot Sensor during ETC cycle
- Significant challenge to total emissions are "transients"

- Cycle specific characterization is necessary to understand processes
- Existing soot instrumentation is neither cylinder nor cycle specific
- 11<sup>th</sup> ETH Conference correlation of exhaust stream and in-cylinder measurements
- 12<sup>th</sup> ETH Conference Exhaust stream measurement of transient soot emissions
- Today: Combination of these two works to consider transient, incylinder processes



### OUTLINE

- Testbench and instrumentation
- Overview of in-cylinder pyrometry
- Overview of exhaust stream measurements
- Detailed analysis of in-cylinder measurements and observed phenomena

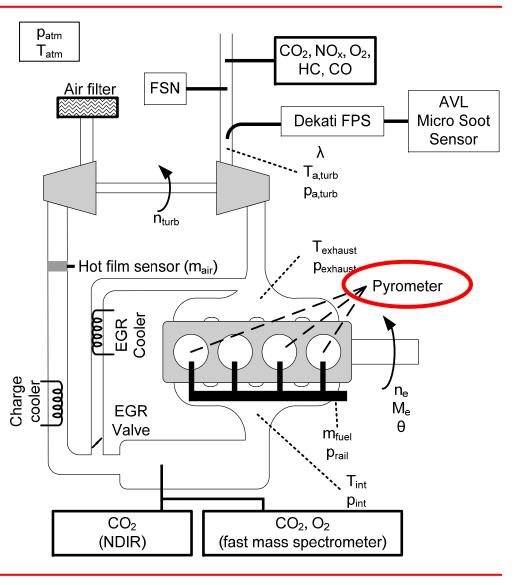
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#### Passenger car, common rail engine:

- DaimlerChrysler OM611
- VTG, EGR, p<sub>inj,max</sub>~1350 bar

#### Exhaust stream soot emissions:

- Dekati Fine Particle Sampler
- AVL Micro Soot Sensor
- Transient characterization of instrumentation (Δt, τ)
- Pyrometers mounted in cylinders 1,3 and 4 provide:
  - Soot concentration
  - Soot temperature





OM611

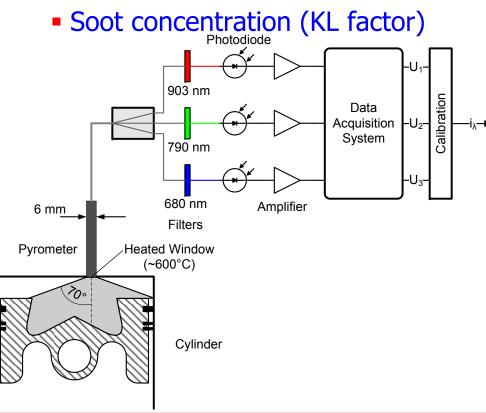
22.06.2009



### SOOT MEASUREMENT

### IN-CYLINDER PYROMETRY

- Multi-color pyrometry considers light intensity to determine incylinder:
  - Soot cloud temperature



Considers only hot ("glowing") soot
Limited to soot within field of view

$$\begin{bmatrix} 1 - \left(\frac{e^{\frac{C_2}{\lambda_1 T}} - 1}{e^{\frac{C_2}{\lambda_1 T_{s,1}}} - 1}\right) \end{bmatrix}^{\lambda_1^{\alpha}} = \begin{bmatrix} 1 - \left(\frac{e^{\frac{C_2}{\lambda_2 T}} - 1}{e^{\frac{C_2}{\lambda_1 T_{s,2}}} - 1}\right) \end{bmatrix}^{\lambda_2^{\alpha}} \\ T\Big|_{\lambda_1,\lambda_2} = T\Big|_{\lambda_1,\lambda_3} = T\Big|_{\lambda_2,\lambda_3}$$

$$KL = -\lambda^{1.39} \ln \left[ 1 - \left( \frac{e^{\frac{C_2}{\lambda T}} - 1}{\frac{C_2}{e^{\lambda T_s}} - 1} \right) \right]$$

$$KL\Big|_{\lambda_1} = KL\Big|_{\lambda_2} = KL\Big|_{\lambda_3}$$

Hottel and Broughton. Ind. Eng. Chem., 1932. 4(2)

# **3 COLOR PYROMETRY**

- System initially developed by LAV<sup>1</sup>; later in conjunction with Kistler AG and Sensoptic<sup>2</sup>
- Uses 3 wavelengths for redundancy (T, KL cross-verification)
- Wide field of view (140°) considers "most" of the cylinder
- Window heated to ~600°C to prevent contamination and provide long-term signal stability
- Very small size permits use in production engines (glowplug adapter, for eg.)

<sup>1</sup> R. Schubiger *et al.* **MTZ**, 2002. **5**(63):342-353 <sup>2</sup> S. Kunte *et al.* **KTI Technical Report**, 2005.



IMPLEMENTED SENSOR

LAV /



 $d_{Sensor} = 4mm$ 

**KISTLER** 

measure. analyze. innovate.



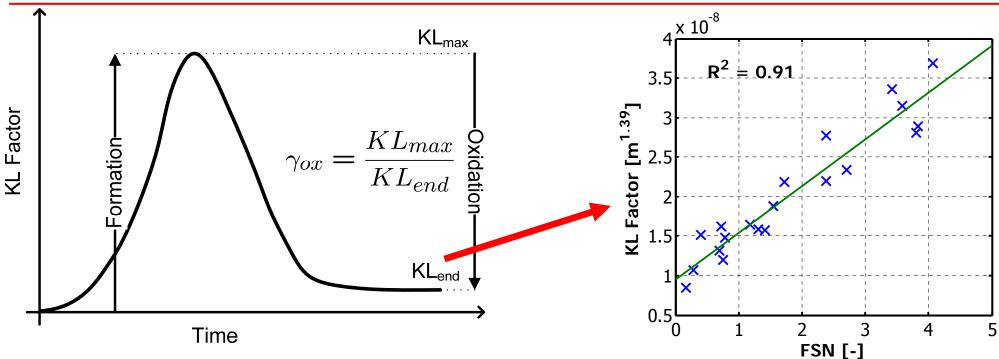


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**USEFUL PARAMETERS** 

### **3 COLOR PYROMETRY**



- KL<sub>end</sub> correlates with exhaust stream measurements (R<sup>2</sup> ~0.8...0.9)
- Cycle resolved engine out soot emissions

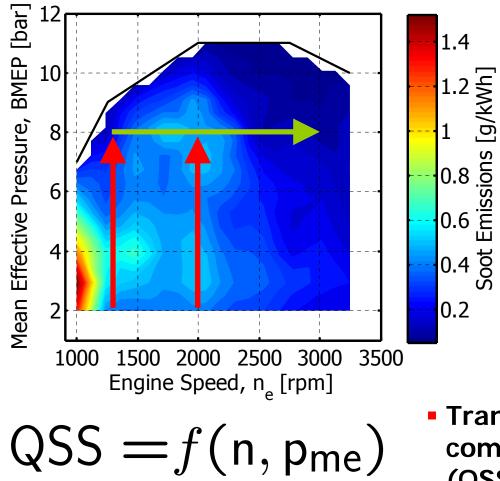
- KL<sub>max</sub> measure of soot formation
- γ<sub>ox</sub> measure of soot oxidation
- Relative characterization of formation and oxidation processes

Kirchen et al. Proc. Stuttgart Symposium, 2008. 2:129-145

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### TRANSIENT MEASUREMENTS CONSIDERED TRANSIENTS



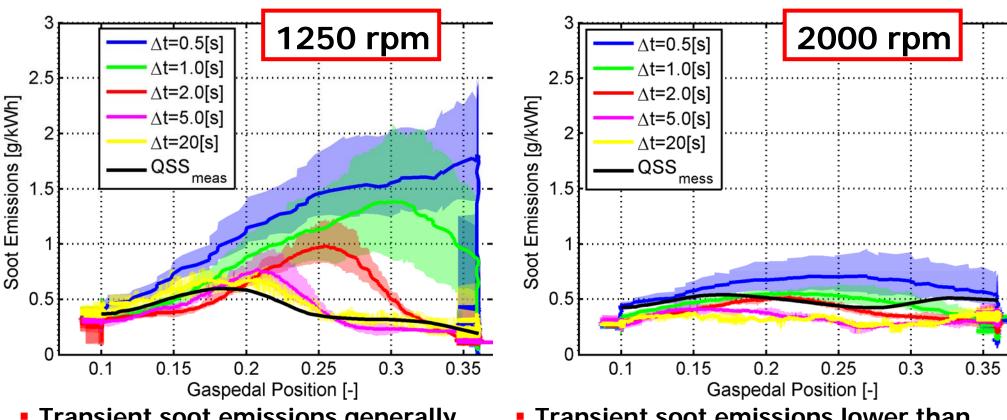
- Acceleration (speed increase)
  - No notable change over steadystate
- Tip-in (load increase)
  - 1250 and 2000 rpm
  - Δt=0.5...5 s

 Transient and steady-state emissions compared using a Quasi Steady State (QSS) approximation<sup>1</sup>

<sup>1</sup>Hagena et al. SAE 2006-01-1151, 2006

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### **TRANSIENT SOOT MEASUREMENTS**



- Transient soot emissions generally higher than steady-state (QSS)
- Faster transients result in much higher emissions

- Transient soot emissions lower than at 1250 rpm
- Only the fastest transient results in increased soot emissions

Kirchen and Boulouchos. SAE, 2009. 2009-01-1904

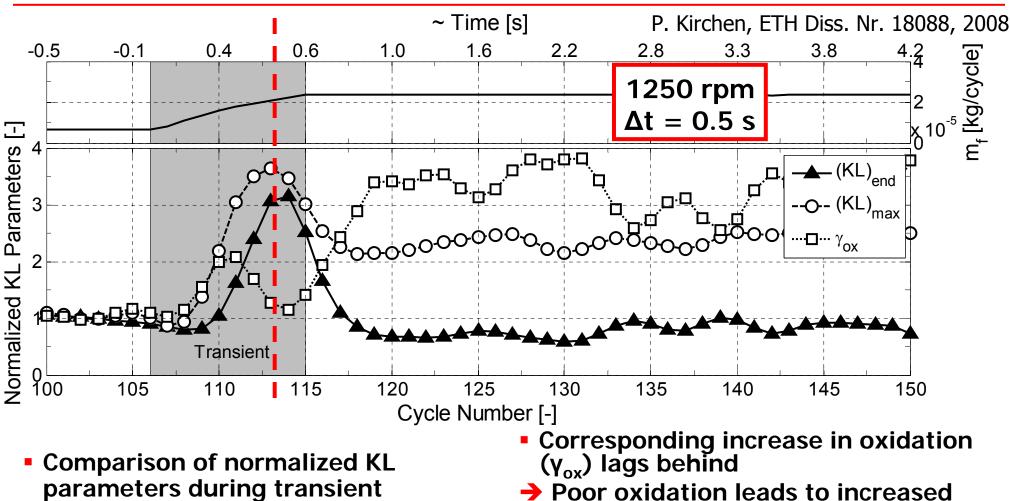
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TIP-IN

# 

TIP-IN TRANSIENT (1250 rpm)

### **IN-CYLINDER PYROMETRY**



- Increase in fuel quantity -> increase in KL<sub>max</sub>
- What causes the poor oxidation?

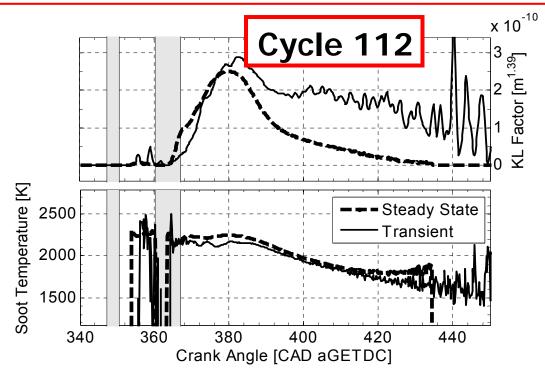
engine-out emissions (KL<sub>end</sub>)

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### PYROMETRY (KL & T)

### TRANSIENT vs. STEADY-STATE



- KL<sub>max</sub> not strongly influenced by transient operation
- Oxidation considerably slower during transient operation and stops earlier
- No significant differences between steady state and transient soot temperatures
- $\rightarrow$  Oxidation inhibited due to lack of O<sub>2</sub>

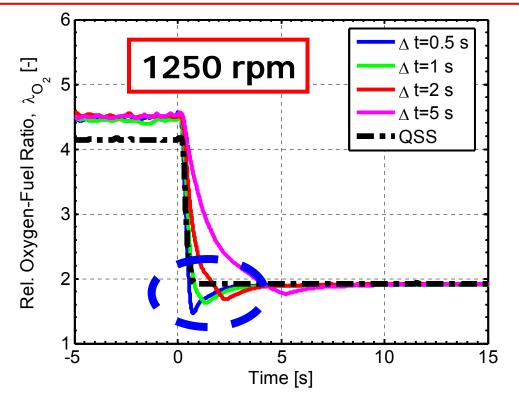
P. Kirchen, ETH Diss. Nr. 18088, 2008





### **OXYGEN AVAILABILITY**

### TIP-IN TRANSIENT (1250 RPM)



- Short term oxygen deficit caused by:
  - Slow EGR valve closing
  - Slow increase in charge pressure
  - Rapid increase in fuel quantity

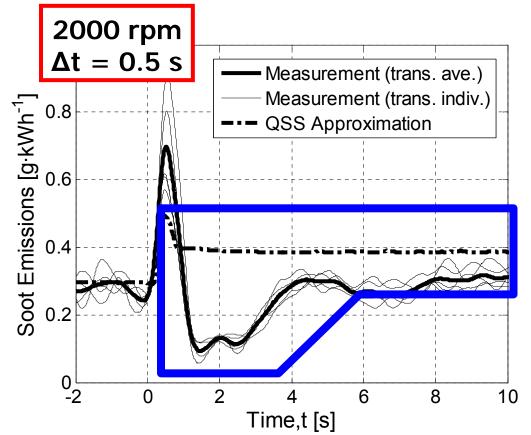
### **POST-TRANSIENT PHENOMENA**

TIP-IN AT 2000 RPM

LAV

• After transient:

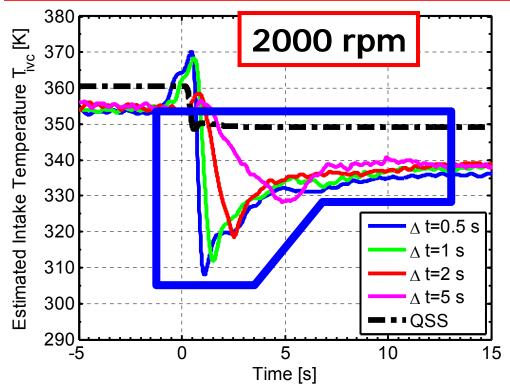
- Transient soot emissions are lower than steady state
- Only gradually increase and reaproach steady state value (~60s)
- Phenomena correlates with a gradual increase in intake charge temperature
- Mechanism for reduction of engine-out emissions is unclear...



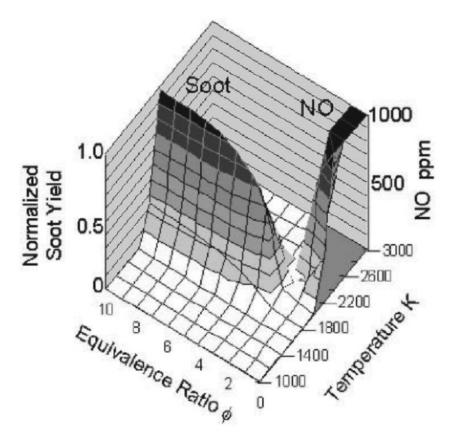
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### **T DURING TIP-IN TRANSIENTS**



- Lower temperature after transient, when compared to QSS
- Intake charge temperature provides estimate of soot formation temperature



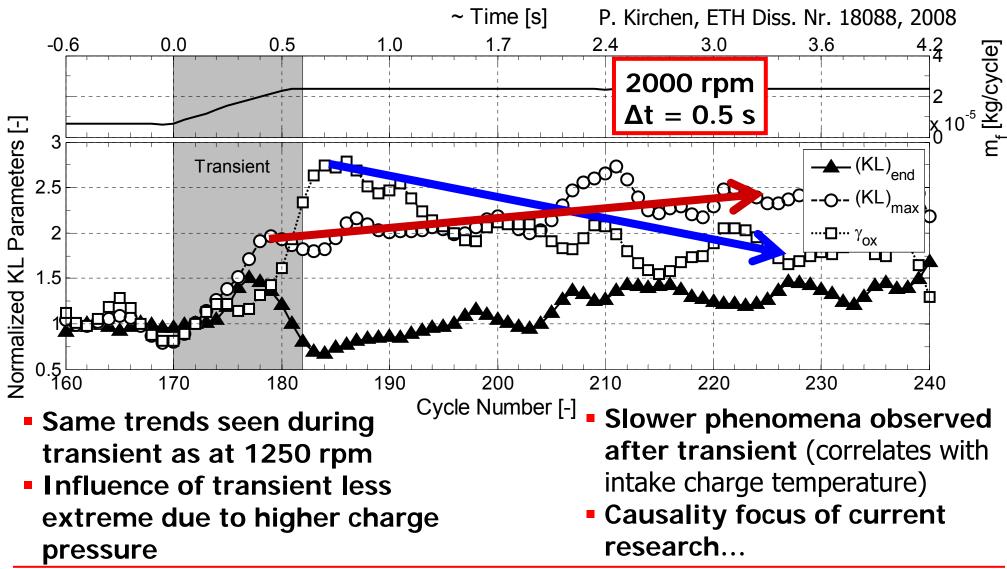
Akihama *et al*. **SAE** 2001-01-0655, 2001

 Reduction of formation temperature results in lower soot emissions

# LAV 🔨

### **IN-CYLINDER PYROMETRY**







### CONCLUSIONS

- Multicolor pyrometry is a powerful tool for the measurements of cylinder and cycle specific, in-cylinder soot concentration and temperature
- During tip-in transients:
  - Soot formation is approximately the same as during steady-state (KL<sub>max</sub>)
  - Soot oxidation is weaker due to an oxygen deficit ( $\gamma_{ox}$ )
- Only 5-10 cycles responsible for high engine out soot emissions
- Slow increase in intake charge temperature after transient results in:
  - Gradual increase in engine-out soot emissions to final steady-state values
  - Gradual increase in KL<sub>max</sub> to steady state value
- Precise influence of charge temperature is not yet completely understood ...



# THANK YOU FOR YOUR ATTENTION!!

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