The chemical tiger in Diesel car exhaust: Search for fluorine-bearing highly acidic gases

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We report on a search for fluorine-bearing highly acidic and corrosive trace gases in Diesel car exhaust. Using an innovative, very sensitive and fast chemical ionization mass spectrometric method, we have made on-line and off-line measurements of highly acidic exhaust gases. The detected gases include sulphuric acid and several species, which most likely contain fluorine. Some of the F-bearing species also contain sulphur. We have also made a search for the source of exhaust fluorine. In commercial Diesel fuel we found highly variable F-mass fractions of up to 114 ppmm and in lubricant oil up to 30 ppmm. The detected Fbearing exhaust gases not only are highly acidic and corrosive, but include also ones having a highly toxic nature. One of the detected F-bearing gases has a very low volatility and therefore may become incorporated into exhaust nanoparticles. It is at least conceivable that exhaust fluorine may also undergo chemical conversion to toxic F-bearing dioxins and furans, but these are less stable than their chlorinated counterparts and may thermally decompose at typical exhaust temperatures. Presently fuel-fluorine is not explicitly regulated. However, we have found that certain catalytic Diesel particle filters, at least partially, remove the detected F-bearing highly acidic gases. Considering their corrosive and toxic nature, F-bearing exhaust species deserve increased future attention the chemistry literature Fluorine is often

- termed "The Tiger of Chemistry" since it is extremely reactive Due to ist large electronegativity, F tends to form very strong acids
- Some contain also S besides F
- Some have an extremely corrosive and toxic nature
- However, no information available on F-species in motorcar exhaust and fuel!

Screening of international literature

- German Federal Environment Agency
- German Fuel producers
- We have made a search for F in exhaust
- We have made a search for F in motor-car fuels and lubricant oils
- F-acids have been detected in exhaust
- Some undergo co-adsorption with H2SO4
- Likely F-source seems to be the fuel
- is F a Diesel OXICAT poison?
- ATS efficiently removes organic F-species
- Is F contained only in fuels from certain refineries?
- Future fuel F-analyses are needed.

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Search for fluorine-bearing highly acidic gases

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Abstract

We report on a search for fluorine-bearing highly acidic and corrosive trace gases in Diesel car exhaust. Using an innovative, very sensitive and fast chemical ionization mass spectrometric method, we have made on-line and offline measurements of highly acidic exhaust gases. The detected gases include sulphuric acid and several species, which most likely contain fluorine. Some of the F-bearing species also contain sulphur. We have also made a search for the source of exhaust fluorine. In commercial Diesel fuel we found highly variable F-mass fractions of up to 114 ppmm and in lubricant oil up to 30 ppmm. The detected F-bearing exhaust gases not only are highly acidic and corrosive, but include also ones having a highly toxic nature. One of the detected Fbearing gases has a very low volatility and therefore may become incorporated into exhaust nanoparticles. It is at least conceivable that exhaust fluorine may also undergo chemical conversion to toxic F-bearing dioxins and furans, but these are less stable than their chlorinated counterparts and may thermally decompose at typical exhaust temperatures. Presently fuel-fluorine is not explicitly regulated. However, we have found that certain catalytic Diesel particle filters, at least partially, remove the detected F-bearing highly acidic gases. Considering their corrosive and toxic nature, F-bearing exhaust species deserve increased future attention.

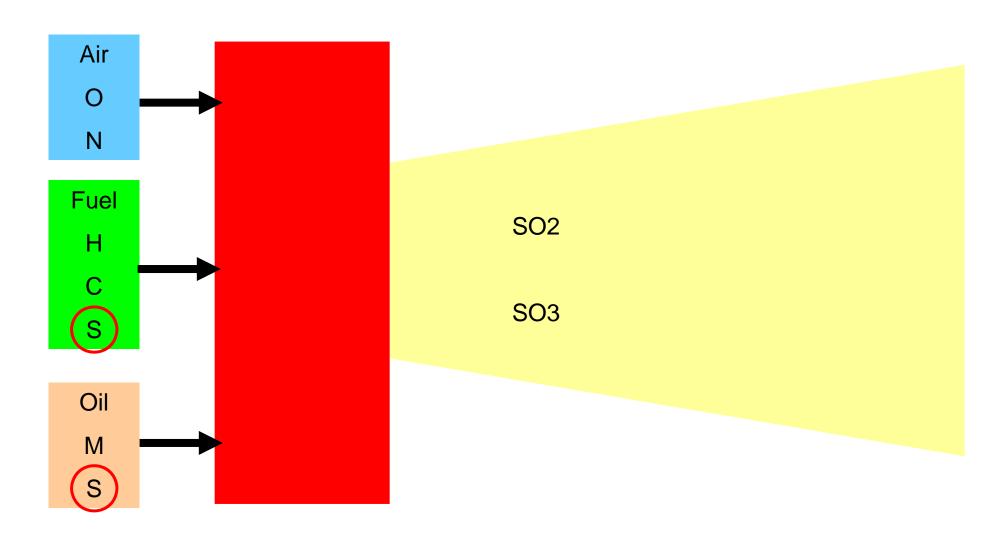
Background

- Motor-car exhaust still represents a major air pollution source in cities and near motorways
- Contains numerous combustion generated species (gases, nanoparticles)
- Some are related to chemical elements contained in the fuel only in ultra-trace mass amounts

Example

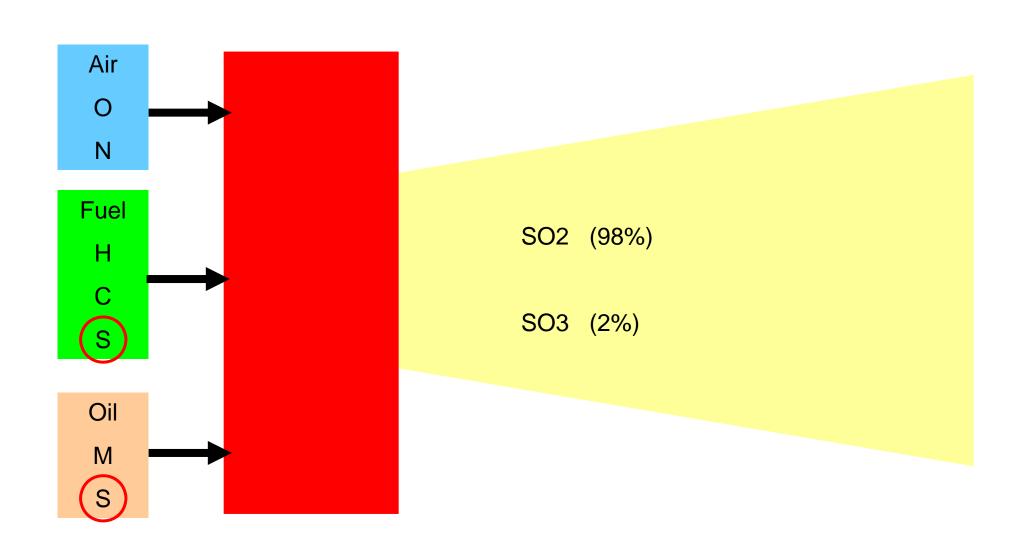
Sulphur (S)

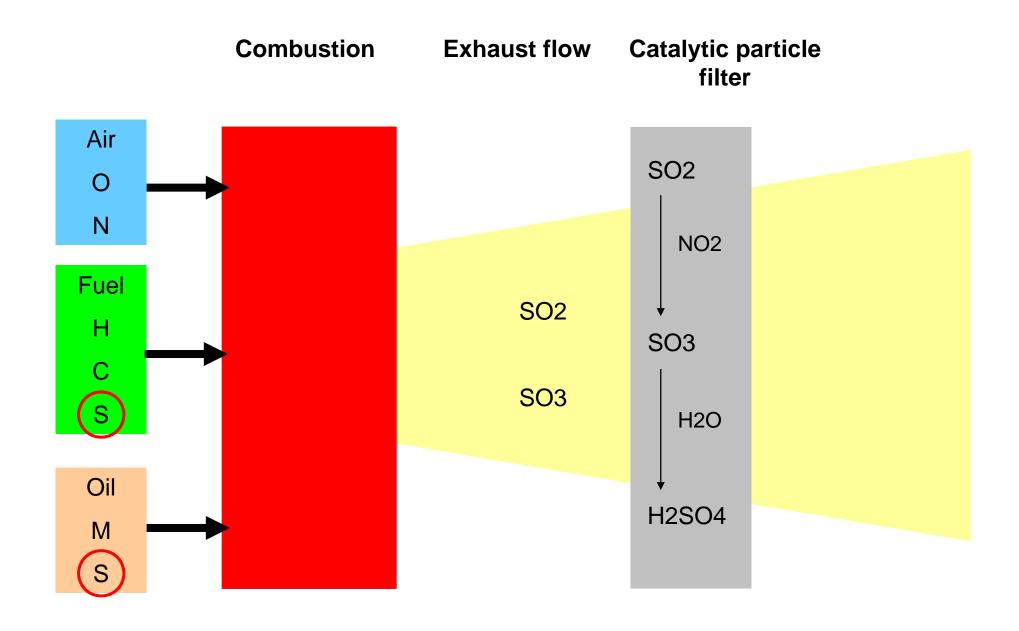
Combustion Exhaust flow

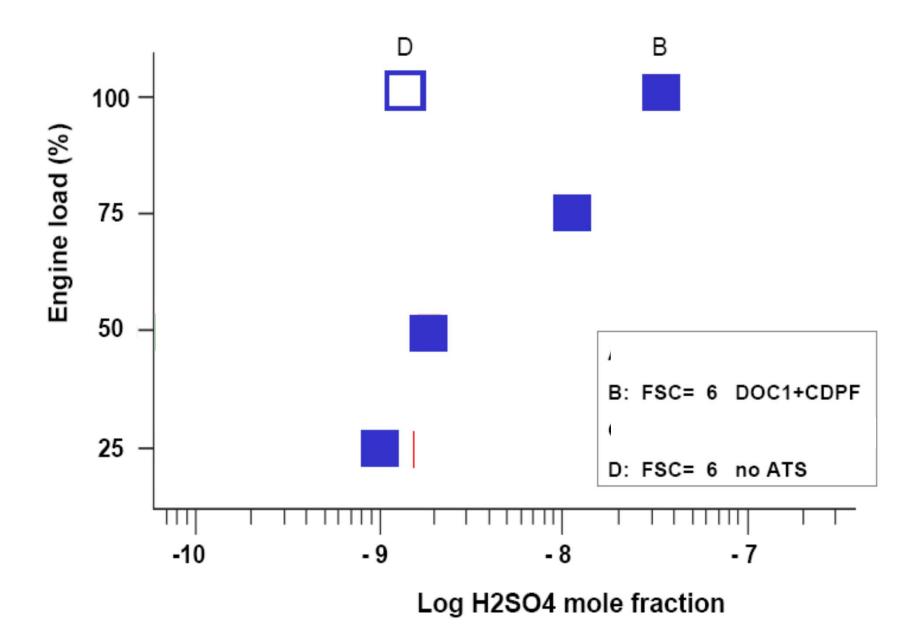


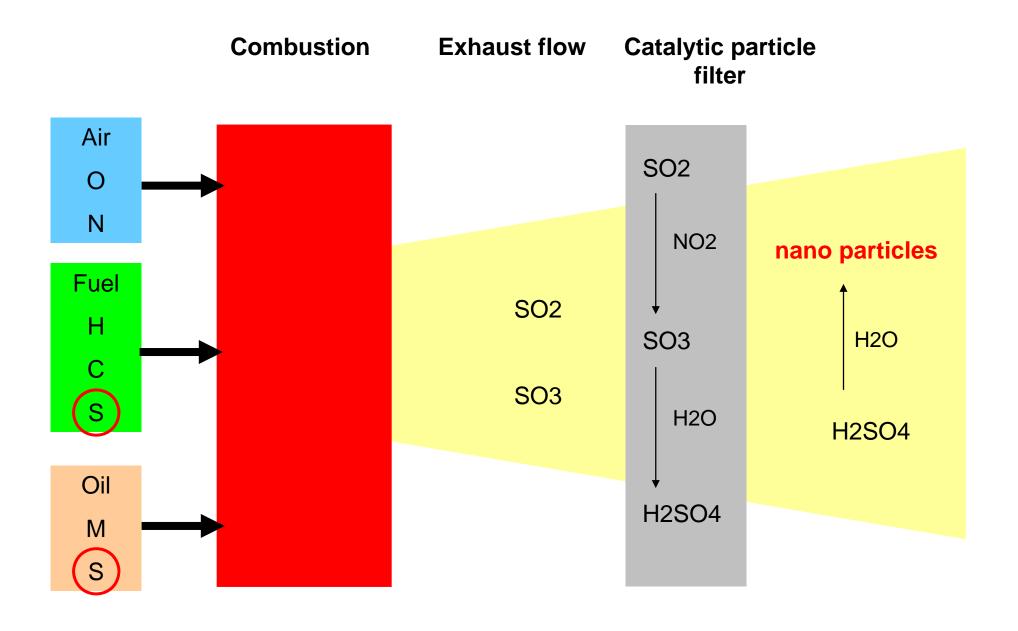
Combustion

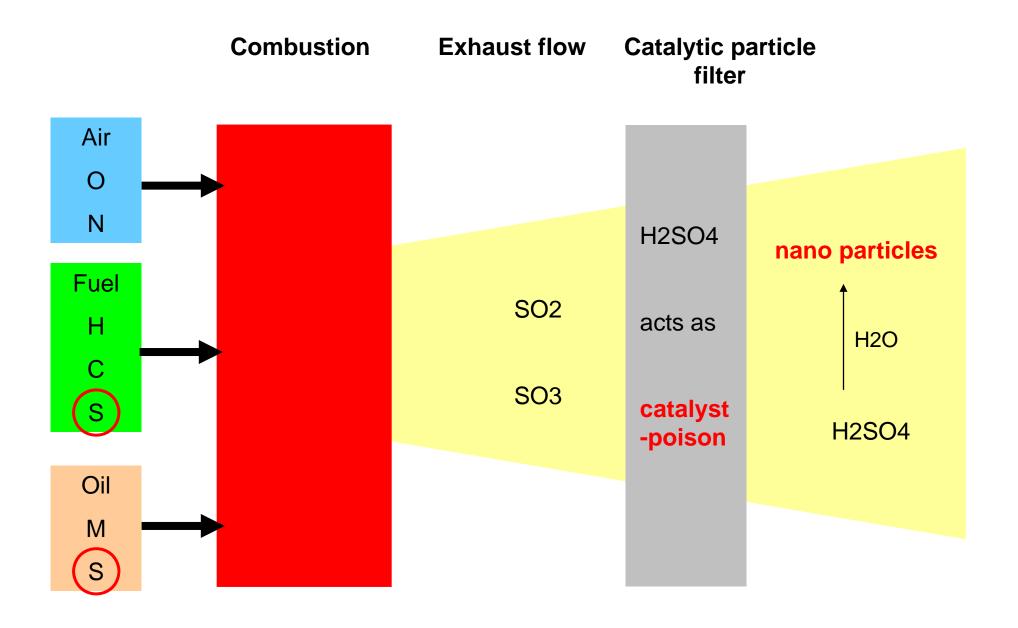
Exhaust flow







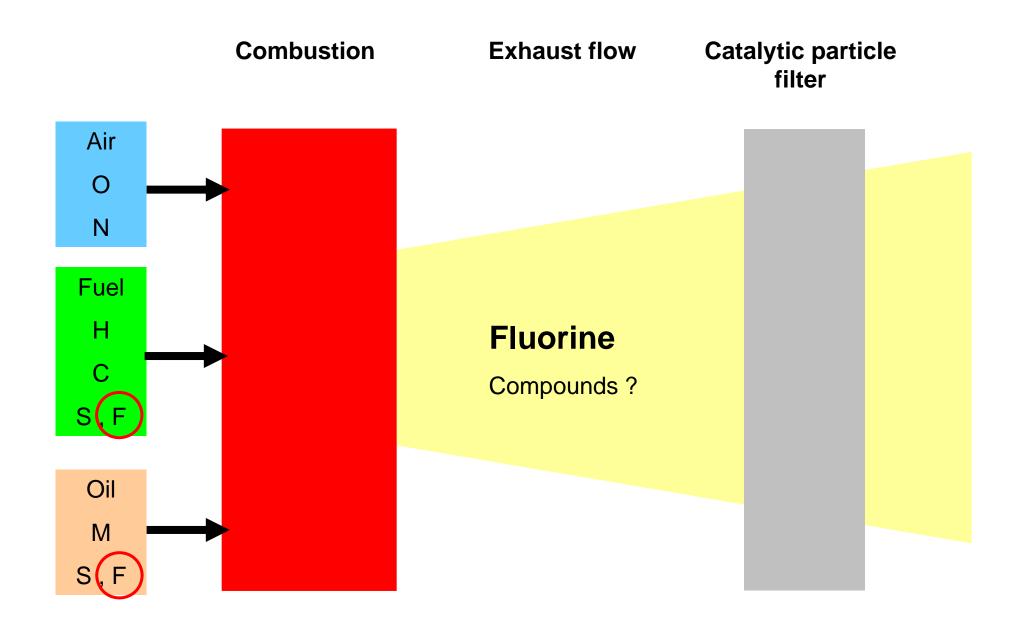




Example

Fluorine (F)

(the "Tiger of chemistry")



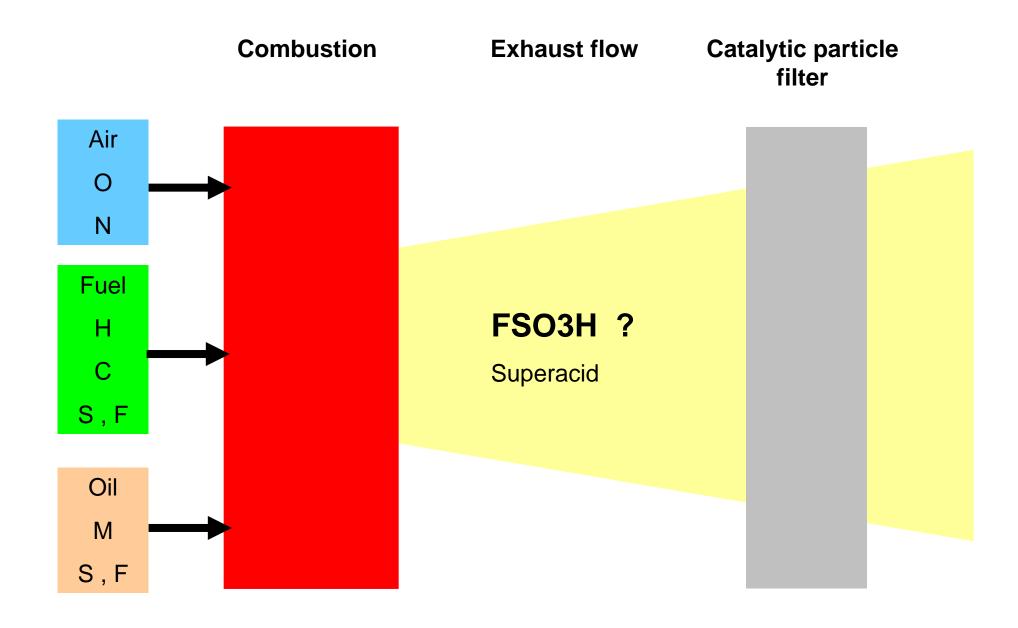
Fluorine

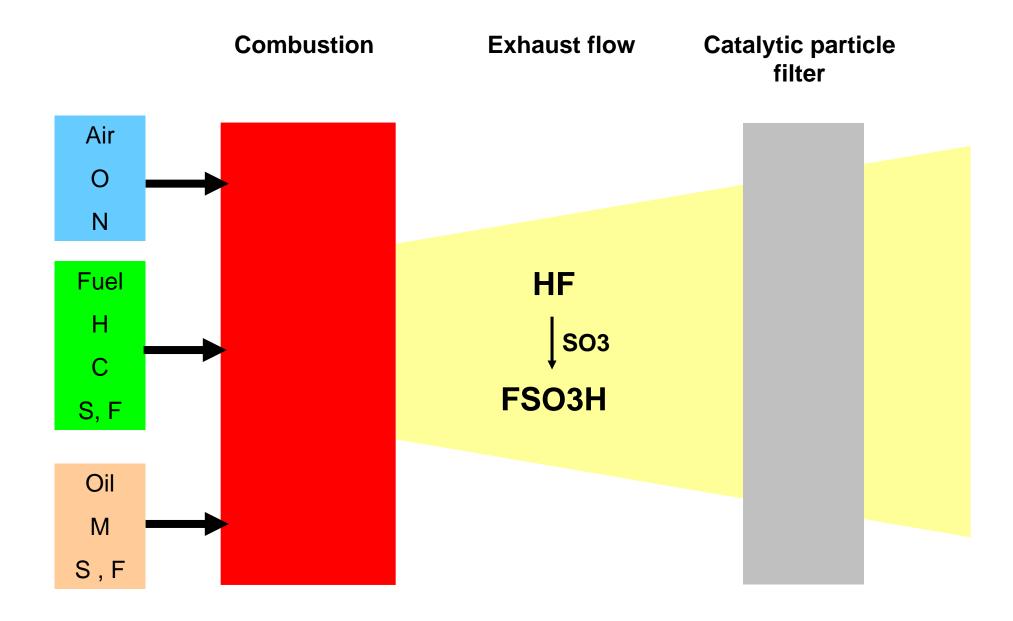
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- Due to ist large electronegativity, F tends to form very strong acids
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Example

of acid bearing F and S

FSO3H





FSO3H

- Superacid
- highly toxic and highly corrosive
- hydrolyzes to release HF
- H2O + FSO3H is violent, similar to H2O + H2SO4
- FSO3H + H2O much more violent than H2SO4 + H2O every single FSO3H - drop makes a little explosion

Conclusions

- F-acids have been detected in exhaust
- Some undergo co-adsorption with H2SO4
- Likely F-source seems to be the fuel
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- ATS efficiently removes organic F-species
- Future fuel F-analyses are needed. Is F contained only in fuels from certain refineries?

Acknowledgements

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 - Helsinki
 - Tampere
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Thank you for your interest