
A Thermophoretic-Thermocouple Method for Soot Measurement in Combustion

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di Napoli Federico II



Istituto di Ricerche
sulla Combustione

Introduction and motivations

SOOT FORMATION IN FUEL-RICH FLAME CONDITIONS



Wide range of molecular weights and sizes



**Nascent Organic
Carbon Nanoparticles**
(2-5 nm)

**Soot
Nanoparticles**
(10-100 nm)

Introduction and motivations

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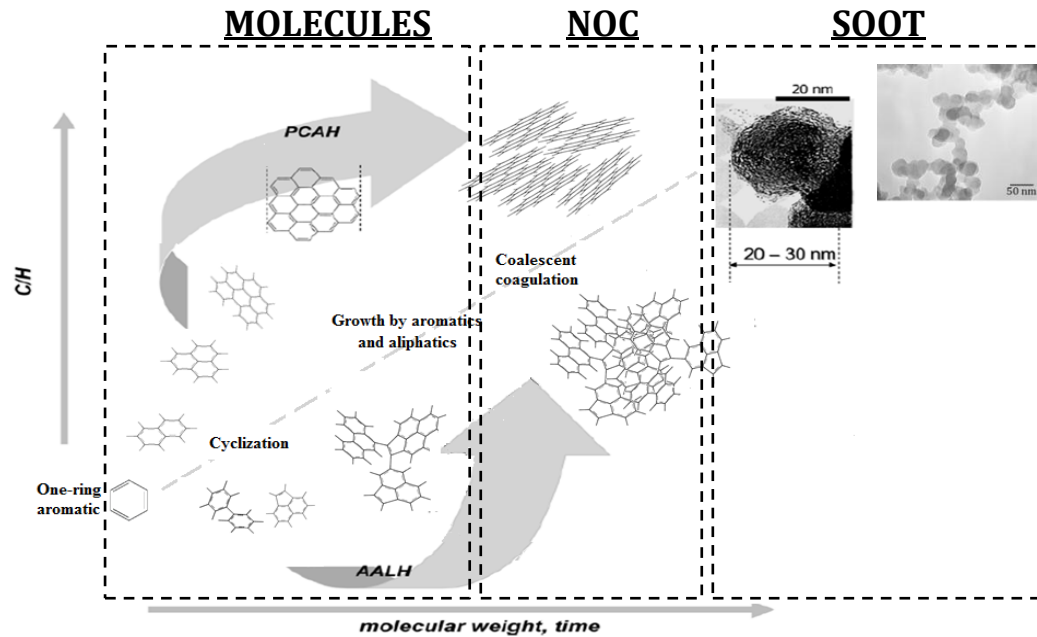
**Soot
Nanoparticles**
(10-100 nm)

Differentiate by:

- ✓ Size and morphology
- ✓ Chemical composition
- ✓ Optical and spectroscopy properties
- ✓ H/C ratio (organic carbon vs. black carbon)
- ✓ Physical status

Introduction and motivations

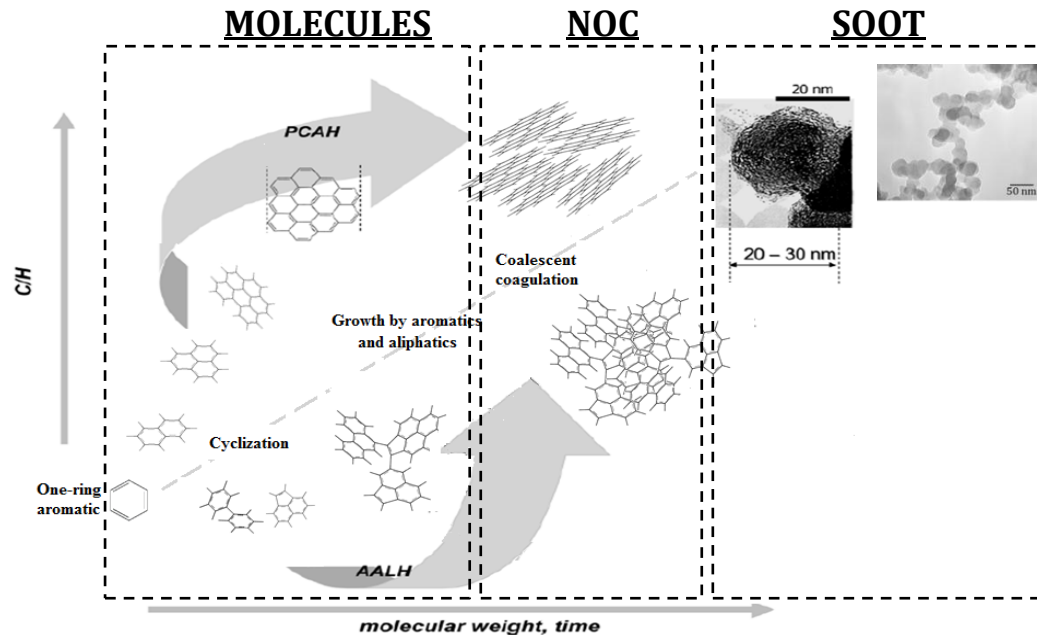
SOOT FORMATION IN FUEL-RICH FLAME CONDITIONS



A. D'Anna, in *Combustion Generated Fine Carbonaceous Particles* (2009)

Introduction and motivations

SOOT FORMATION IN FUEL-RICH FLAME CONDITIONS



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Diagnostic methods

QUANTITATIVE INFORMATION
(Particle concentration)

QUALITATIVE INFORMATION
(Particle nature and composition)

OBJECTIVE OF THIS WORK

To develop and apply a diagnostic method able to measure *in-situ* in combustion systems:

- Thermal emissivity of carbon nanoparticles
(QUALITY)
- Particle volume fraction
(QUANTITY)

TPD-Based Thermophoretic-Thermocouple Method

Thermocouple Particle Densitometry (TPD)



Combustion and Flame

Volume 109, Issue 4, June 1997, Pages 701-720



Soot volume fraction and temperature measurements in laminar nonpremixed flames using thermocouples

Charles S. McEnally , Ümit Ö. Köylü, Lisa D. Pfefferle, Daniel E. Rosner

Department of Chemical Engineering and Center for Combustion Studies, Yale University, New Haven, CT 06520-8286 USA

Available online 8 May 1998

 Show less

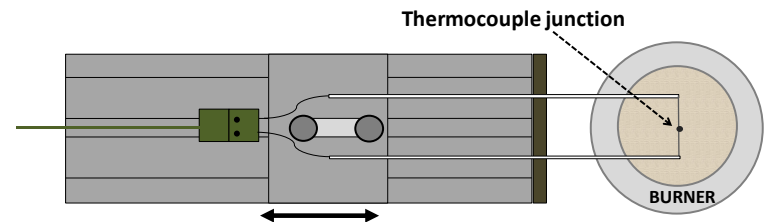
doi:10.1016/S0010-2180(97)00054-0

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Abstract

Thermocouple particle densitometry (TPD), a new method for measuring absolute soot volume fraction in flames which was suggested by Eisner and Rosner, has been successfully implemented in several laminar nonpremixed flames. This diagnostic relies on measuring the junction temperature history of a thermocouple rapidly inserted into a soot-containing flame region, then optimizing the fit between this history and one calculated from the principles of thermophoretic mass transfer. The TPD method is very simple to implement experimentally, yields spatially resolved volume

RAPID INSERTION PROCEDURE



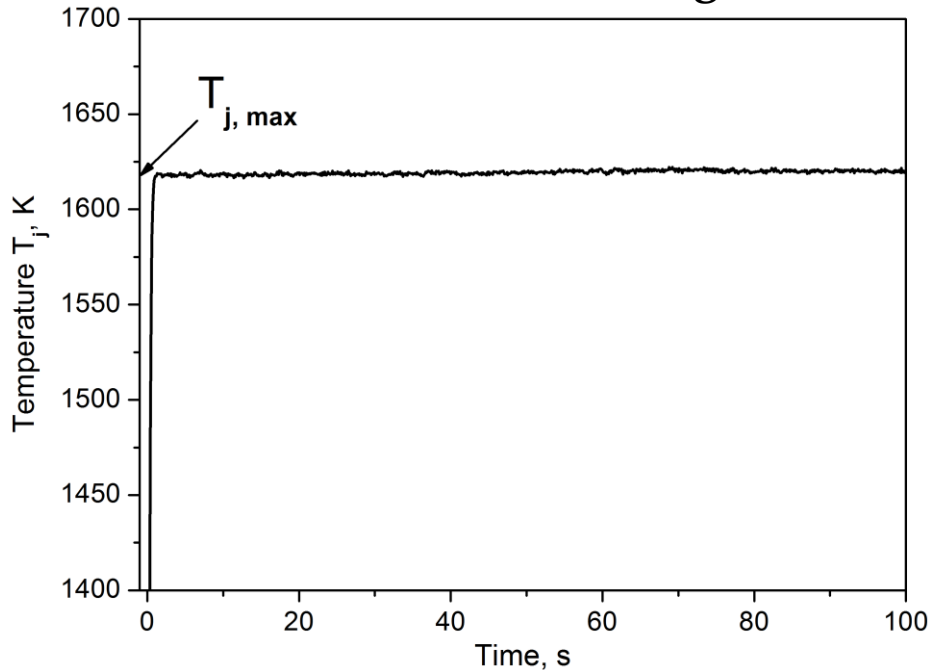
C.S. McEnally et al. *Combust. Flame* **109** (1997)

A.D. Eisner and D.E. Rosner *Combust. Flame* **61** (1985)

TPD-Based Thermophoretic-Thermocouple Method

GAS TEMPERATURE DETERMINATION

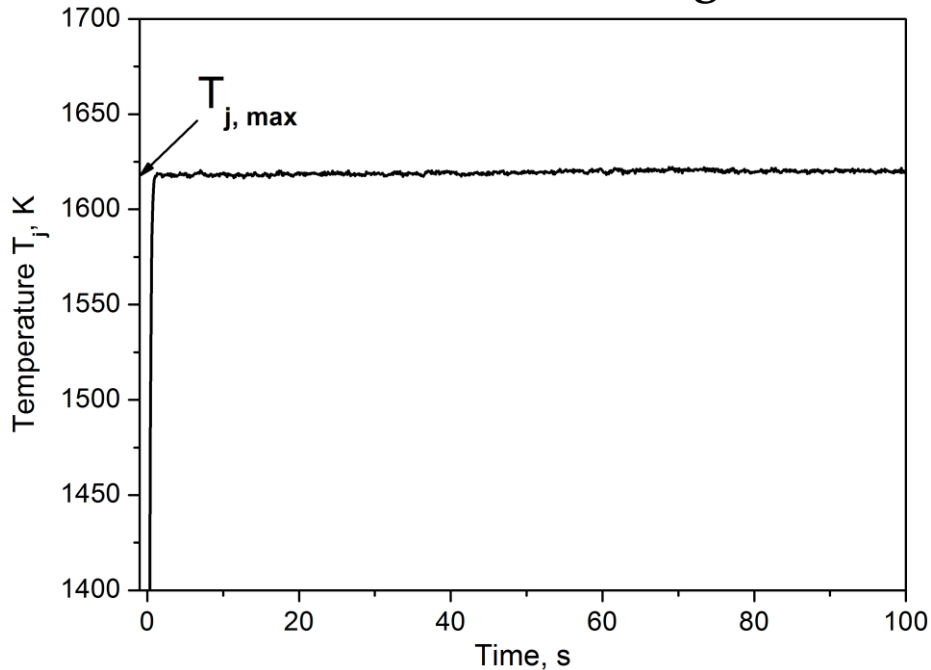
Particle-free flame region



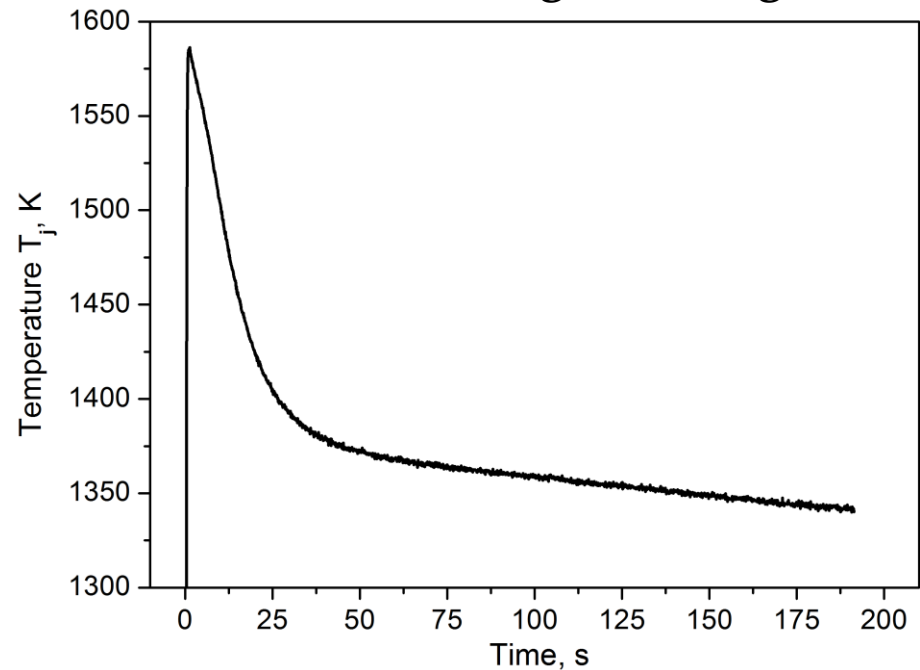
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GAS TEMPERATURE DETERMINATION

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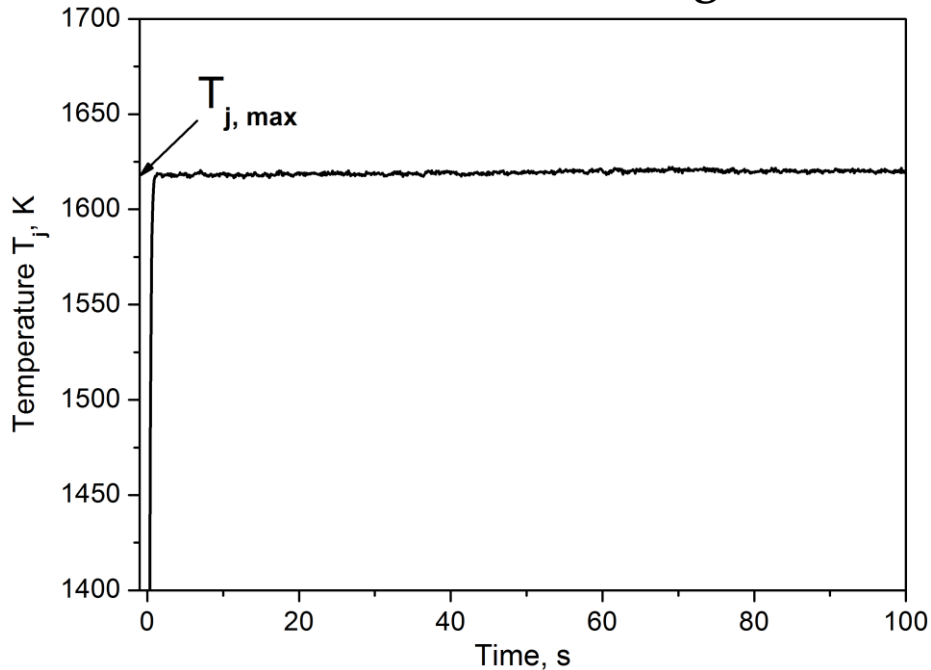
Soot-containing flame region



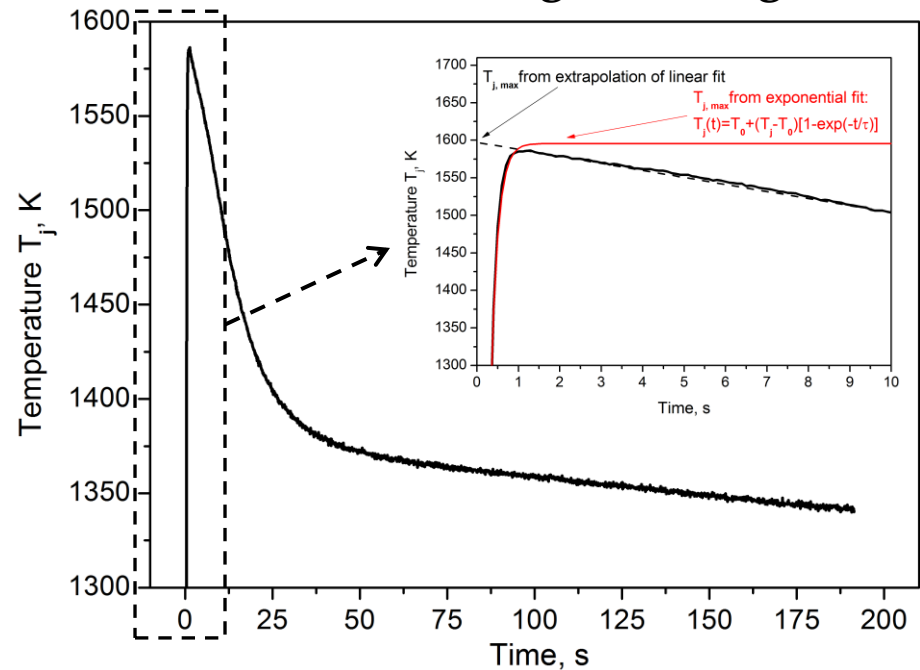
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Soot-containing flame region

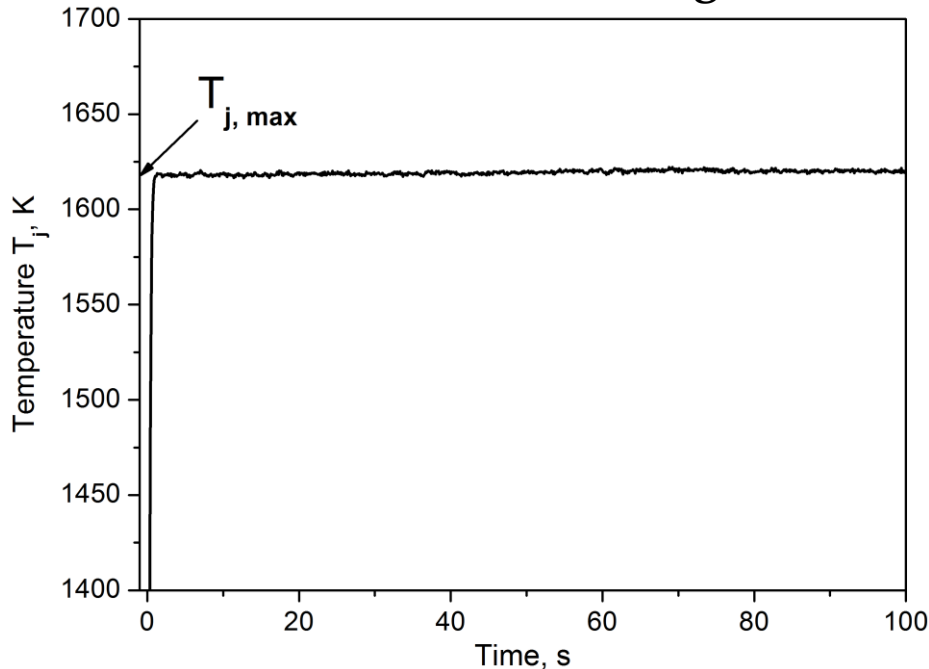


C.S. McEnally et al. *Combust. Flame* **109** (1997)
Z. Xu and H. Zhao *Combust. Flame* **162** (2015)

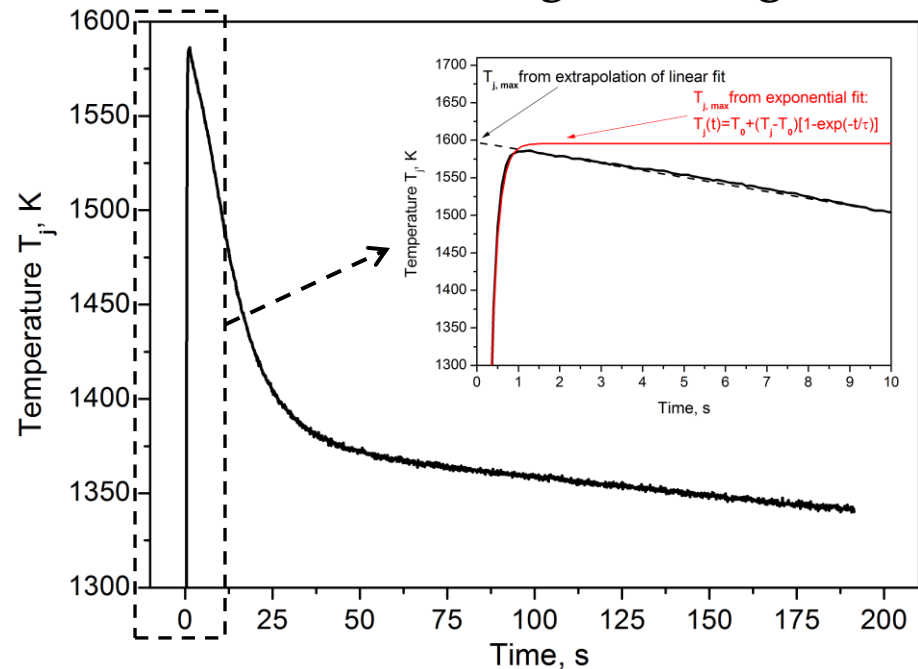
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C.S. McEnally et al. *Combust. Flame* **109** (1997)
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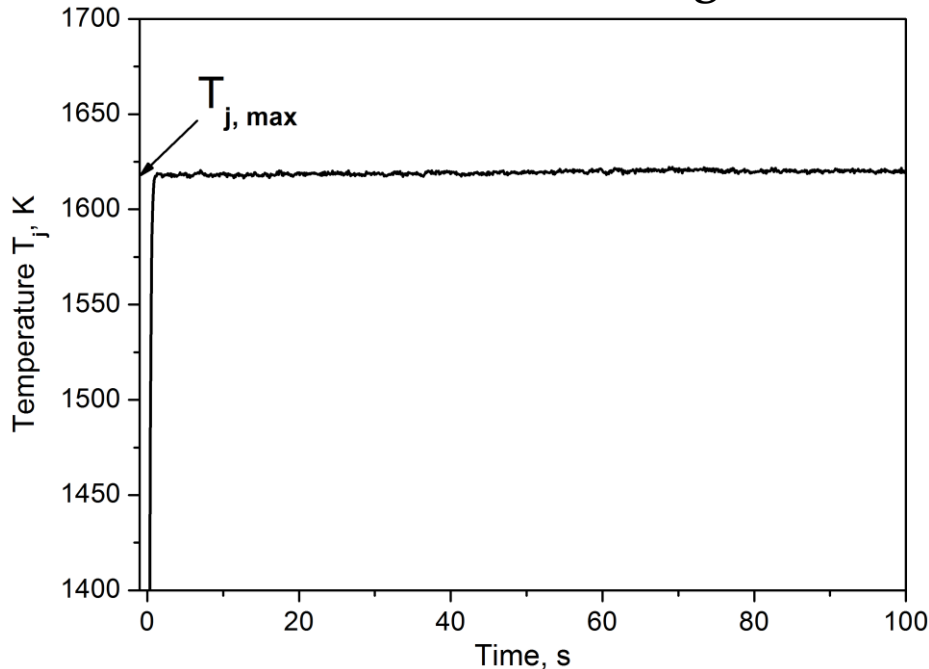
Energy balance at the thermocouple junction

$$\varepsilon_j \sigma T_{j, \max}^4 = \left(\frac{K_{g0} Nu_j}{2d_j} \right) (T_g^2 - T_{j, \max}^2)$$

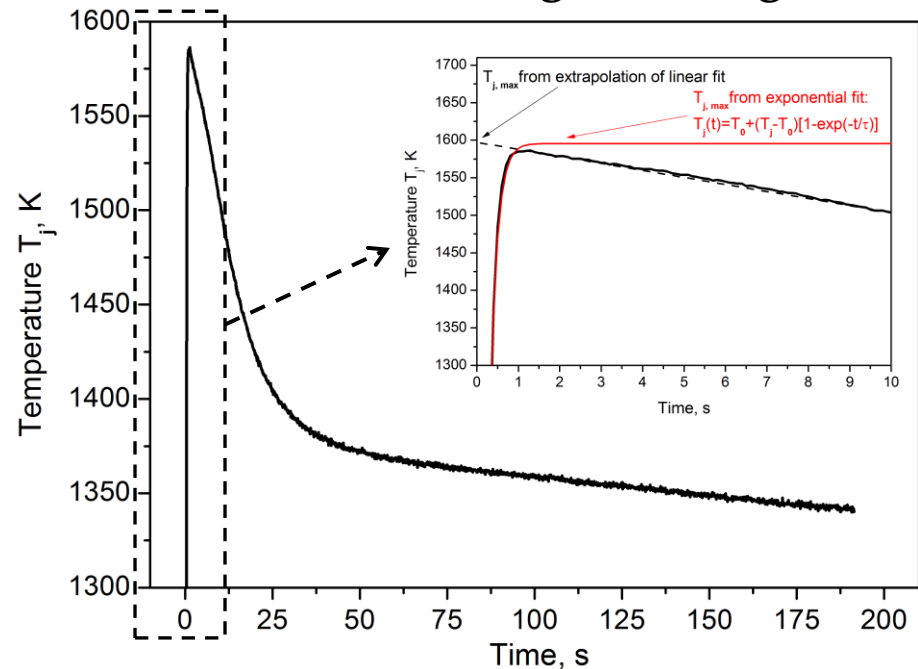
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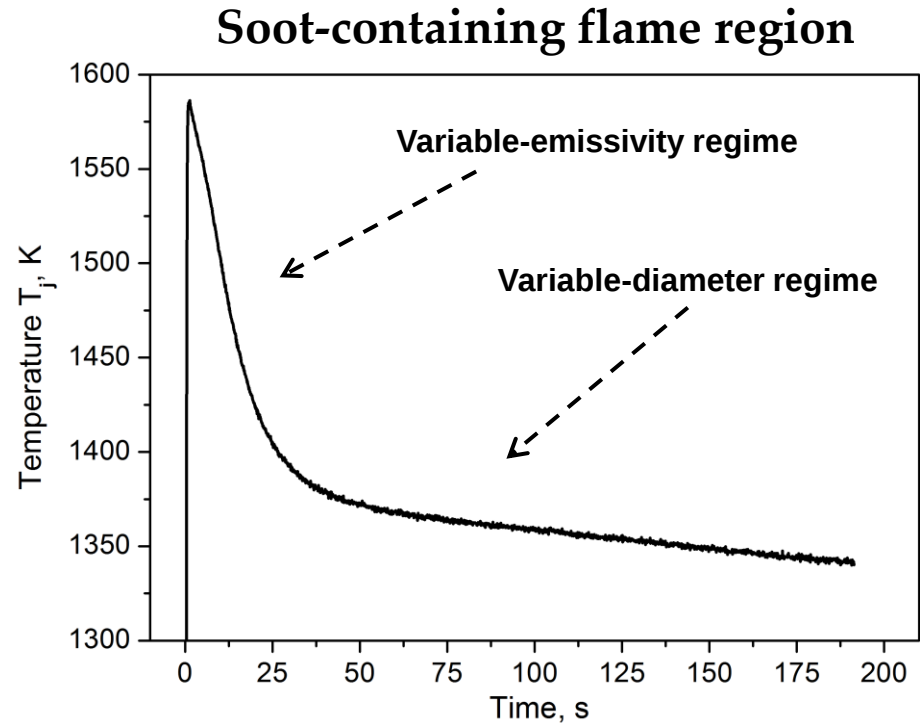
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Energy balance at the thermocouple junction

$$\varepsilon_j \sigma T_{j, \max}^4 = \left(\frac{K_{g0} Nu_j}{2d_j} \right) (\overset{\circ}{T_g^2} - T_{j, \max}^2)$$

TPD-Based Thermophoretic-Thermocouple Method

PARTICLE EMISSIVITY DETERMINATION

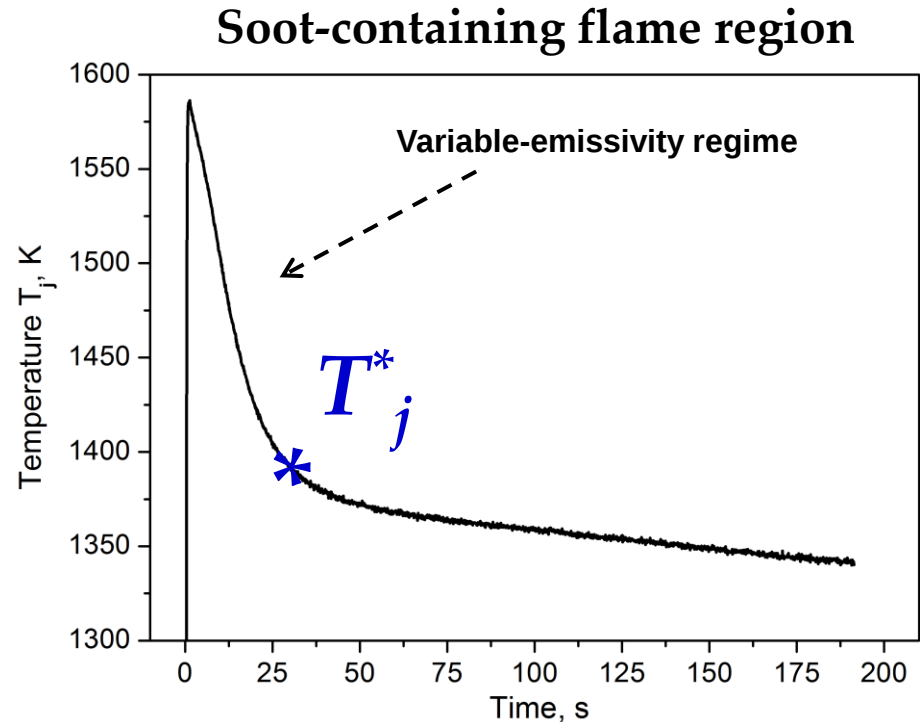


TPD-Based Thermophoretic-Thermocouple Method

PARTICLE EMISSIVITY DETERMINATION

Energy balance

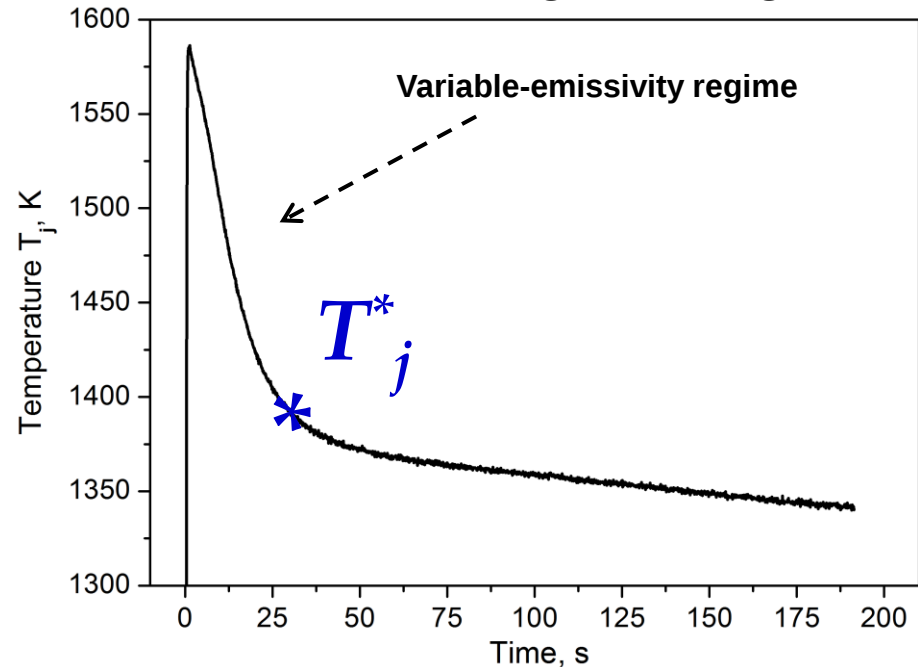
$$\epsilon_{\text{par}} \sigma T_j^{*4} = \left(\frac{K_{g0} \text{Nu}_j}{2d_j} \right) (T_g^2 - T_j^{*2})$$



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TPD-Based Thermophoretic-Thermocouple Method

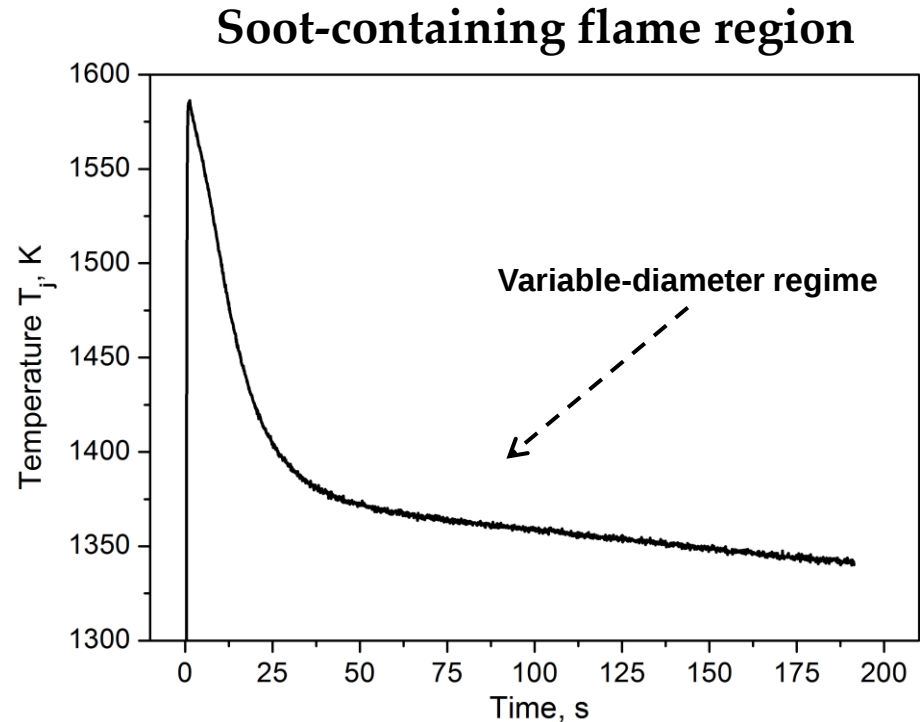
PARTICLE f_v DETERMINATION

Energy balance
+
Mass balance
(*Thermophoretic flux*)

$$G = \frac{1}{4} \left(\frac{T_g}{T_j} \right)^8 - \frac{1}{6} \left(\frac{T_g}{T_j} \right)^6 = G_0 + \frac{2D_T \epsilon_{\text{particle}}^2 \sigma^2 T_g^4}{(\rho_d / \rho_p) K_{g0}^2 Nu_j} f_v t$$

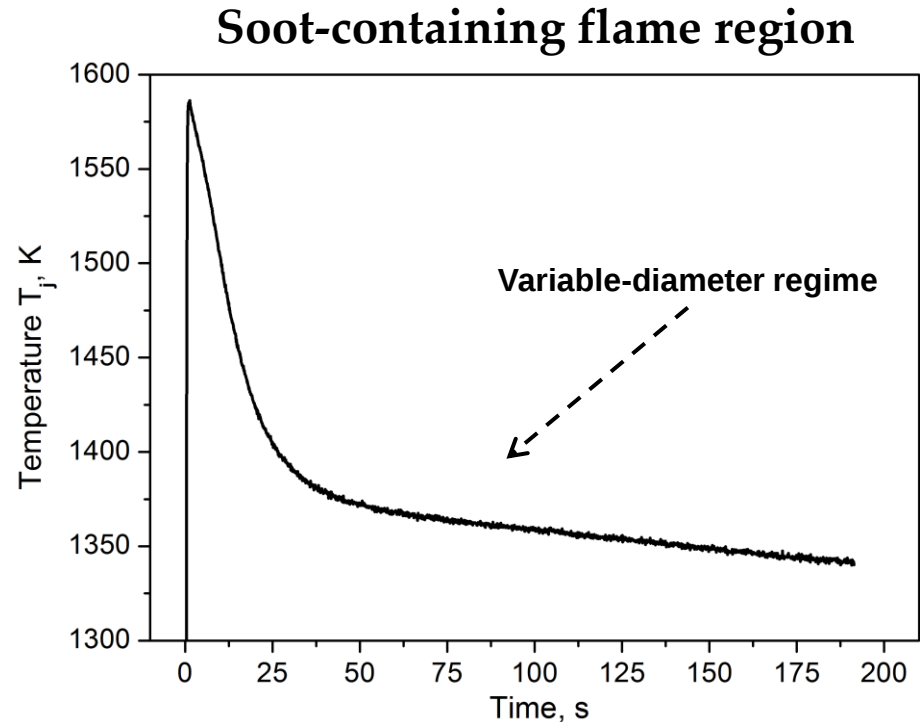
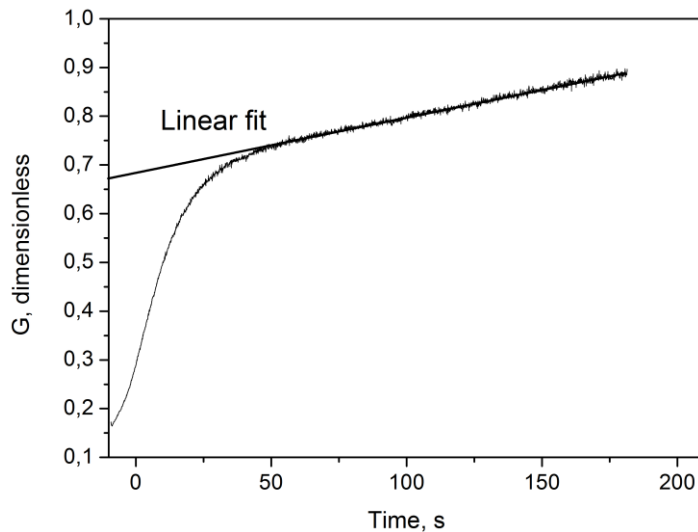
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Laminar Premixed Flame: Method Validation



- Ethylene/Air Laminar Premixed Flame
- McKenna Burner
- $v_{cg}=9.8 \text{ cm/s}$, $\Phi=2.01$

Type-R thermocouple, bead diameter = $235 \text{ }\mu\text{m}$

M. Commodo et al. *Combust. Flame* **162** (2015)

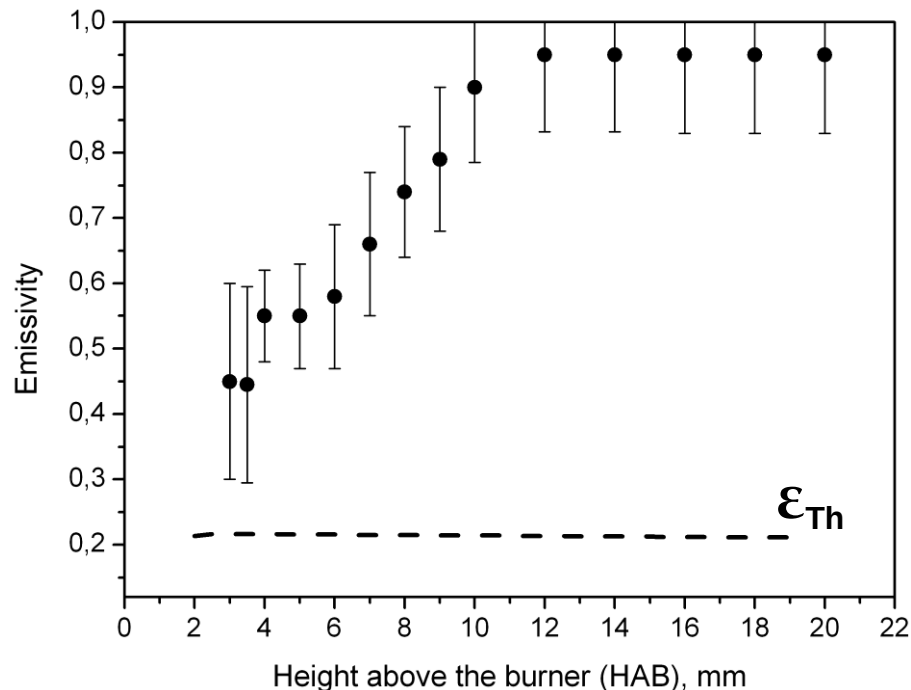
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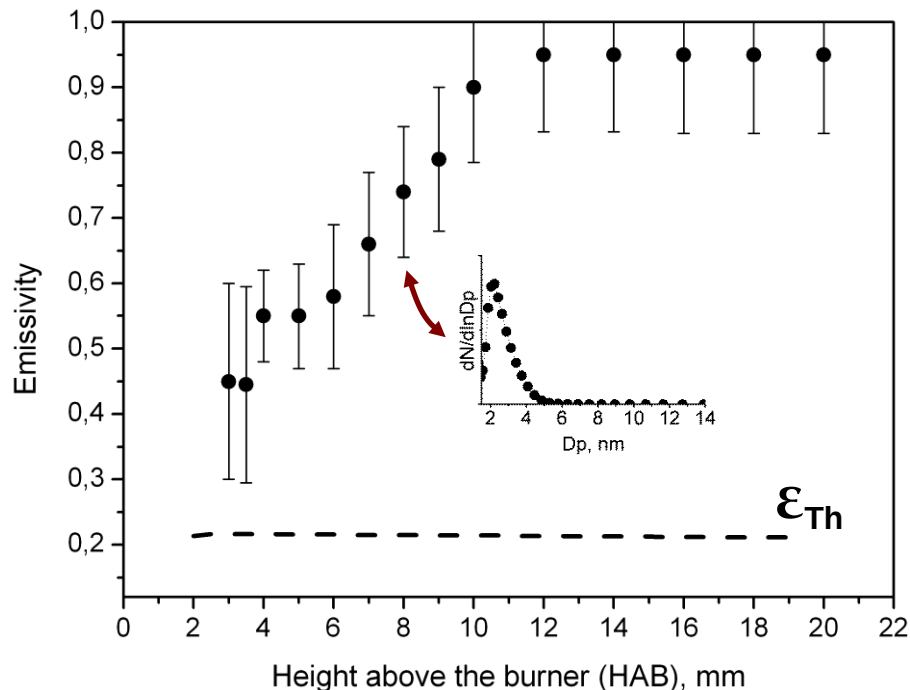
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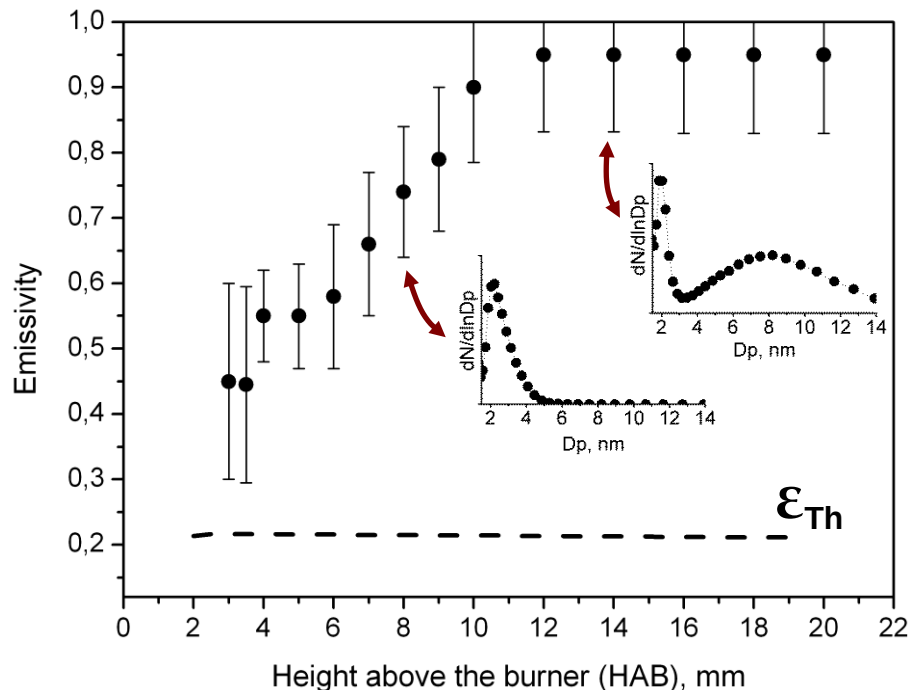
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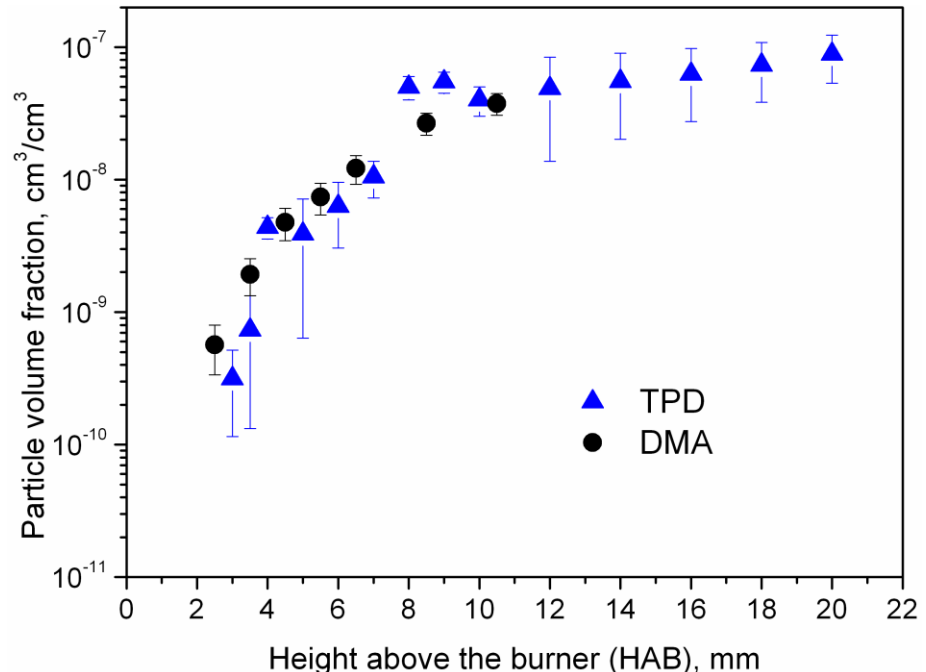
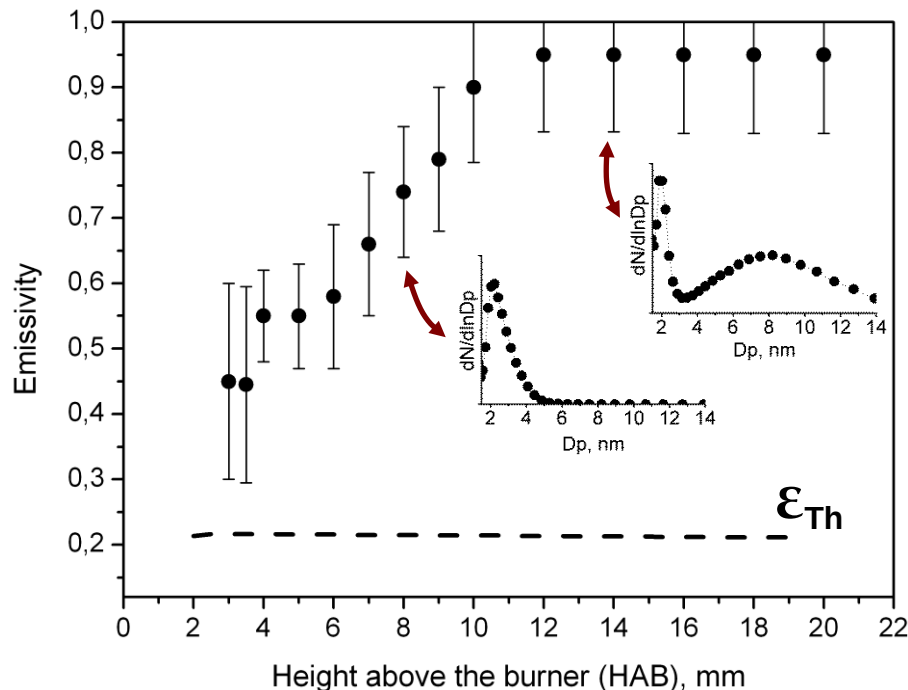
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M. Commodo et al. *Combust. Flame* **162** (2015)



Laminar Diffusion Flame

- Ethylene/Nitrogen Co-Flow Laminar Diffusion Flame
- ISF target flame (ISF-3 Co-flow2, Condition a)

C.S. McEnally and L.D. Pfefferle *Combust. Flame* **121** (2000)

Type-R thermocouple, bead diameter = 235 μm

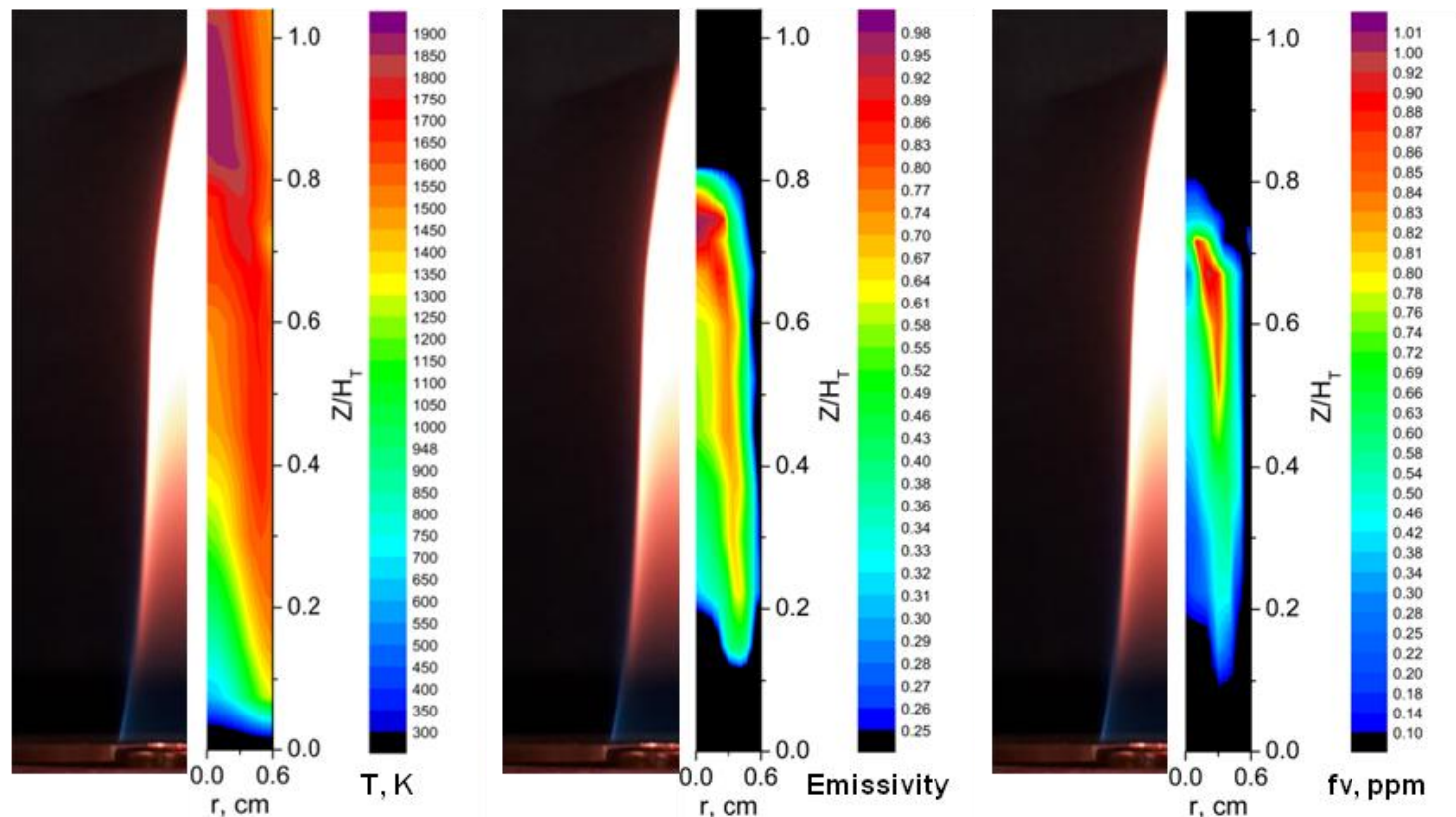


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Laminar Diffusion Flame

THE EFFECT OF OXIDATION

MASS BALANCE AT THE THERMOCOUPLE JUNCTION

$$\left(\frac{\rho_d}{2}\right) \frac{d(d_j)}{dt} = \left(\frac{D_T Nu_j f_v \rho_p}{2 d_j}\right) \left(1 - \left(\frac{T_j}{T_g}\right)^2\right)$$

(Thermophoretic flux)

Laminar Diffusion Flame

THE EFFECT OF OXIDATION

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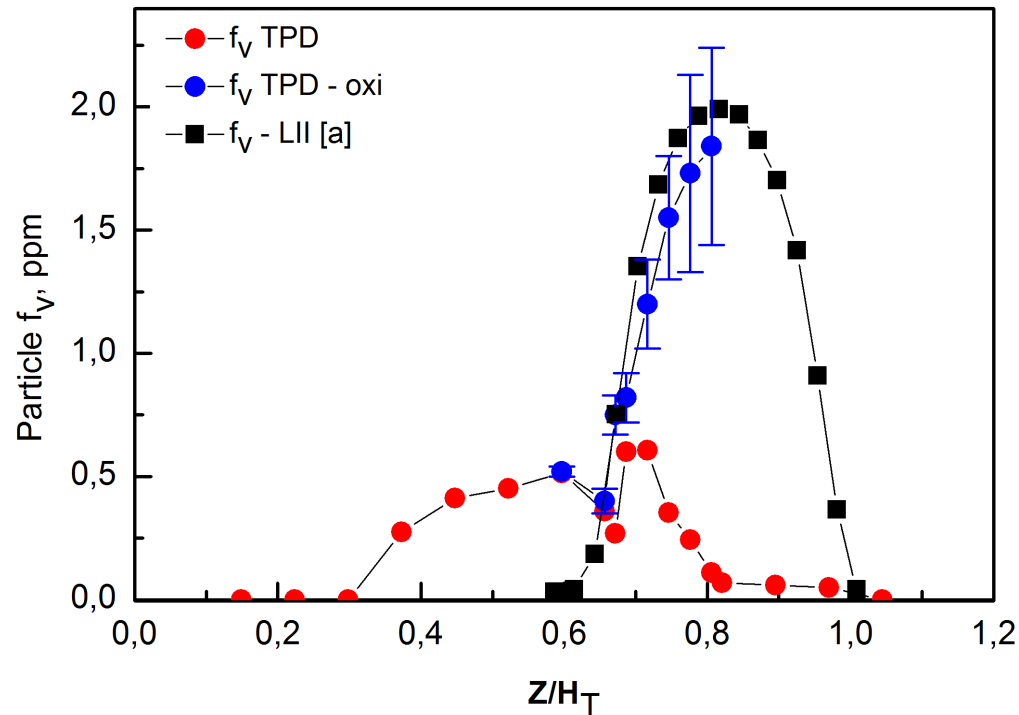
(Thermophoretic flux)

(Soot burnout rate by OH)

H. Ghiassi et al. *Energy Fuel* **30** (2016)

Laminar Diffusion Flame

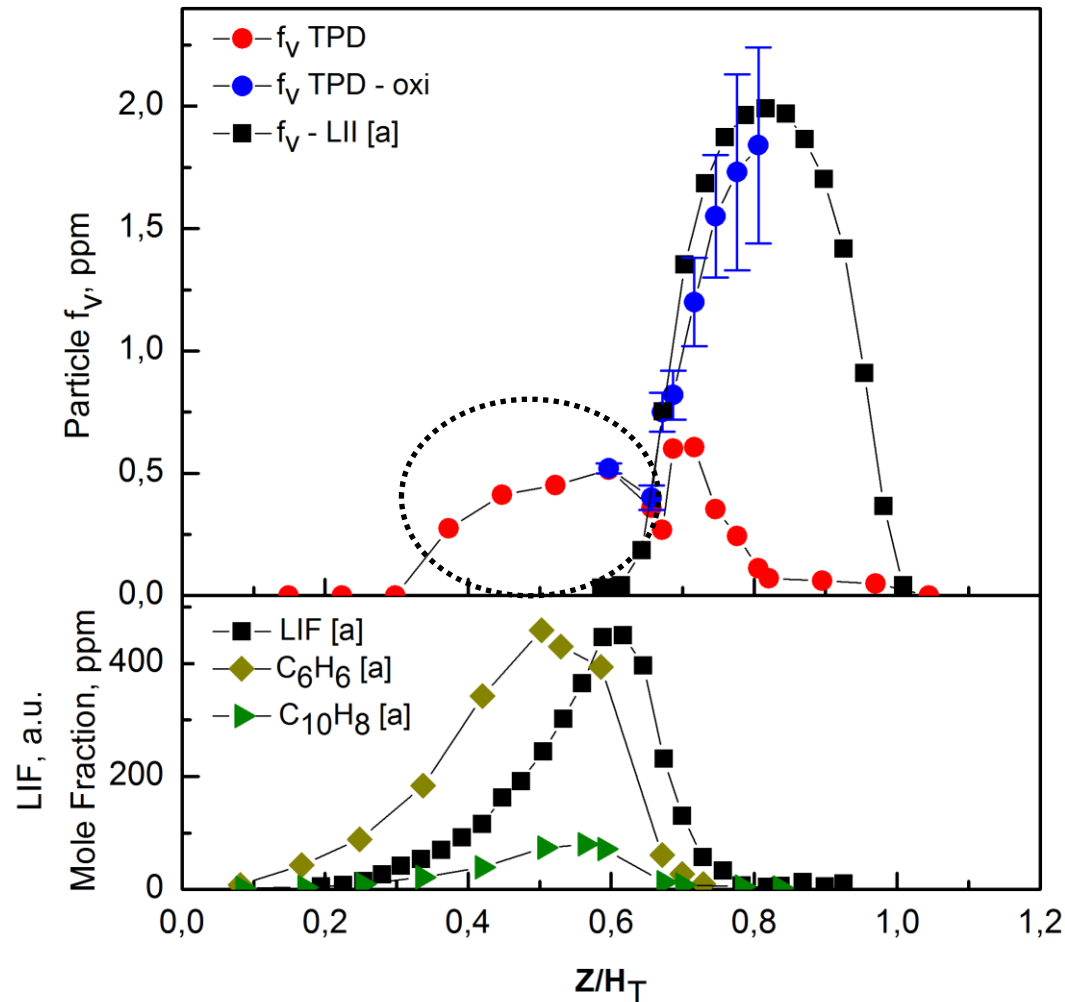
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Laminar Diffusion Flame

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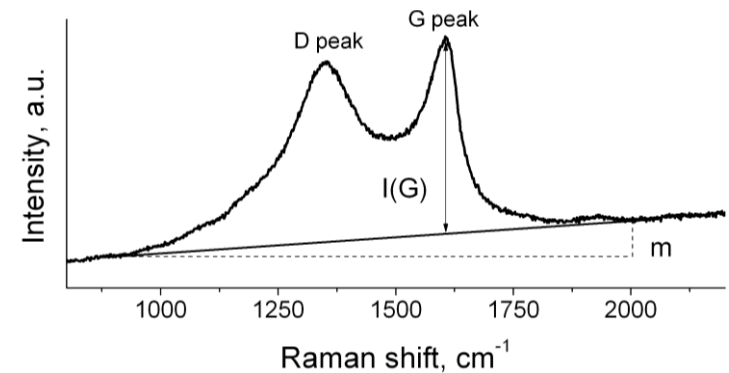


^a C.S. McEnally and L.D. Pfefferle *Combust. Flame* **121** (2000)

Laminar Diffusion Flame

THE EVOLUTION OF PARTICLE NATURE/COMPOSITION

Raman Spectroscopy

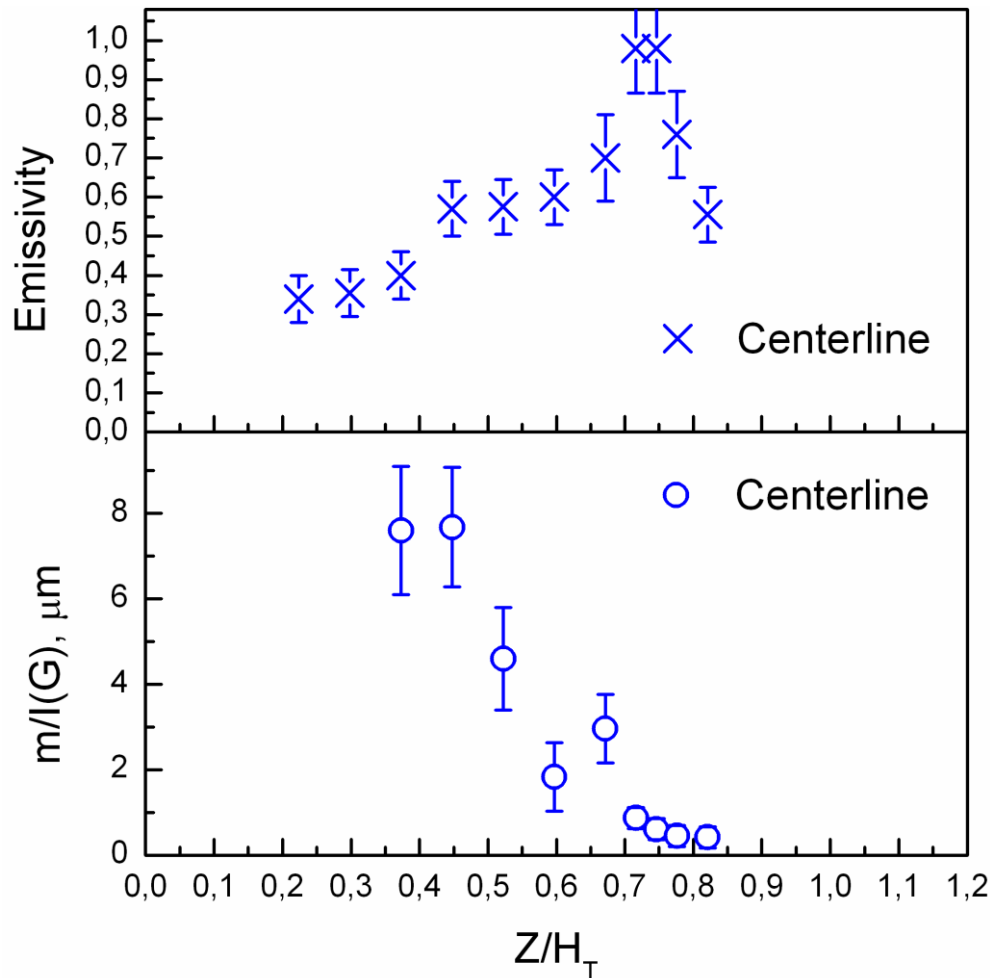


$$\frac{m}{I(G)} = \frac{\text{Fluorescence}}{sp^2 \text{ Carbon}}$$

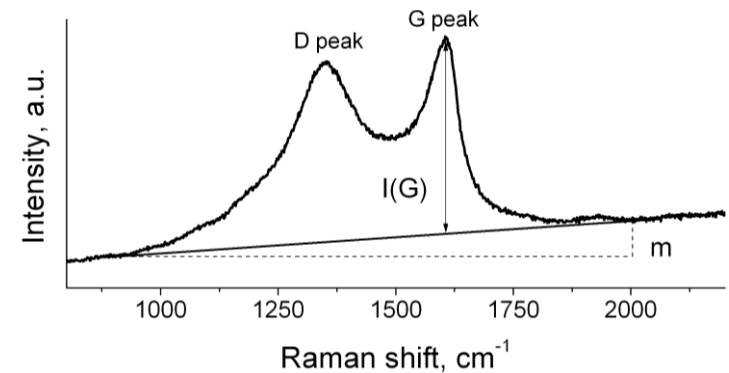
C. Casiraghi et al. *Diam. Relat. Mater.* **14** (2005)

Laminar Diffusion Flame

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C. Casiraghi et al. *Diam. Relat. Mater.* **14** (2005)

Final Remarks /1

- ❖ A Thermophoretic-Thermocouple method based on TPD has been developed for the investigation of particle evolution in flame;
- ❖ The method here reported has the capability to measure simultaneously particle volume fraction and emissivity, being at the same time very simple, fast and cheap to operate;
- ❖ The method has been first successfully validated on a laminar premixed flame, and then applied for studying a laminar diffusion flame.

Final Remarks /2

- ❖ Precursor particles formed in the center of the flame have an emissivity $\varepsilon \sim 0.4-0.6$ and can emit fluorescence;
- ❖ Emissivity of particles increases following a carbonation pathway up to $\varepsilon \sim 0.95$, their capability to emit fluorescence strongly decreases and TPD-fv coincides with that measured by LII;
- ❖ In a moderately oxidative flame region, the inclusion of OH oxidation in the mass balance equation for TPD allows evaluating properly particle volume fraction.

*THANK YOU
FOR YOUR
ATTENTION*