20th ETH Conference on Combustion Generated Nanoparticles June 13th – 16th, 2016 ETH Zurich, Switzerland Section 6B: Health Session

Dosimetry, Dose-metrics, Bioprocessing and Human Risk Extrapolation

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Focus Event: Particle Filter Quality under Real World Conditions



REACTIVITY OF ULTRAFINE PARTICLES

High specific surface area: m²/g

Surface Reactivity as Dose-Metric, *e.g.*, ROS inducing potential to determine response per unit particle surface area

DTT (dithiothreitol) assay DCFH-DA (2'-7' dichlorofluorescin-diacetate) assay FRAS (ferric reducing ability of serum) assay Vit C assay ESR others...

as screening tool for categorization of UFPs based on reactivity in cellular or cell free assays for Hazard Ranking [Bello et al., 2009; Rushton et al., 2010]

Noncellular Nanoparticle ROS Summary (Carbon Particles)

Particle Mass Correlation



Noncellular ROS Summary (Carbon Particles) Particle Surface Area Correlation





Rochester PM Center

CONCLUSIONS, UF PARTICLE BOUND ROS (oxidative potential)

- ROS activity/m³ of ambient UFP can vary widely
- ROS activity of lab-generated nanoparticles seems to reasonably well predict acute in vivo responses
- ROS activity of ambient UFP does not necessarily parallel their number or mass concentration
- Do results of epidemiological studies indicate association between UFP reactivity and effects at organ of entry or secondary organs?

Challenges re UFP Standard:

Physico-chemical properties of UFP are different from different sources:

Elemental carbon Organic carbon compounds Inorganics (metals) Agglomeration/aggregtion Surface properties Solubility Volatility

influence Toxicity

UFP source $A \neq$ UFP source B \implies different biol./toxicol. effects

Proposed Concept for Standard

- UFP standard should best be source-specific
- Based on number concentration of emitted UFP
- Need to identify sources that emit most reactive UFP
- Regulate these sources *(rather than all UFP)* by introducing a number emission standard, based on UFP risk assessment

What is surface reactivity of Coal/Biomass Combustion Generated Nanoparticles?

What is surface reactivity of Coal/Biomass Combustion Generated Nanoparticles? *Chem. Characterization by HR-TEM/STEM/EELS*







<u>Case study with inhaled Nano SiO₂NPs:</u>

Exposure-Dose-Response Inhalation Study in Rats to determine No Effect Level

Case study with inhaled Nano SiO₂NPs:

Exposure-Dose-Response Inhalation Study in Rats to determine No Effect Level

Dosimetric extrapolation modeling to assess human risk

Silica/SiO₂ Starting Materials



 Majority of the NPs are spherical or semi-spherical and ~ 20-40 nm in size.

• Nanoparticles tend to form dense agglomerated aggregates.

• Nanoparticles have smooth surfaces without etching or dissolution patterns.

 Particles are not zoned or show different densities (core to surface).

• Particles are amorphous

SiO₂ Starting Nanoparticles

This HR-TEM shows the amorphous nature of the supplied SiO₂ NPs.

Aggregation and Agglomeration is part of NPs Formation.





Means ± SD

Dosimetric Extrapolation of Inhaled Particles from Rats to Humans



Assumption: If retained dose is the same in rats and humans, then effects will be the same

Estimation of Chronic NOAEL from Subchronic Rodent Study using MPPD Model



From: Oberdörster, 2002

MPPD v3.04

File Input Data Calculations Report Results Plot Results Help Get Started



$\Theta \cap \Theta$	Airway Morphometry	
Species	✓ Human)
Model	kat Mouse Rhesus Pig Rabbit	C
FRC	3300.0	ml
URT Volume	50.0	ml
Default	Help	ОК

	Exposure Scenario	
Acceleration of Gravity	981.0	cm/s²
Body Orientation	Upright \$	
Body Orientation: α	Leaning Forward	•
Body Orientation: β	On Back On Stomach	0
Body Orientation: y	On Right Side On Left Side	•
Aerosol Concentration	Upside Down 1.0	mg/m ³
Breathing Frequency	12.0	per minut
Tidal Volume	625.0	ml
Inspiratory Fraction	0.5	
Pause Fraction	0.0	
Denething Commis	. A Nacal	
Default Help	Oral Oronasal-Mouth Breath Oronasal-Normal Augr Endotracheal	ner nenter

$\Theta \cap \Theta$	Particl	e Properties	
Density	1.0	g/cm³	
Aspect Ratio	1.0	=1 for spherical	
			Single
Diameter	1.0	μm	Multiple
			Multimodal
\odot	смд 🛛 🔿 м		AD
	🗌 Inhalabil	ity Adjustment	
GSD (diam.)	1.0]	
GSD (length)	1.0]	
Correlation	0.0]	
	🗌 Equiv. Diam	. Model	
Diff. Diameter	1.0	μm	
Sed. Diameter	1.0	μm	
Imp. Diameter	1.0	μm	
Int. Diameter	1.0	μm	
Help			ОК

MPPD Model

Input Choices

Clearance Settings	
Tracheal Mucous Velocity	5.5 in
Fast Human Clearance Rate	0.02 ys
Medium Human Clearance Rate	0.001 ys
Slow Human Clearance Rate	0.0001 ys
Lymph Node Human Clearance Rate	0.00002 ys
Rat Clearance Parameter 'a'	0.03341
Rat Clearance Parameter 'b'	1.7759
Rat Clearance Parameter 'c'	0.3123
Rat Clearance Parameter 'd'	0.00071642
Lymph Node Rat Clearance Rate	0.00105652
Exposure Time Settings:	
Number of Hours Per Day	6
Number of Days Per Week	5
Number of Weeks	1
Max. Post-Exposure Days	0
Default Help	ОК



Determining Aerosol Density for Input into MPPD Model

Several terms and meanings for density (p = mass/volume): specific; material; packing, effective, relative

Deposition of airborne particles in the respiratory tract is affected by effective or actual density of aerosols

A number of suggested methods to determine aerosol density: Charvet et al., 2014, 2015; Maricq et al., 2004, Spencer et al., 2007 Miller et al., 2013; Park et al., 2003; Hering and Stolzenburg, 1995; Wang et al., 2015

Result of MPPD derived ρ_{eff} for SiO₂ slurry aerosols using data of 4-hr. rat inhalation study:

 $\rho_{\rm eff} = 0.165 \, {\rm g/cm^3}$

Compare to SiO_2 material density of 2.65 g/cm³!

b_{tot} of biosoluble particles in the lung is the sum of mechanical (AM-mediated) removal and of dissolution:

 $\mathbf{b}_{\text{tot}} = \mathbf{b}_{\text{mech}} + \mathbf{b}_{\text{diss}}$

 $\mathbf{b}_{\mathbf{diss}} = \mathbf{b}_{\mathbf{tot}} - \mathbf{b}_{\mathbf{mech}}$

Note:

Species differences of particle clearance rates in rat and human lung: <u>mechanical clearance rate</u>: very different (T½ rat ~ 70 days, human ~400-700 days) <u>dissolution clearance rate</u>: assumed to be the same in mammalian lungs



Verifying in vivo dissolution of SiO₂ NPs by HR-TEM/STEM/EELS analysis



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After 27 days p.E., STEM shows many areas that are enriched in Si, but the nanoparticle size is so small that at the depicted Magnification, the particles appear like clouds (outlined with the yellow lines).

The Si-enriched zones have very small nanoparticles that could be Identified to have Si, but it is not Determined whether they are SiO_2 , or Si-phosphates (see next slide).

The formation of the Si-enriched areas is a clear indication, that after 27 days, the original SiO_2 NPs have undergone at least partial in vivo processing. We observe dissolution patterns (rough surfaces, pore formation in the starting materials, edge pits with areas of high solubility) In addition we see formation of precipitates that are << 2 nm and are part of what appears as Siclouds. The Si-nanoparticles inside clouds are well dispersed suggesting, t hat there is some insitu mechanism that prevents particle agglomeration. More work needed to identify coronas.

HEC Calculation from 4 week rat inhalation study with SiO₂ slurry aerosol:

Environmental Exposure

Deposition in **human lung** of inhaled SiO₂ aerosol of same particle size as in rat study, predicted by MPPD model with MMAD = 0.38 µm, GSD = 2.0, ρ = 0.165: **6.3** % deposition in alveolar region, **3.75** % in tracheo-bronchial region environmental setting: TV 625 ml; BF 12 min⁻¹ (resting breathing)

Alveolar Surface Area Basis

<u>NOAEL rat: 13.6 μg</u> (retained lung dose at end of 4 week exposure in rats) Normalized by alveolar surface area: **5.97 ng/cm²** surface area Equivalent total retained dose in humans' lungs to be reached after continuous exposure: <u>3,787 μg</u>

> **HEC** to reach this lung burden over total life at 24 hours/day, 7 days/ week, resting breathing, 365 days/year: <u> $304 \ \mu g/m^3$ </u>

Daily deposited dose in humans: 208 µg/day (additional safety/assessment factors to define Reference value?)

Risk Assessment and Risk Management Paradigm For Engineered Nanoparticles (NPs)



Modified from Oberdörster et al., 2005

Risk = f (hazard; exposure)

