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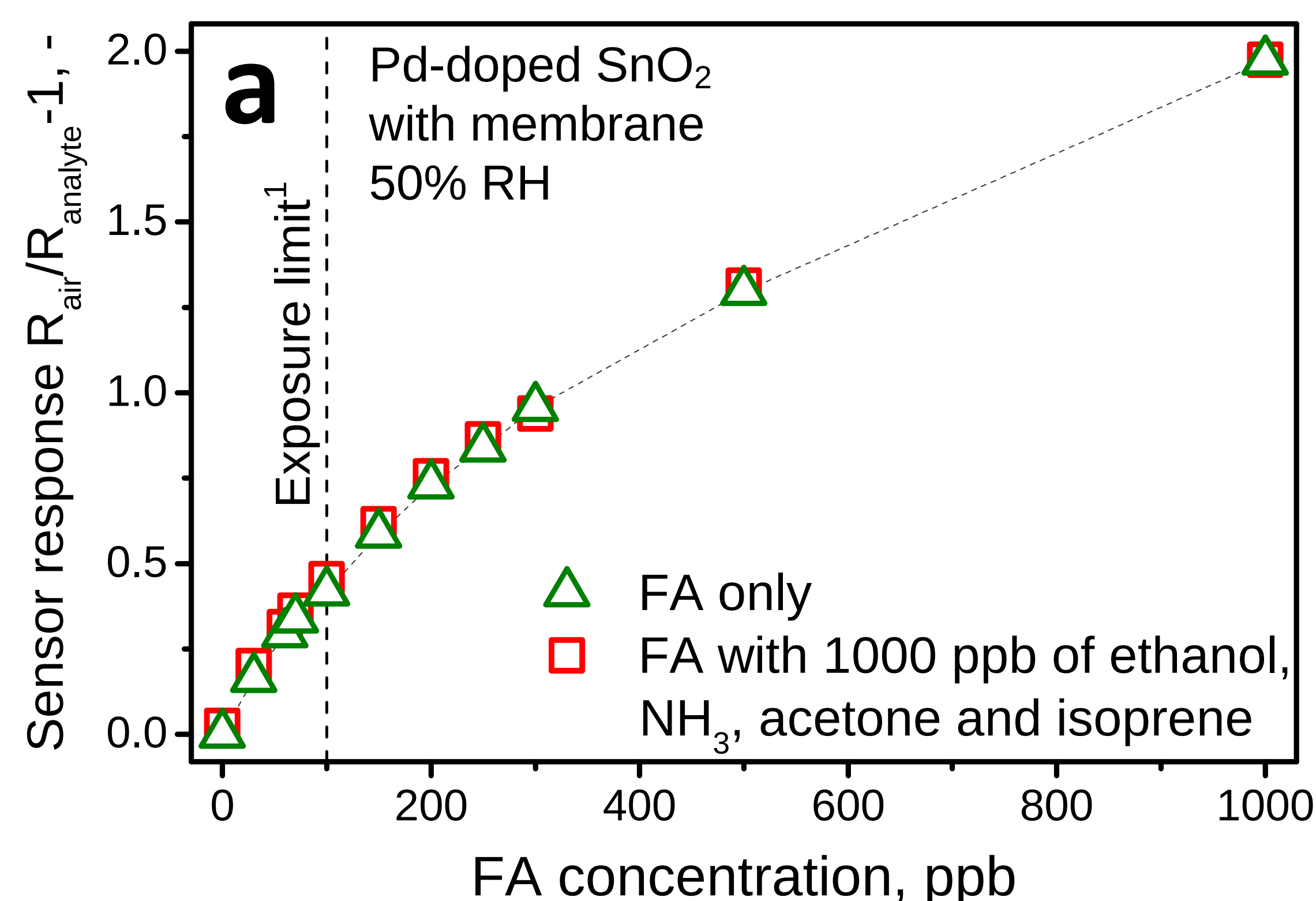
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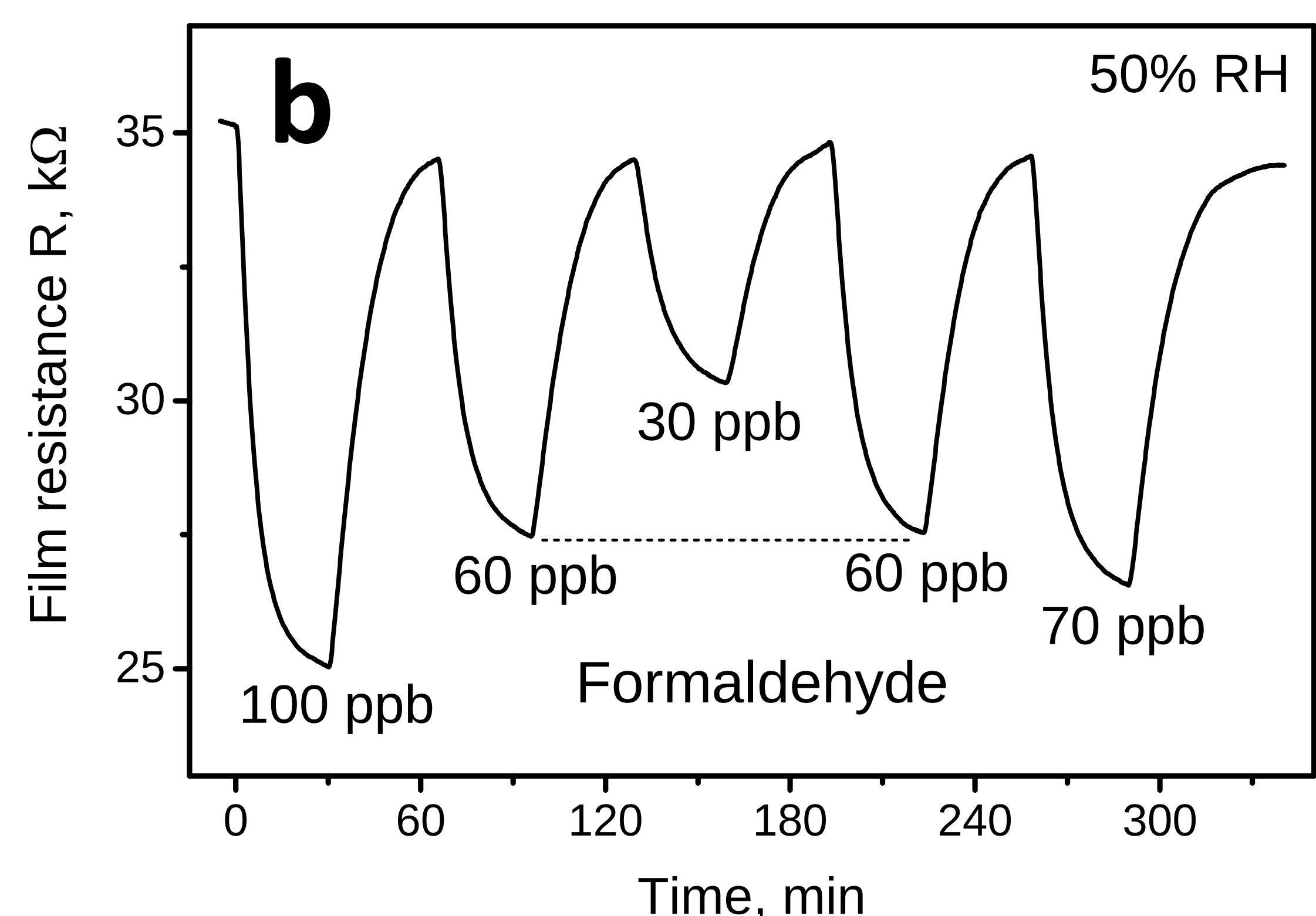
Air quality monitoring

Monitoring gaseous compounds in aerosols is of high interest. Formaldehyde (FA), for instance, is a *carcinogenic* indoor air pollutant released from wood-based furniture, laser-printers or paints,¹ and one of the proposed lung cancer markers in exhaled breath.² The recommended indoor air exposure limit should not exceed 100 ppb,¹ but suitable devices to monitor FA emission are missing. Chemo-resistive metal-oxide sensors made by flame spray pyrolysis (FSP) are quite attractive as they can detect sufficiently low FA levels, offer fast response and recovery times and can be produced cost-effectively.³ Also, they feature small size and low power needed for integration into autarkic FA monitors. However, they lack selectivity. Here, a modular sensor system is developed that overcomes this limitation by placing a highly selective membrane ahead of the sensor, enabling ultra-low FA detection in simulated gas mixtures.⁴

Outstanding performance in gas mixture

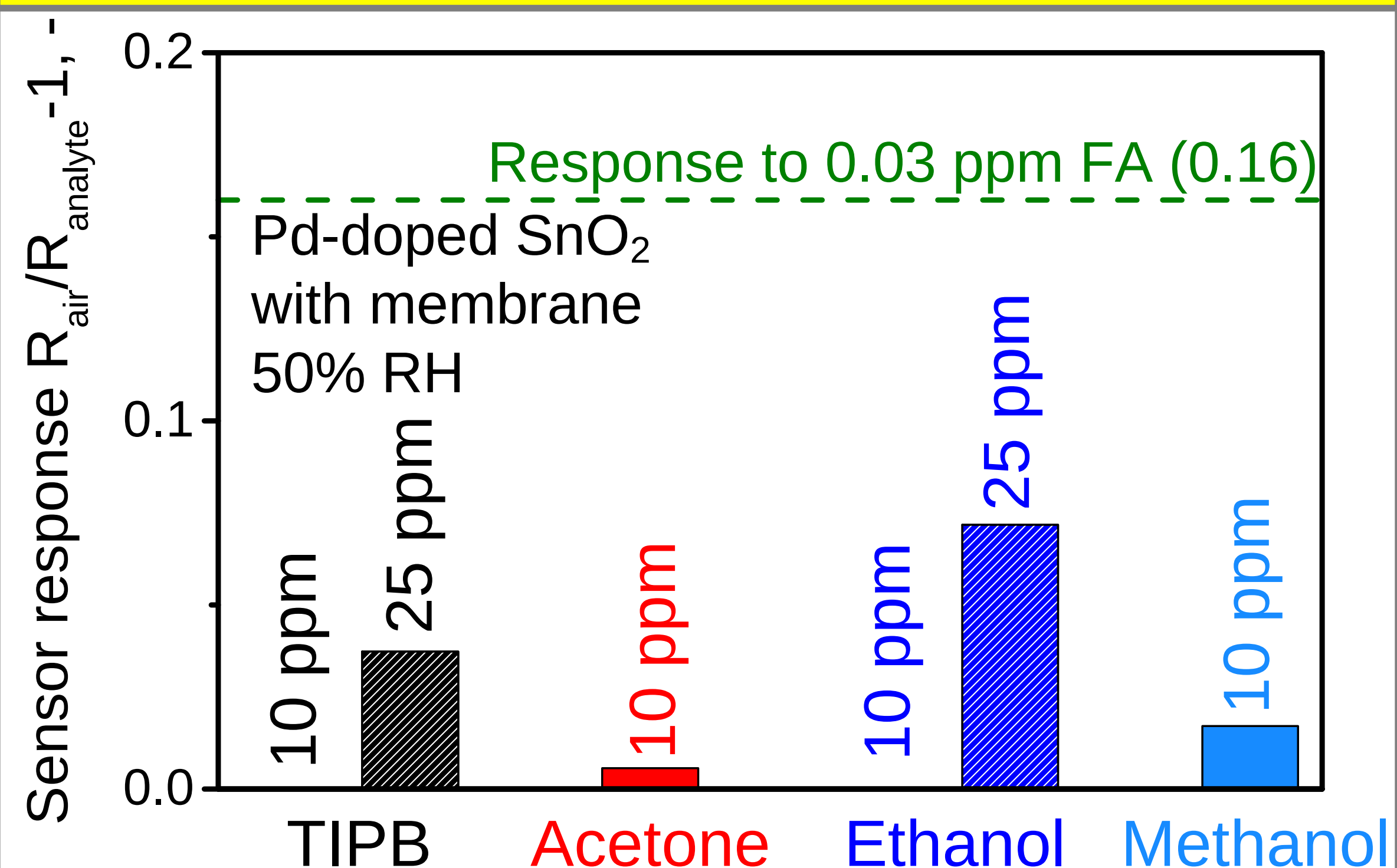


FA calibration curve at 50% RH. The membrane/sensor assembly can accurately detect FA down to 30 ppb even in a gas mixture containing NH₃, acetone, isoprene and ethanol each at 1000 ppb.



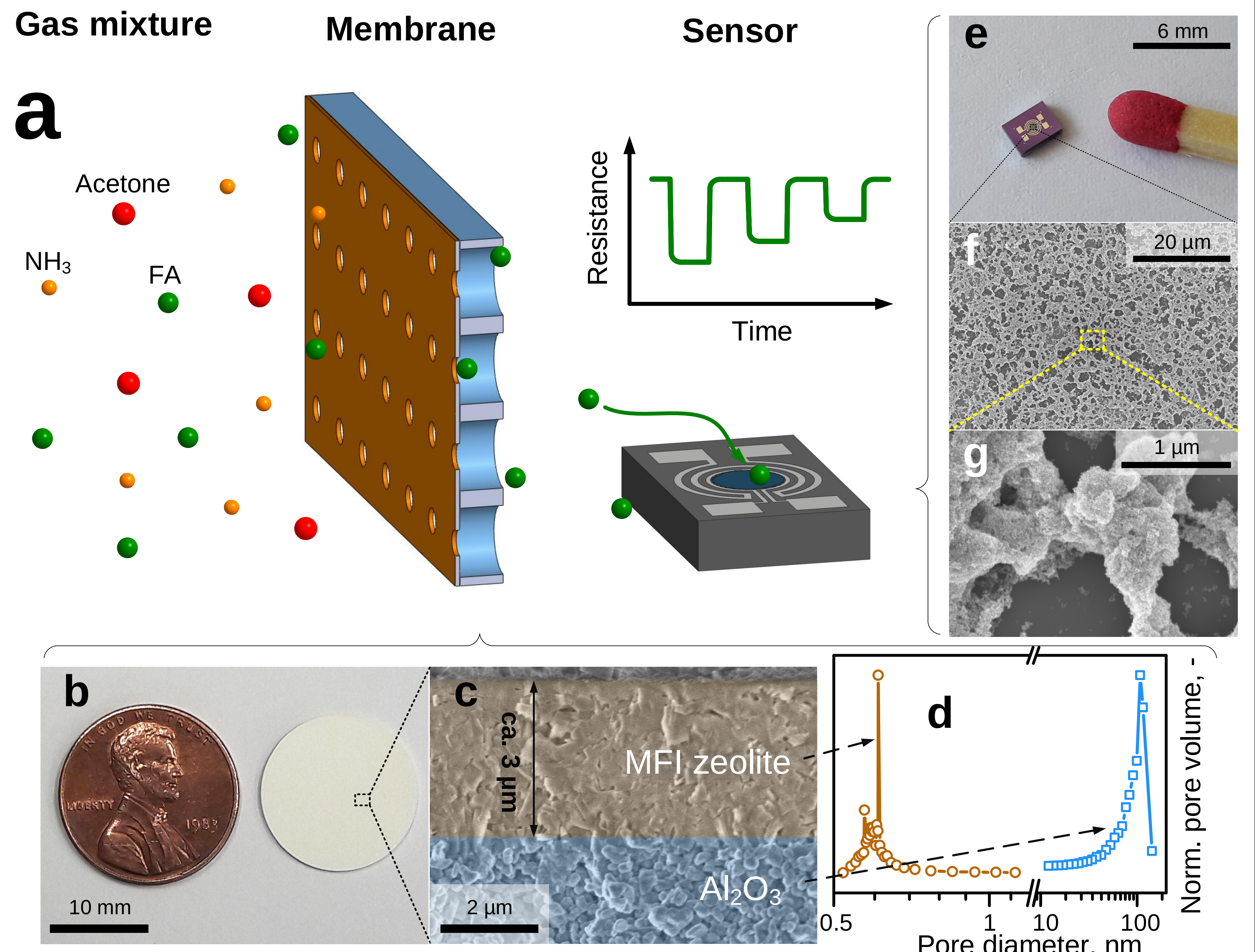
Sensor film resistance upon exposure to 100, 70, 60 and 30 ppb of FA at 50% RH. These ultra-low analyte concentrations below the recommended exposure limit¹ are detected reproducibly with high signal-to-noise ratio (> 70).

Interference at higher concentrations



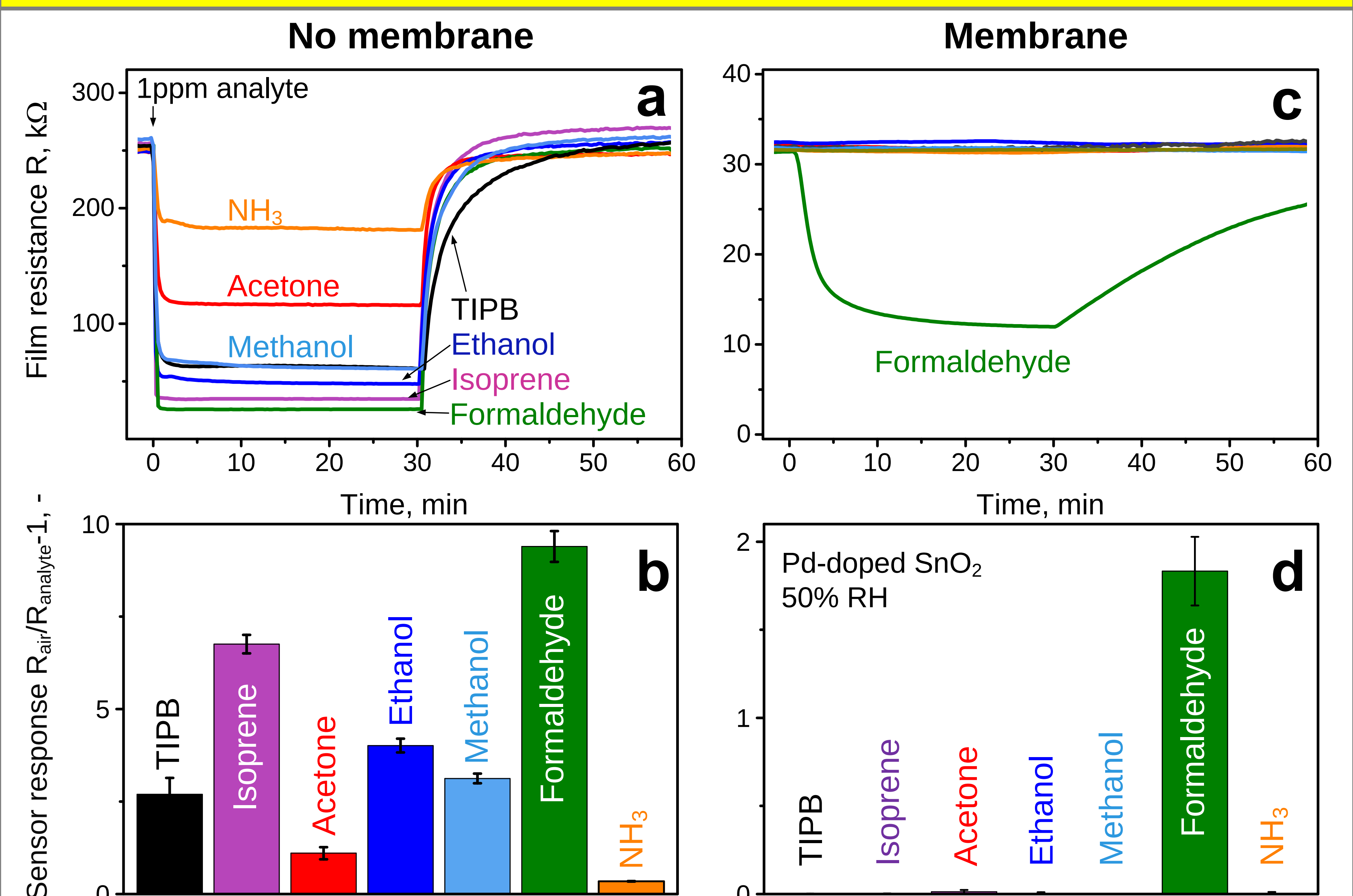
Comparison of responses to interfering gases at a concentration of 10 and 25 ppm and the response to 30 ppb FA. This outstanding selectivity enables FA detection even at much higher interfering concentrations.

Membrane-sensor concept for highly selective FA detection



A microporous MFI zeolite membrane pre-separate the gas mixture by molecular sieving and chemical separation, ideally allowing only FA to permeate. This enables highly selective FA detection with a chemoresistive sensor consisting of flame-made Pd-doped SnO₂ (1 mol%) nanoparticles that aggregate to a fine and extremely porous network.

FA selectivity enhancement by microporous membrane



Real-time resistance and evaluated sensor responses without (a,b) and with the microporous membrane (c,d) tested at 50% RH. Without membrane, the sensor reacts to all tested analytes and cannot selectively detect the target analyte. With the membrane, only FA is detected, enabling selective FA monitoring by this membrane/sensor assembly.

Conclusions

- Unprecedented FA selectivity enabled by combination of MFI/Al₂O₃ membrane with a Pd:SnO₂ sensor
- Sufficiently low FA levels are selectively detected down to 30 ppb in simulated mixtures enabling FA level monitoring
- Easy integration into portable detectors due to small and modular design
- Paradigm change in gas sensor development?

References

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