

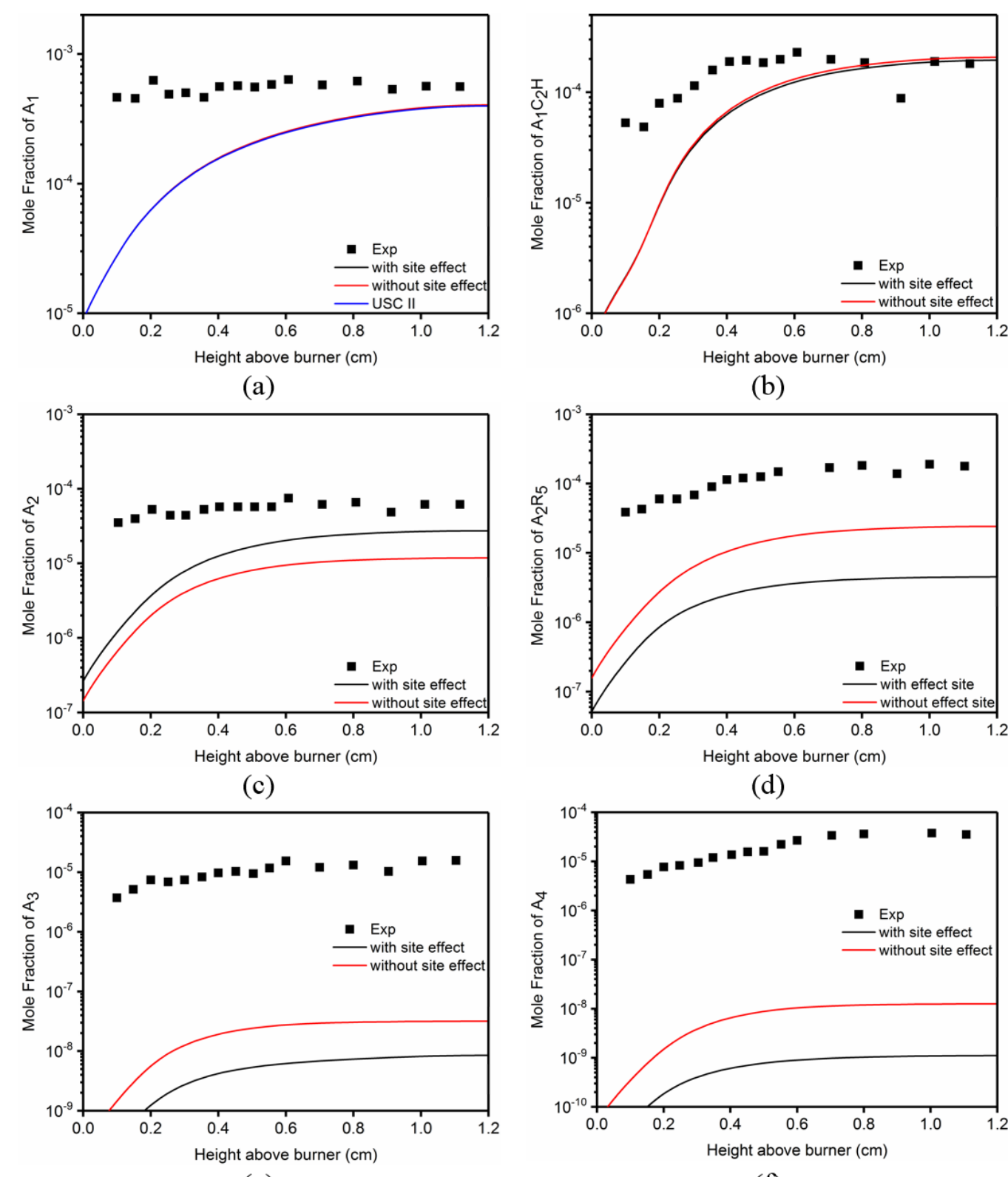
# A THEORETICAL STUDY OF PAHS GROWTH BY PHENYLACETYLENE ADDITION

Zepeng Li<sup>a,b</sup>, Peng Liu<sup>a</sup>, Peng Zhang<sup>b</sup>, Hong He<sup>b</sup>, Suk Ho Chung<sup>a</sup>, William L. Roberts<sup>a</sup>

<sup>a</sup> King Abdullah University of Science and Technology (KAUST), Clean Combustion Research Center, Saudi Arabia  
<sup>b</sup> Research Center for Eco-Environmental Sciences, Chinese Academy of Sciences, China

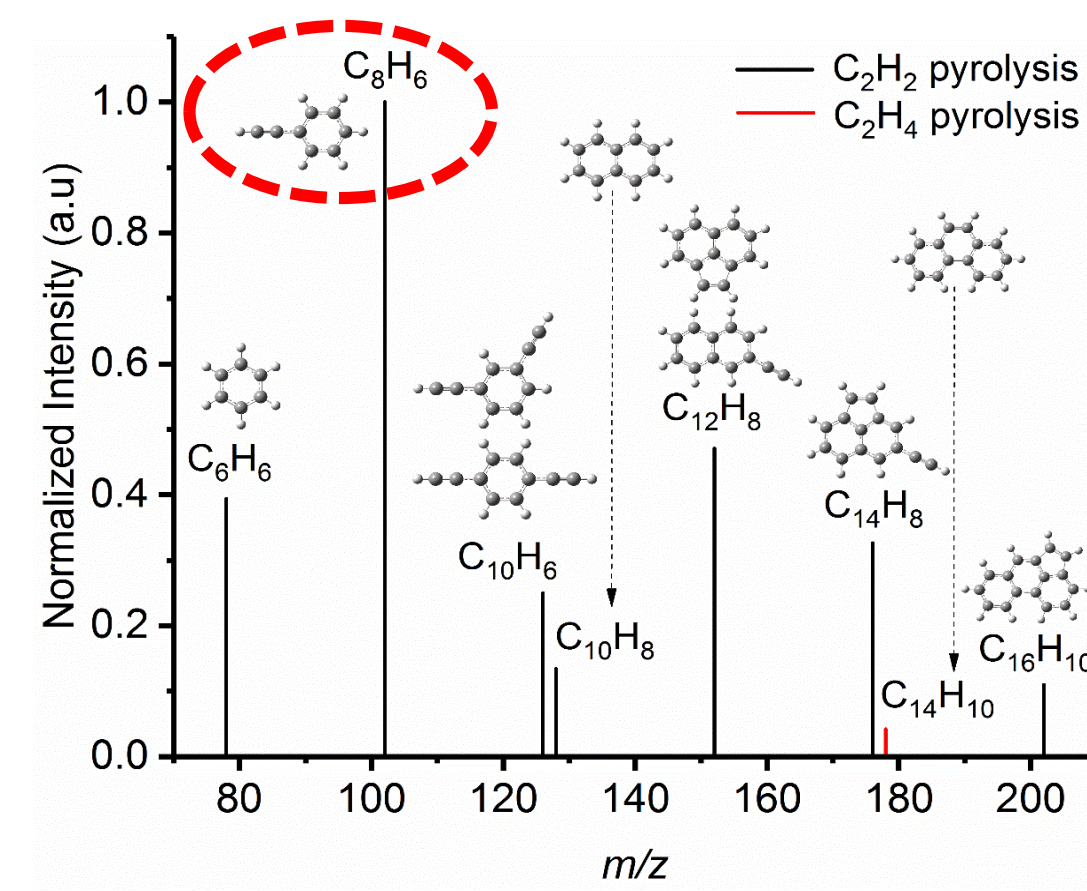
## Background

### Phenylacetylene ( $C_6H_5C_2H$ , $C_8H_6$ , or $A1C2H$ )

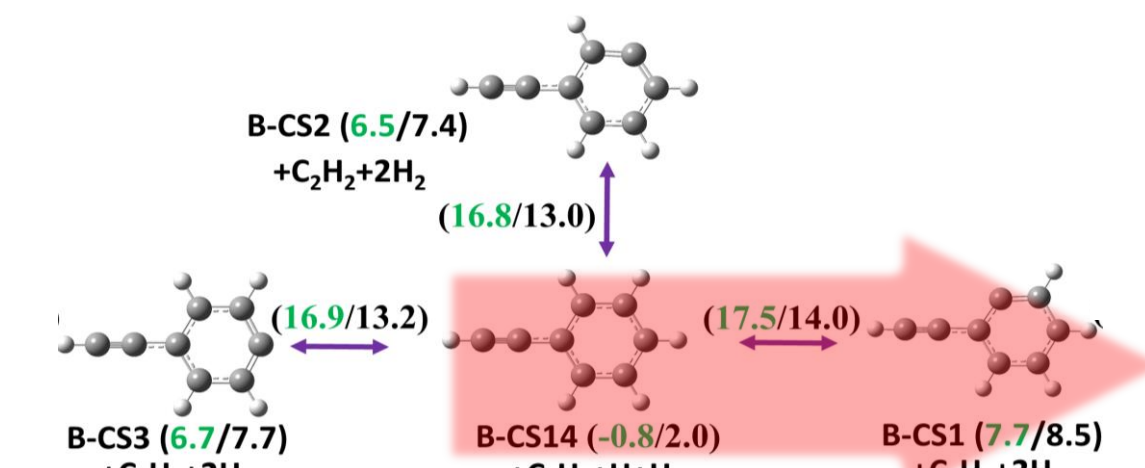


**Fig. 1** Comparison between experimental and calculated mole fraction of benzene and various PAHs in  $C_2H_4$  premixed flame [1]. (1 atm,  $C_2H_4/O_2/N_2 = 21.3/20.9/57.8$ ).

**Fig. 2** The mass spectra of acetylene and ethylene pyrolysis products observed at 1485 K [2]



**Fig. 3** Potential energy surface with energies calculated at CBS-QB3 (labeled as green) and 6-311+g(d,p) (labeled as black), respectively



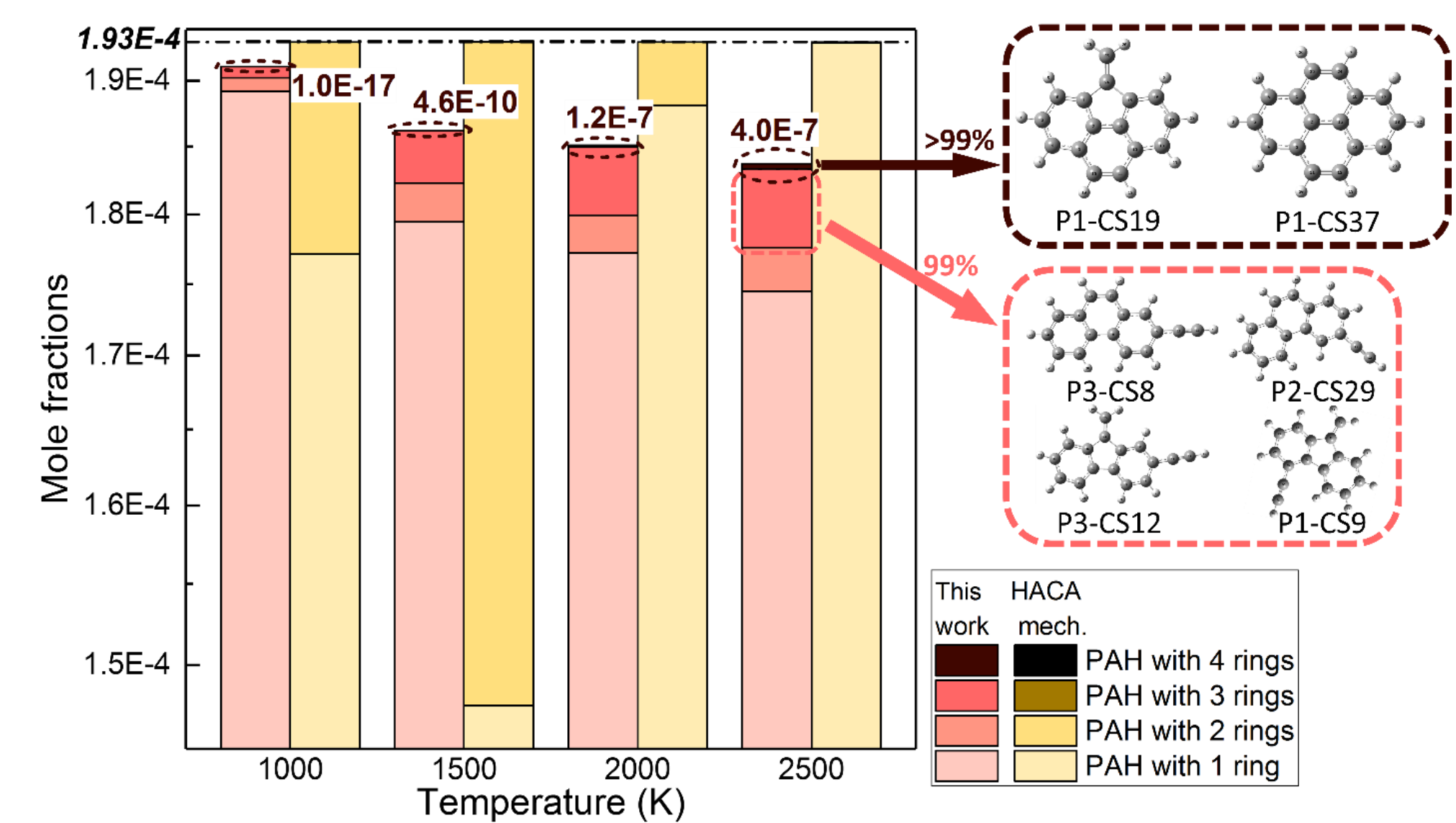
## Calculation details

- Potential energy surface of the reaction system
  - Optimize the structures and calculate energies at DFT B3LYP 6-311+G(d,p) level
  - Refine energies at CBS-QB3 level
- Kinetic analysis
  - Calculate reaction rate coefficients by TST theory
  - Check pressure dependence by RRKM theory
- Simulated using 0-D batch reaction in Chemkin

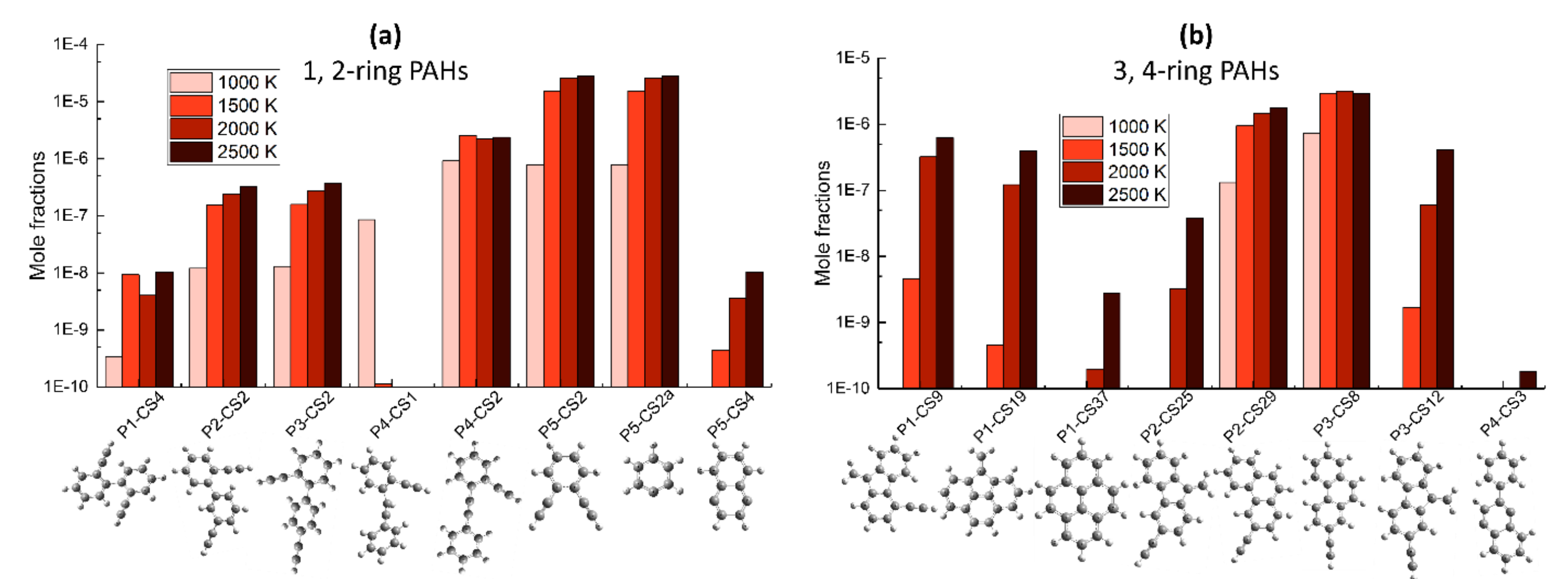
## Yield distributions in 0-D batch reactor

In the whole  $C_6H_5C_2H-C_2H_2-H-Ar$  reaction system, the initial reactants are set the same as the experimental premixed ethylene flames in Castaldi's work [3]: 193 ppm, 100 ppm, 0.045 and 0.954707 for  $C_6H_5C_2H$ , H, and  $C_2H_2$  and Ar, respectively.

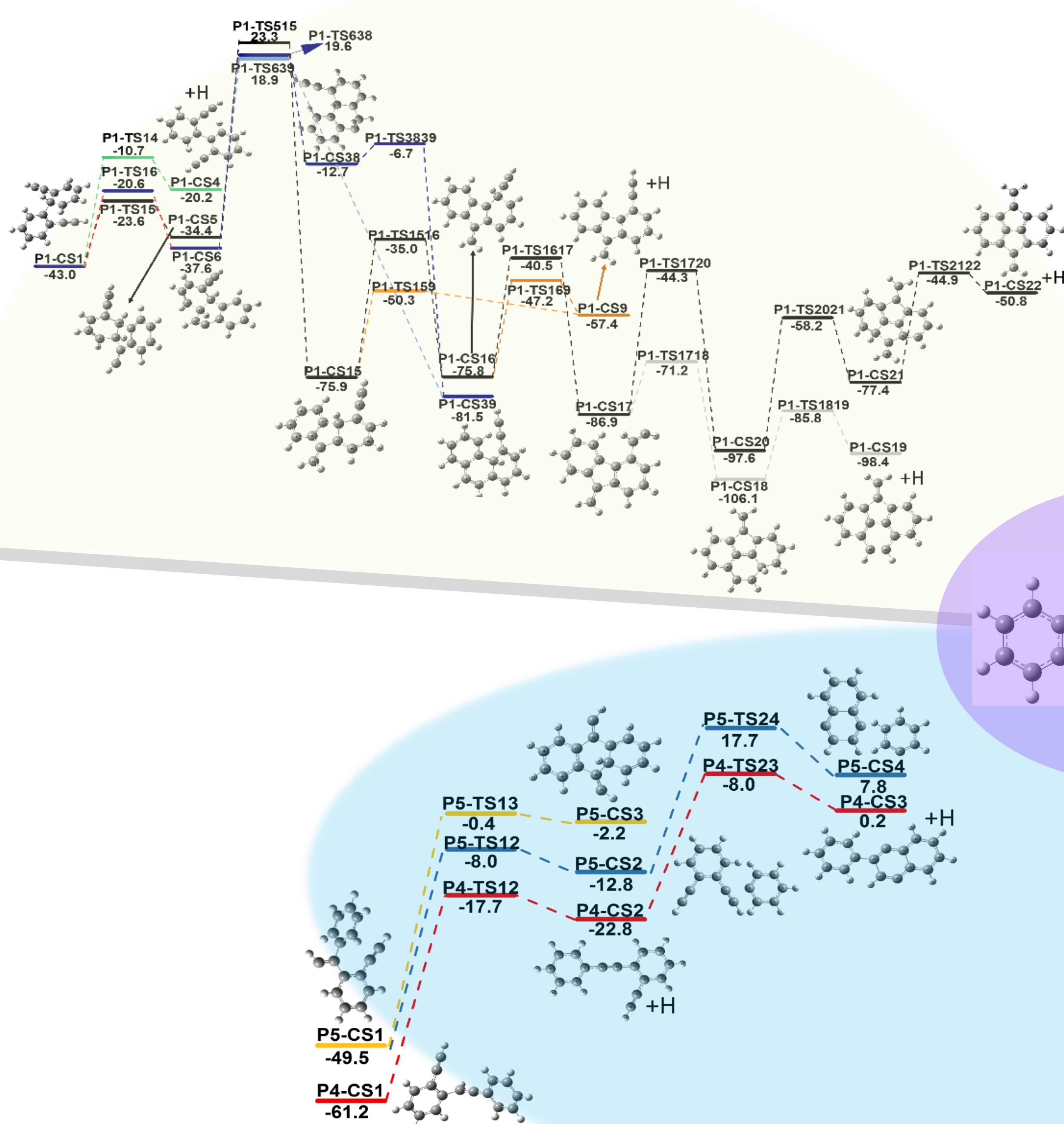
**Fig. 4** Comparison of HACA mechanism and this reaction network in the production of PAH with different sizes at elevated temperatures.



**Fig. 5** Simulated mole fractions of the major products in the  $C_6H_5C_2H-C_2H_2-H-Ar$  system in a 0-D batch reactor via the  $C_6H_5C_2H$  reaction network.



## Potential energy surfaces of the reaction system



Explore the formation of 3-, 4-ring PAHs by the addition reaction of  $C_6H_5C_2H$  and  $C_6H_4C_2H$  without the consideration of other carbon sources

- Energies are calculated at CBS-QB3 level.
- Only major channels are shown in the above potential energy surfaces.

## References

[1] M.J. Castaldi, N.M. Marinov, C.F. Melius, J. Huang, S.M. Senkan, W.J. Pit, C.K. Westbrook, Experimental and modeling investigation of aromatic and polycyclic aromatic hydrocarbon formation in a premixed ethylene flame, *P. Combust. Inst.* 26 (1996) 693-702.  
 [2] Liu, Peng, et al. "The site effect on PAHs formation in HACA-based mass growth process." *Combustion and Flame* 199 (2019): 54-68.  
 [3] M.J. Castaldi, Marinov, N. M., Melius, C. F., Huang, J., Senkan, S. M., Pit, W. J., Westbrook, C. K., Experimental and modeling investigation of aromatic and polycyclic aromatic hydrocarbon formation in a premixed ethylene flame, *Symposium (International) on Combustion* 26 (1996) 673-702