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Secondary nanoparticles formation and composition from open and residential wood combustion

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Background Methods SOA formation SOA composition Summary

Biomass burning organic aerosols



Background Methods SOA formation SOA composition Summary

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Forest fire and residential wood combustion

- Forest fire (wildfire) has increased in frequency, intensity, and area in the past few decades, and is predicted to continue to do so.
- Residential wood combustion is one of the major sources of organic aerosol in Europe during winter (15-60%).



McClure & Jaffe, PNAS, 2018, 7901.

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Methods – Overview

Open burning (Forest fire) - twigs, barks and needles



Residential wood burning - wood logs

Objectives:

- Quantify the SOA formation at a wide range of photochemical ages.
- Molecular-level SOA composition and its evolution over oxidation.

Background \rightarrow Methods \rightarrow SOA formation \rightarrow SOA composition \rightarrow Summary

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Oxidation Flow Reactor (OFR)



Laminar-flow OFR (formerly ECCC-OFR):

- Improved design can reduce jetting and recirculation in the reactor.
- Reduced wall loss and enhanced SOA yields.

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• Extractive electrospray ionization mass spectrometry (**EESI-MS**) is the cutting-edge technique to characterize molecular-level composition of aerosols.



Lopez-Hilfiker et al., Atmos Meas Tech, 2019, 4867.

Techniq ue	Time resolution	Molecular resolution	Fragmentatio n
ESI-MS*	Hours (off- line)	High	Minimal
AMS*	Seconds to minutes	Low	High due to El
TD- CIMS*	Minutes to hours	High	Medium (high temperature)
EESI-MS	Seconds	High	Minimal

Red: Bad; Yellow: Neutral; Green: Good

*ESI-MS: Electrospray ionization MS; AMS: Aerosol MS; TD-CIMS: Thermodesorption chemical ionization MS.

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SOA formation under different NOx conditions



- SOA mass first increases and then decreases with increasing photochemical age.
- SOA production varies between experiments.

SOA formation normalized by VOCs



- Open and stove burning are very similar after normalization.
- High-NOx condition slightly increases the SOA formation.

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• Using measured VOCs to calculate SOA



Bruns et al., Sci Rep, 2016, 27881.

Measured VOCs can only explain 10-20% of the formed SOA in the OFR, very different from previous smog chamber study.

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Using measured VOCs to calculate SOA



Possible reasons:

- OFRs and smog chambers have different photochemical ages.
- > The contribution of unmeasured low-volatility compounds (e.g., IVOCs).

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SMPS vs AMS vs EESI



- AMS and SMPS have very similar trend in SOA concentration.
- EESI is slightly different, likely due to different sensitivities of different category of products.

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Carbon and Oxygen distribution



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Average C#, O#, and O/C



The lower C# and higher O/C under high-NOx condition is observed for all photochemical ages.

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FE Average O/C



- No significant difference between open and stove burning.
- Higher O/C under high-NOx condition.

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Clustering analysis



- 12 clusters based on k-means clustering analysis.
- They evolve differently with increasing photochemical age.

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Clustering analysis – O# vs C#



Later-generation products have lower C# and higher O#.

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Clustering analysis – H/C vs O/C



Later-generation products have lower H/C and higher O/C.

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Conclusion and outlook

- Measured VOCs cannot explain the observed SOA production in OFR. Likely due to the contribution of IVOCs.
- High-NOx condition enhances O/C of SOA by reducing carbon number (molecular size).
- Using EESI-MS, we can track the molecular information of different generation of products in the oxidation of biomass burning emissions.

Outlook:

- Using other techniques (e.g. VOCUS PTR-MS) to quantify IVOCs and their contributions to SOA production.
- Performing PMF analysis with EESI and AMS data, and compare with the clustering analysis results.



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Supplementary Slides

LF-OFR



- CFD simulation indicates that the velocity in LF-OFR is generally uniform.
- The residence time distribution (RTD) is similar to that of ideal laminar flow.
- The H₂SO₄ yield from SO₂ oxidation is ~100%, significantly higher than other OFRs.



