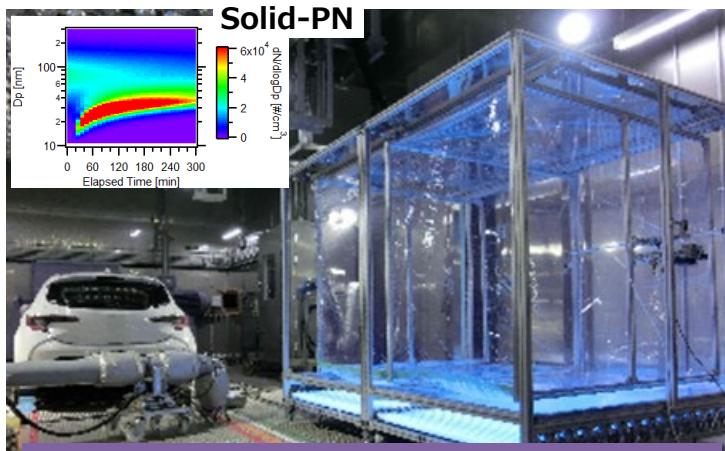


# Environmental Temperature Effects on Formation of Secondary Organic and Inorganic Aerosols formed from Vehicle Exhausts

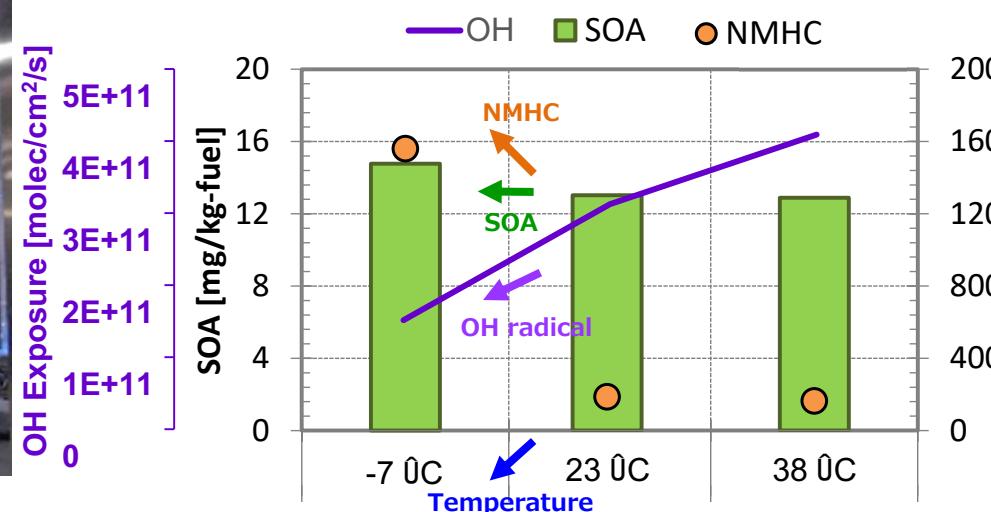
**Hiroyuki HAGINO, Risa UCHIDA**  
Japan Automobile Research Institute

## Environmental Chassis Dyno & Photochemical Smog Chamber

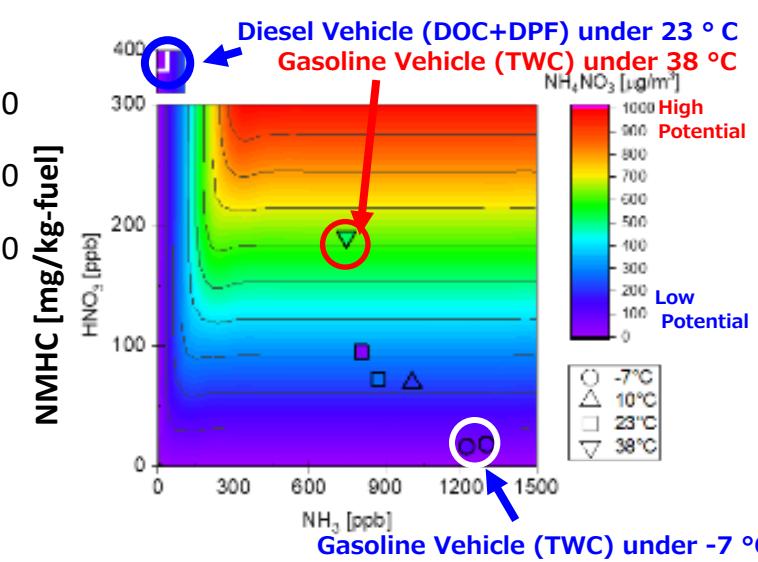


- Aims of This Research**
- Measurement of Formation Potential
  - Sensitivity Analysis of Precursors to Environmental Impact Substances

## Secondary Organic Aerosols (SOA) [VOC, IVOC, SVOC] + OH<sup>·</sup> → SOA



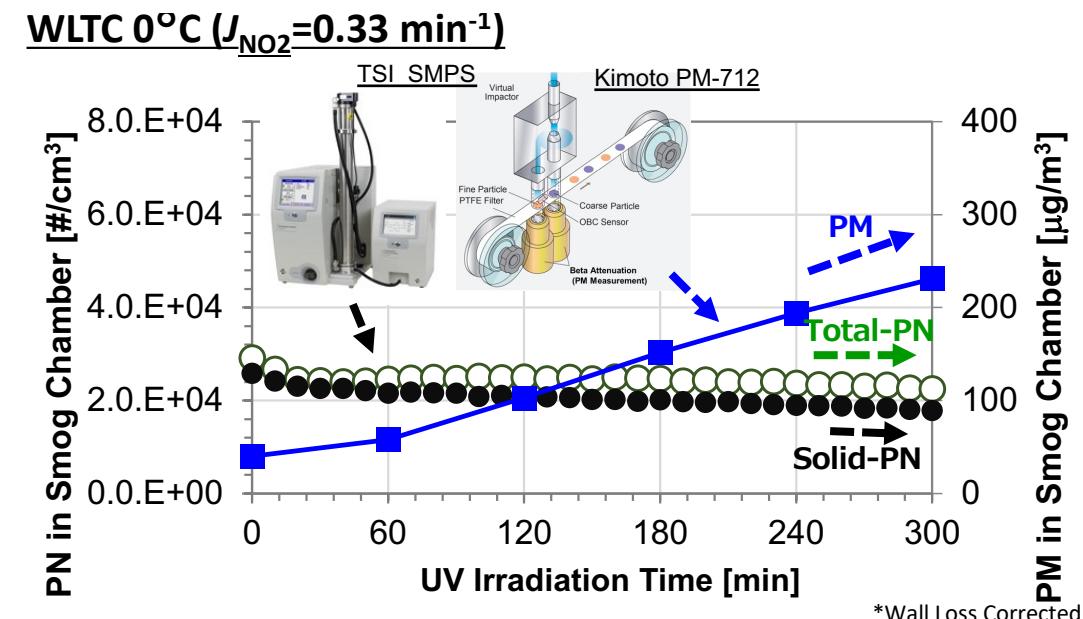
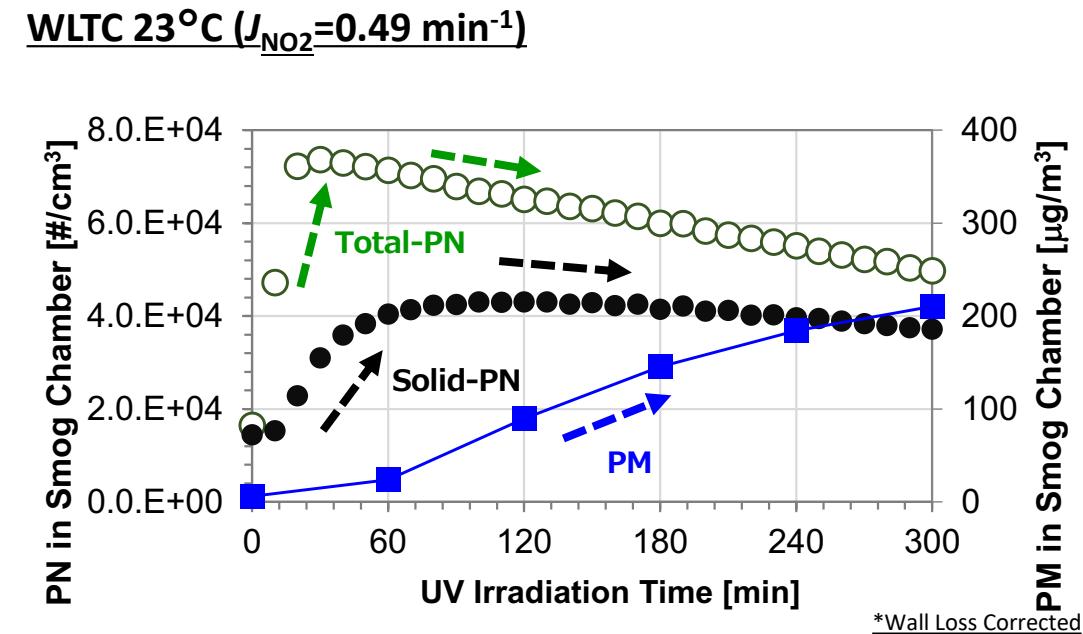
## Secondary Inorganic Aerosols (SIA) $\text{NH}_3 + \text{HNO}_3 \rightleftharpoons \text{NH}_4\text{NO}_3$



# PM vs PN : Secondary Formation Potential Indicators

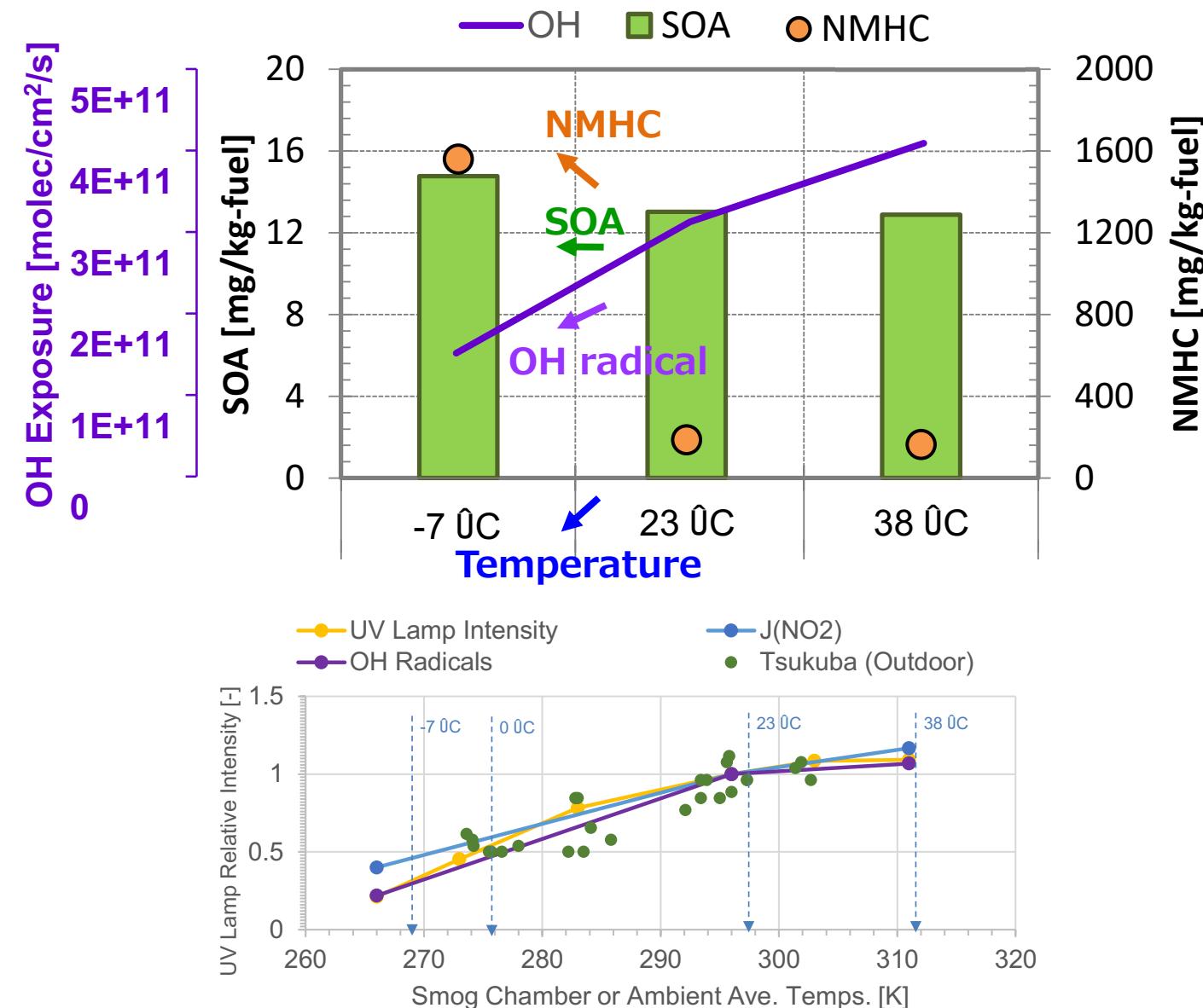
2

- PN Concentration is NOT generally Increased, since PN since Gas Condenses in Soot Pores or Aggregate Shrinks. Therefore, PN is difficult to determine Secondary Formation.
- PM Concentration is Simply Increased, and it is possible to determine Secondary Formation.
- **PM is a Simply and Robust Indicator of Secondary Particle Formation potential.**



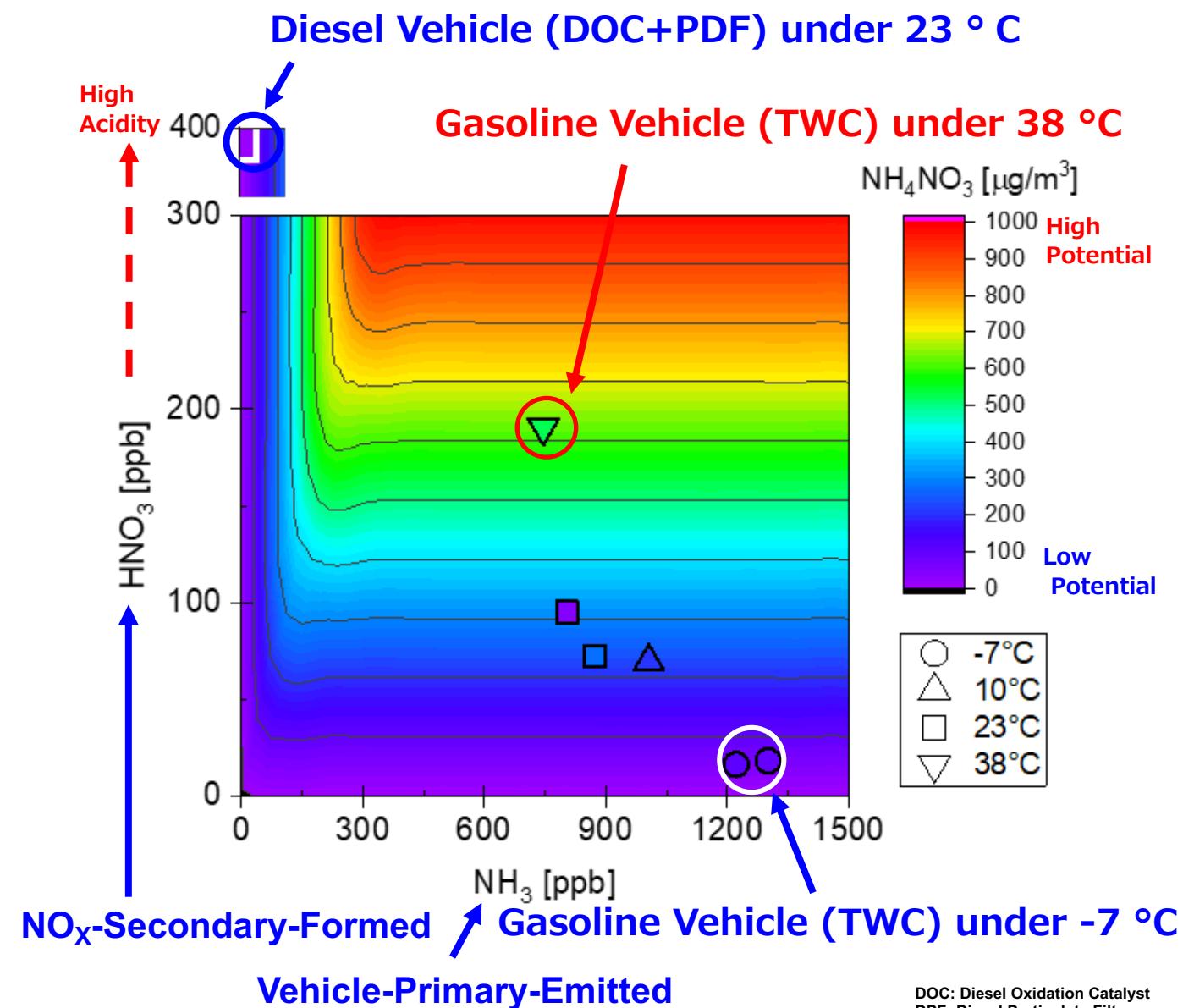
# SOA Formation

- Increase in SOA is NOT LINEAR with increase in Precursor NMHC at Low Temperatures.
- OH radicals required for Oxidation Reaction are reduced in Low-Temperature Environments.
- **NMHC Increase at Low Temperatures may NOT Linearly reflect a Worst-Case Scenario for SOA Formation Potential.**



# NH<sub>4</sub>NO<sub>3</sub> formation

- Contour map showing Secondary Formation of NH<sub>4</sub>NO<sub>3</sub> Particles from NO<sub>x</sub>-derived HNO<sub>3</sub> and Vehicle-emitted NH<sub>3</sub>.
- Secondary formation of NH<sub>4</sub>NO<sub>3</sub> is NOT Linear and it is important to further study the balance of Neutralizing Nitric Acid formed from NO<sub>x</sub> with NH<sub>3</sub>.**



# Take-Home Messages

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- PM is a Simple and Robust Indicator of Secondary Particle Formation.
- NMHC Increase at Low Temperatures may NOT Linearly reflect a Worst-Case Scenario for SOA Formation Potential.
- Secondary formation of  $\text{NH}_4\text{NO}_3$  is NOT Linear, and it is important to further study the balance of Neutralizing Nitric Acid formed from  $\text{NO}_x$  with  $\text{NH}_3$ .

## Next Steps

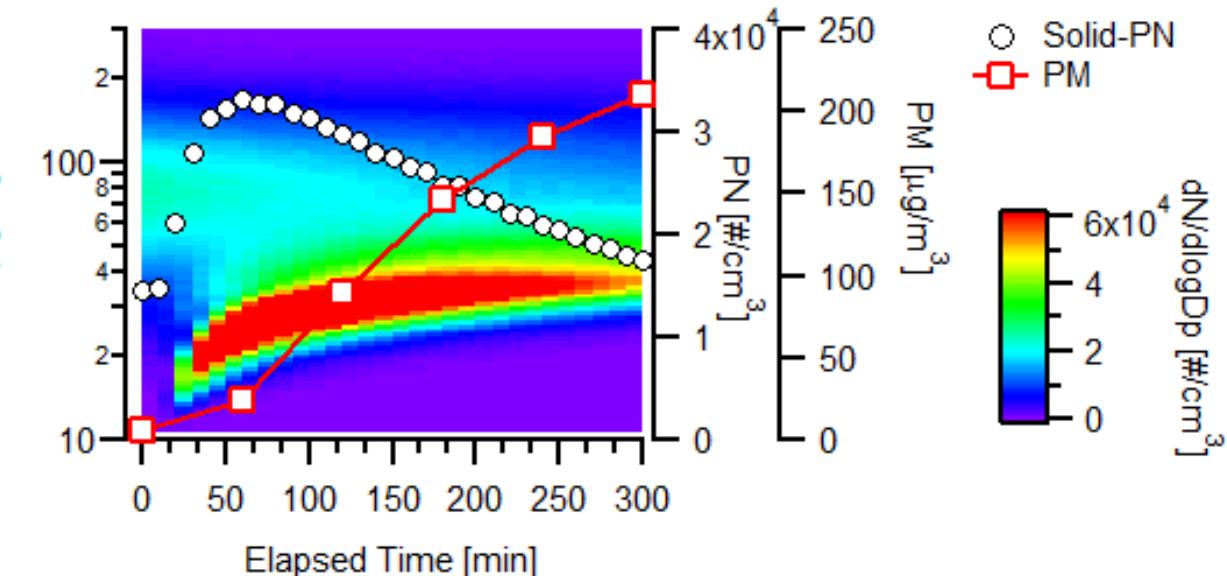
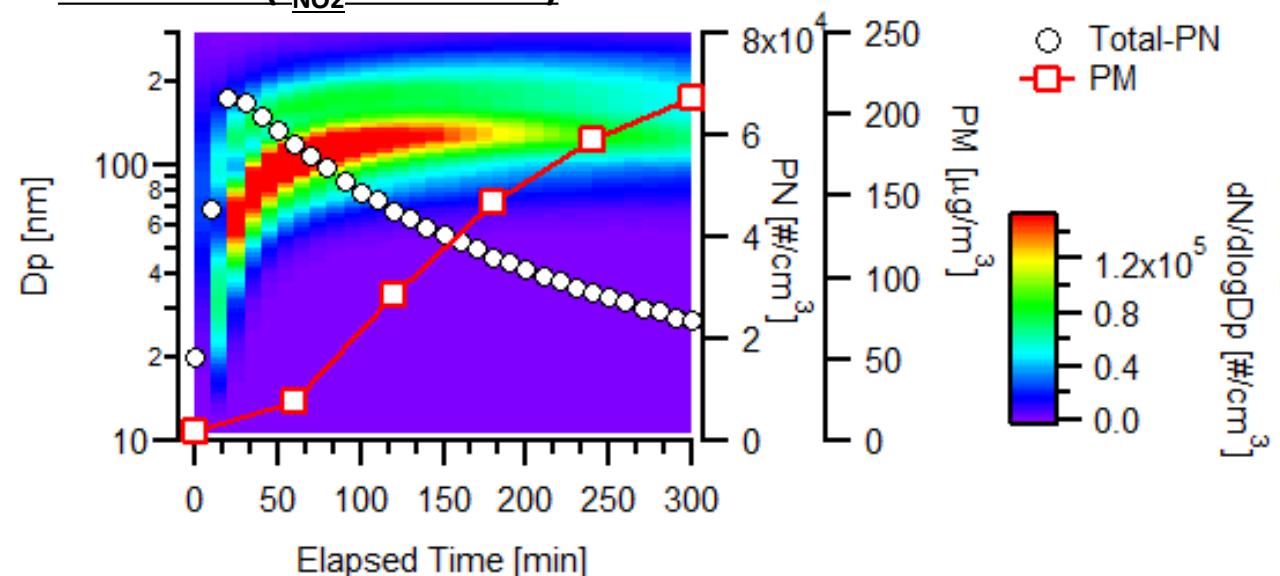
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- Relationship between SOA formation and Oxidative Potential (e.g. DTT assay) in case of NMHC Reduction.
- Effects of Acidity (or pH) on  $\text{NH}_3$  Reduction.
- Correlation of SOA and SIA Formation Potentials with Total PN and Solid PN Emissions.

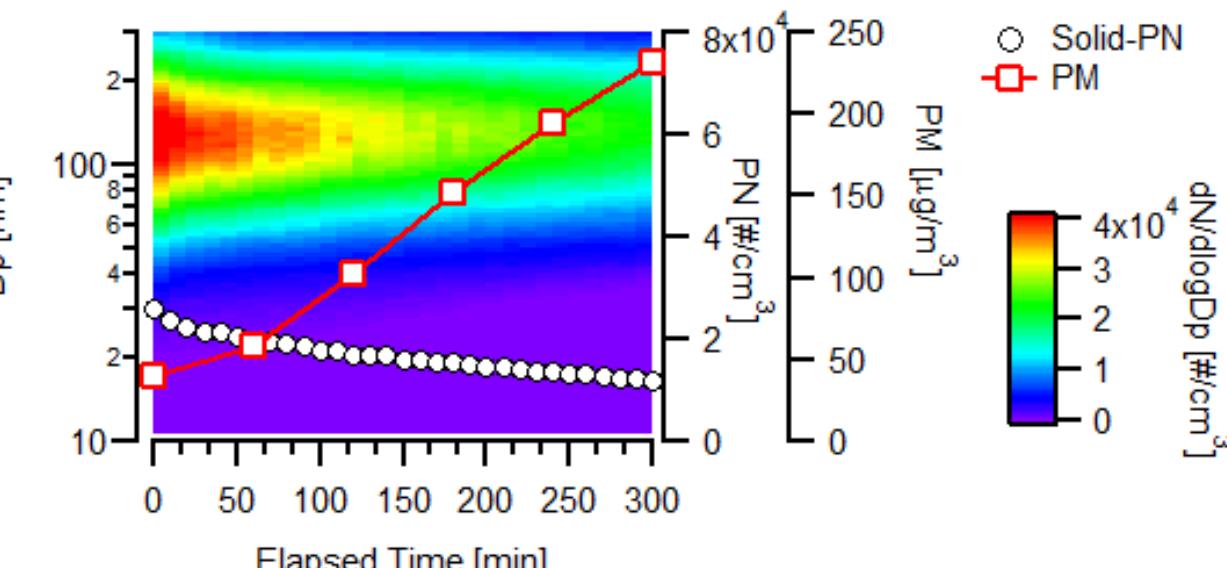
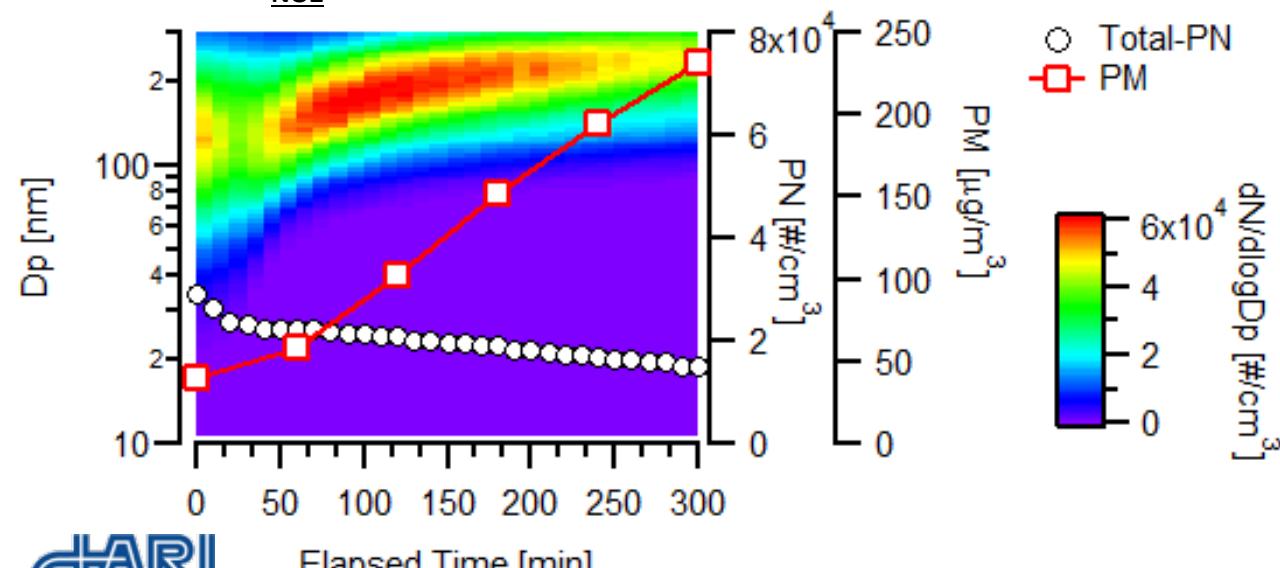
# Appendix

# PM vs PN Sizes

WLTC 23°C ( $J_{NO_2}=0.49 \text{ min}^{-1}$ )



WLTC 0°C ( $J_{NO_2}=0.33 \text{ min}^{-1}$ )



# Research Aims & Experimental Method

- Aim of This Study is to measure Formation Potential of SOA and SIA.
- Vehicle Emissions (WLTC (Worldwide-Harmonized Light Vehicles Test Cycle, 4 Phase) were delivered to Photochemical Smog Chamber ( $7.5 \text{ m}^3$  ) from Ejector Diluter attached Directly to Tail Pipe.
- Environmental Temperatures (Vehicle and Smog Chamber) were set form - 7, 0, 10, 23, 38 °C, etc.

